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An integrated thermal and hydrometallurgical process for the recovery of Silicon and Silver from end-of-life crystalline Si photovoltaic panels

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INTRODUCTION

End-of-life photovoltaic panels: an environmental problem

Photovoltaic (PV) panels are classified as Waste Electrical and Electronic Equipment (WEEE)

They require dedicated treatment at their End of Life (EoL)



Weckend, S., Wade, A. & Heath, G. (2016). End-of-life management: solar photovoltaic panels. International Renewable Energy Agency (IRENA) and International Energy Agency - Photovoltaic Power Systems (IEA-PVPS).

Silicon-based photovoltaic panels: structure

Silicon-based PV modules hold the dominant market share (90% in 2014 to 45% in 2030)



Padoan, F.C.S.M., Altimari, P. & Pagnanelli, F. (2019). Recycling of end of life photovoltaic panels: A chemical prospective on process development. Solar Energy 177: 746-761.

Silicon-based photovoltaic panels: composition

Material	% ww
Glass	70-75
Aluminum	10-18
EVA	5.0-6.5
Tedlar	1.5-3.5
Silicon	3.35-3.65
Copper	0.60
Tin	0.12
Zinc	0.12
Lead	0.06
Silver	0.004-0.06

- Silver is the most expensive component per unit of mass of a Si-based panel
- Silicon is a critical raw material (EU 2020)

The amount of the recovered materials (Si, Al, Cu) suggests a potential benefit

Silicon-based photovoltaic panels: recycling



MATERIALS and METHODS

Characterization of PV materials

Unused Si cells

Front surface



Back surface



PV materials

- EoL Si-based PV panels (Hyundai)
- Unused Si cells (Shenzhen Yima Technology Com)

Characterization

- Scanning Electron Microscopy (SEM)
- X-Ray Diffraction (XRD)
- Thermogravimetry (TG) & Differential Thermal Analysis (DTA)
- Wavelength Dispersive X-Ray Fluorescence (WD-XRF)
- Atomic Absorption Spectrometry (AAS)

Thermal treatment and leaching/recovery tests

Thermal treatment

- Shredding
- Sintering
- Separation/Classification
- Ball milling

Metal extraction

- Nitric acid
- Sulfuric acid

Metal recovery

- NaOH etching to remove the anti-reflection coating / recover Si
- Ag recovery via precipitation as AgCl
- Ag electrowinning

RESULTS and DISCUSSION

Internal structure of Si solar cell, Cu conductors and Ag, Al electrodes



Thermal treatment



The decomposition of the EVA and Tedlar[®] layers is completed in the temperature range of 540-550 °C

Separation/classification



Cu ribbons manually separated

Glass particles > 2000 μm fraction mass: 85% w/w Si-cells 200-2000 μm fraction mass: 7.5% Ash < 200 μm

Si-cells were mixed with the fine fraction to avoid any losses of useful materials in the ash and ball milled to -90 μ m

Chemical composition

Element	EoL cells (% ww)	Unused cells (% ww)
Si	81.4 - 83.0	81.0 - 87.0
Al	5.0 - 5.4	9.5 - 12.0
Ag	0.7 - 0.8	1.0
Cu	0.5	0.2
Pb	0.2 - 0.3	0.1
Sn	0.1	0.1

Aluminum content is about 10 times the silver content and this should be considered during the subsequent hydrometallurgical treatment

Leaching for Ag and Al extraction





Recovery of Silicon



Front (a) and back (b) surface of Si flakes upon leaching with 5 M $\rm HNO_3$

Recovery of Silicon



Front (a) and back (b) surface of Si flakes upon alkaline etching with 2.5 M NaOH





Silver electrowinning from leachates upon extraction with 5 M HNO_3 (S/L: 30%; init. Ag conc: 2.5-5 g/L; constant current I=0.24 A)



Spongy Ag with Al and Pb impurities deposited on the cathode

Flowsheet I

One-step extraction with \mbox{HNO}_3 and separation of metals by

- (i) successive precipitations
- (ii) electrowinning of Ag and precipitation of Al





Conclusions

- This study proposes an integrated process for the recovery of raw materials from EoL Si-based photovoltaic panels.
- Initially, the PV panels were dismantled and thermally treated for delamination. Then they were leached for the extraction of silver and aluminum, in one step with HNO₃ or selectively in two steps with H₂SO₄ and then HNO₃. Finally, the Si cells were etched with NaOH leaving the silicon surface free of metallic impurities.
- The recovery of silver may be achieved by two different processes: (i) precipitation of Ag as AgCl or (ii) electrowinning of Ag. Soluble aluminum precipitates as aluminum hydroxide upon neutralization.
- The proposed process is simple and efficient as well as totally in line with urban mining of raw materials and improving the circularity of the photovoltaic market.