



## Nitrogen and phosphorus recovery from swine manure



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## INTRODUCTION

The use of organic fertilizers such as livestock wastewaters for agricultural production, will be a practical demonstration of circular economy in action, with many economical and environmental advantages.



Nevertheless, the efficiency of nutrient use through the whole food chain should be increased

Nutrient Use Efficiency for Nitrogen = 20%

Nutrient Use Efficiency for Phosphorus = 30%

**Low NUE** = large leakage of nutrients = health problems and environmental issues



## INTRODUCTION

Environmental issues driven by nutrient loss:

- Water: eutrophication due to N and P leaks (reduction of water quality and biodiversity)
- Atmosphere: NO<sub>x</sub> and NH<sub>3</sub> reduce air quality and affect human health. N<sub>2</sub>O and methane contribute to climate change



## INTRODUCTION

**What to do:** To avoid nutrient losses and increase nutrient recovery, increasing NUE.

**How to do it:** Applying technologies that can be used in the key waste streams to recover nutrients with sufficient quality, safe, easy to store, and easy to use by farmers.



**OBJETIVE OF THIS WORK**



## METHODOLOGY

The experiments designed and implemented for the N and P recovery from swine manure, had two phases:

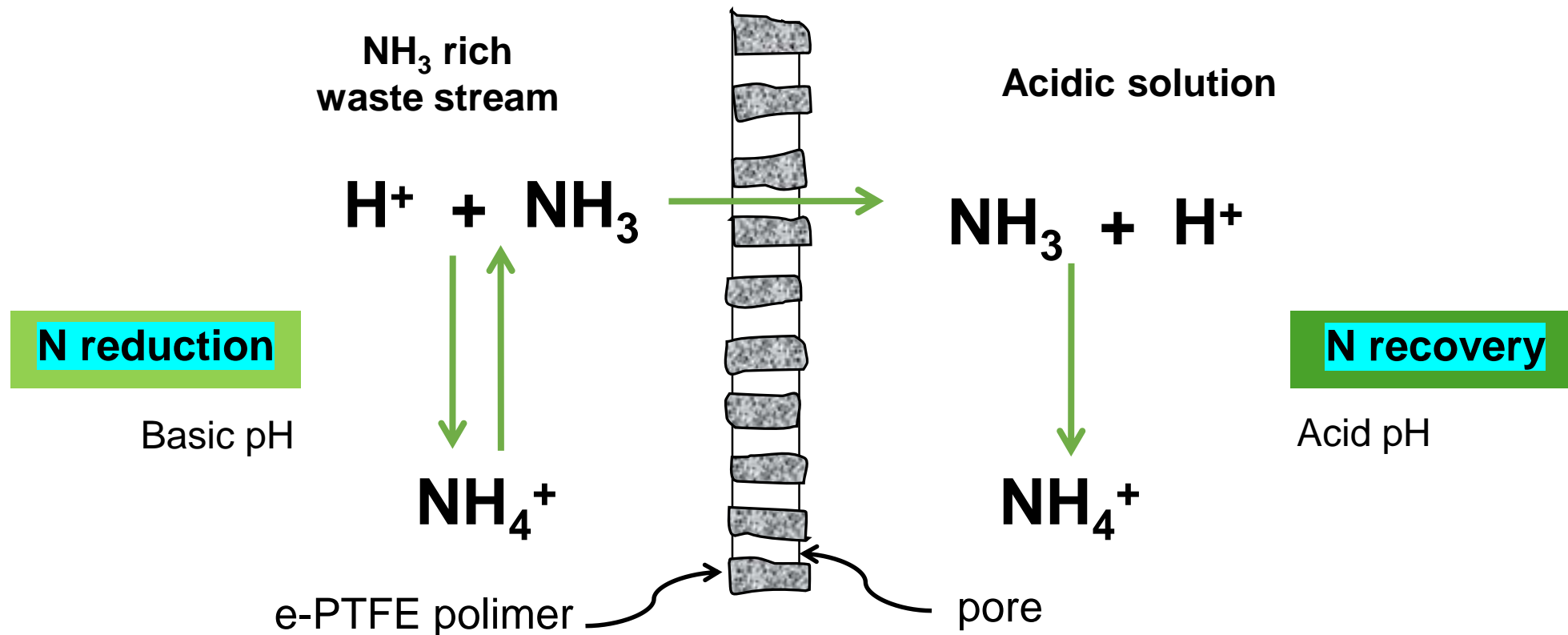
**A) FIRST PHASE:** N was recovered from swine manure using gas-permeable membrane (GPM) technology

**B) SECOND PHASE:** P was recovered from the previously treated swine manure, using an electro-dialytic process, which involves the use of ion-exchange membranes



METHODOLOGY: FIRST PHASE

N recovery with gas-permeable membrane technology





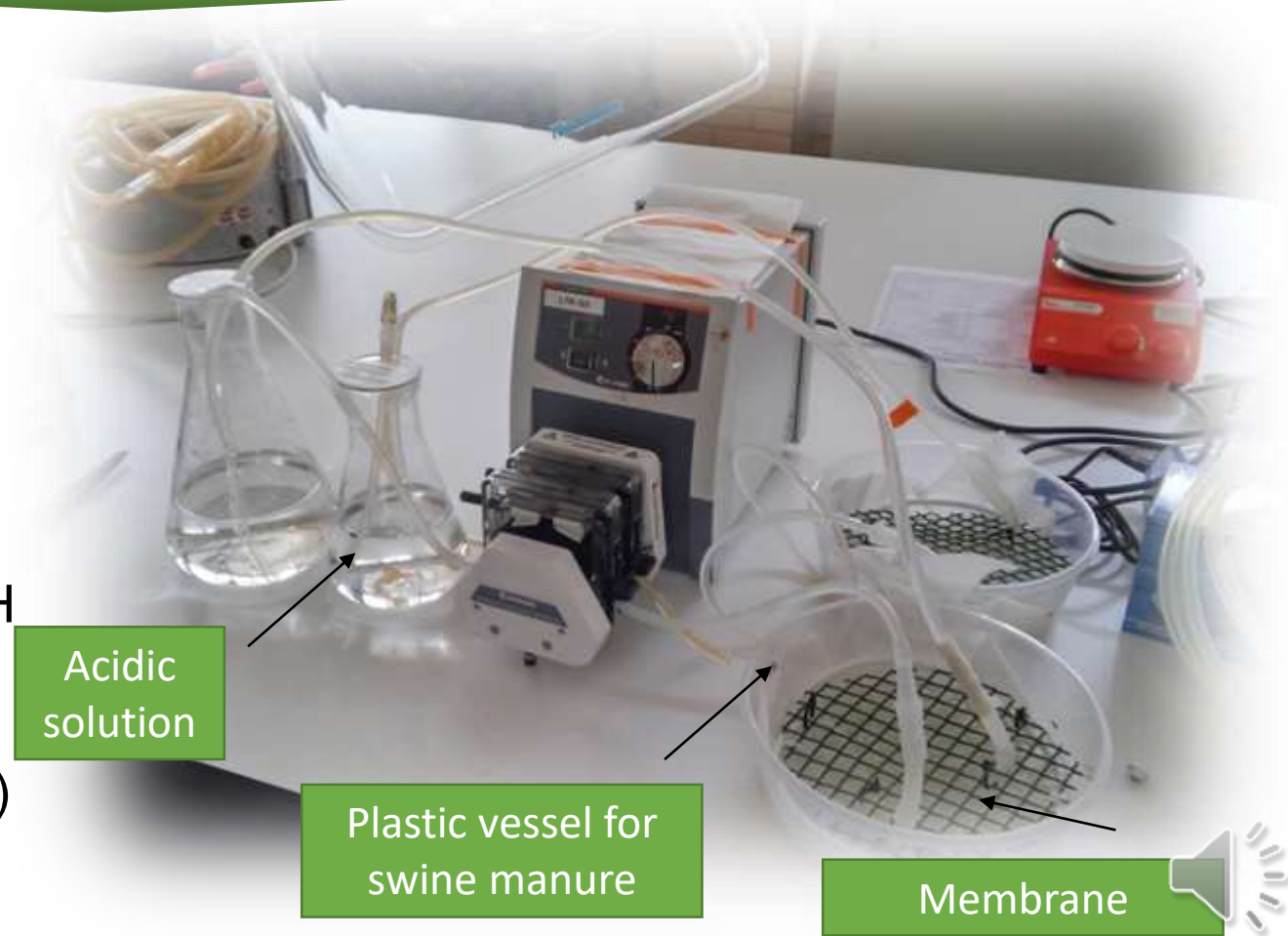
## METHODOLOGY: FIRST PHASE

Inside the membrane submerged in the swine manure, recirculates the acid solution ( $H_2SO_4$ ), recovering the ammonia as ammonium sulphate solution.

The swine manure was constantly agitated and low aeration was used.

To maximize this recovery the acidic solution pH was kept at  $pH < 2$

The amount of N recovered daily ( $mg NH_4 + L^{-1}$ ) and the pH were monitored.

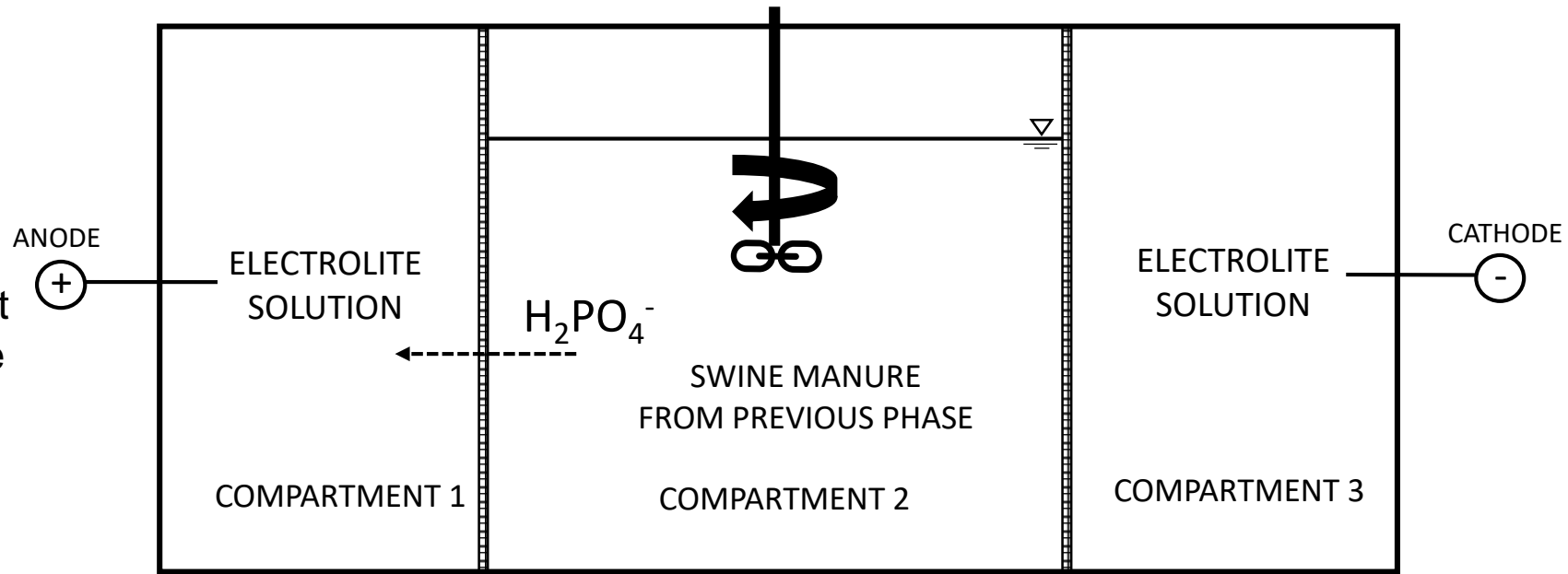


## METHODOLOGY: SECOND PHASE

### P recovery with electrodialytic process

The previously treated swine manure was put in compartment 2, and it was constantly agitated.

The electrical current is responsible for the ions movement from the residue to the electrodiolitic solution.



Compartments 1 and 3 contained  $\text{NaNO}_3$  as electrolyte, to recover cations and anions respectively

Anionic exchange membrane (AEM)

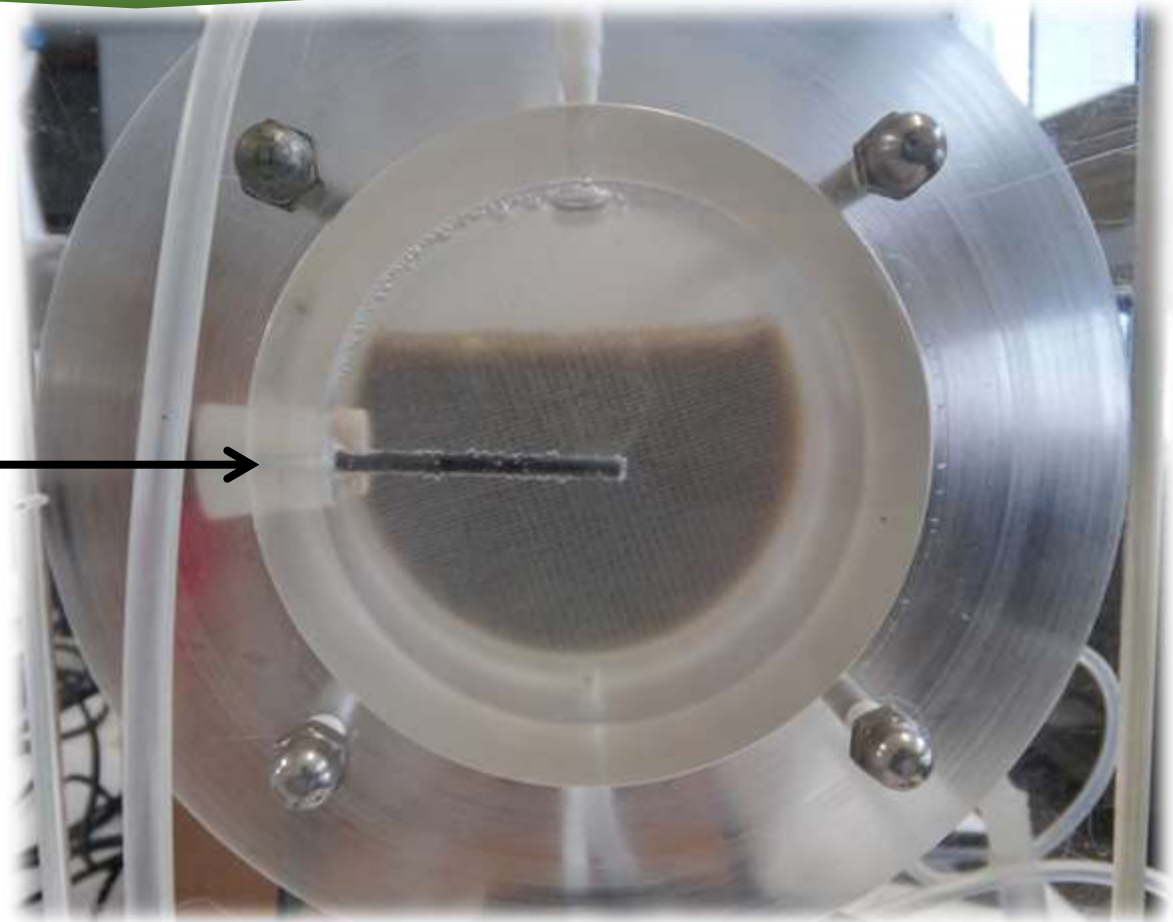
Cathionic exchange membrane (CEM)





METHODOLOGY: SECOND PHASE

Water splitting in the anionic membrane keeps low the pH of the residue

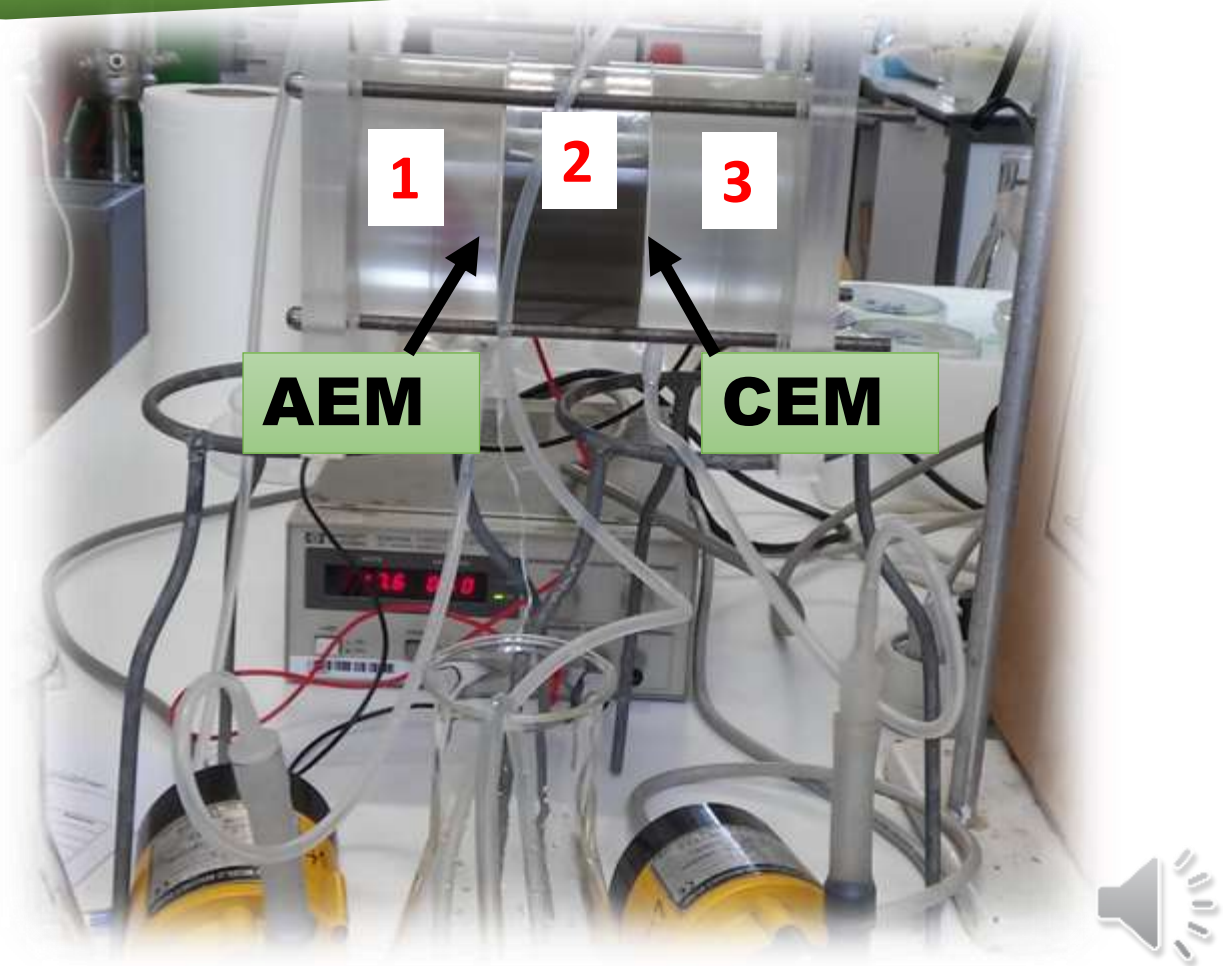


## METHODOLOGY: SECOND PHASE

The recirculation of the electrolytic solution as made with a pump, between the corresponding compartment and an external flask. The power supply maintained a current of 50 mA.

Samples of the manure and the electrolyte solution in each flask were taken daily to monitor the pH and conductivity and to analyse the amount of P.

The voltage and the electric current were also monitored daily.

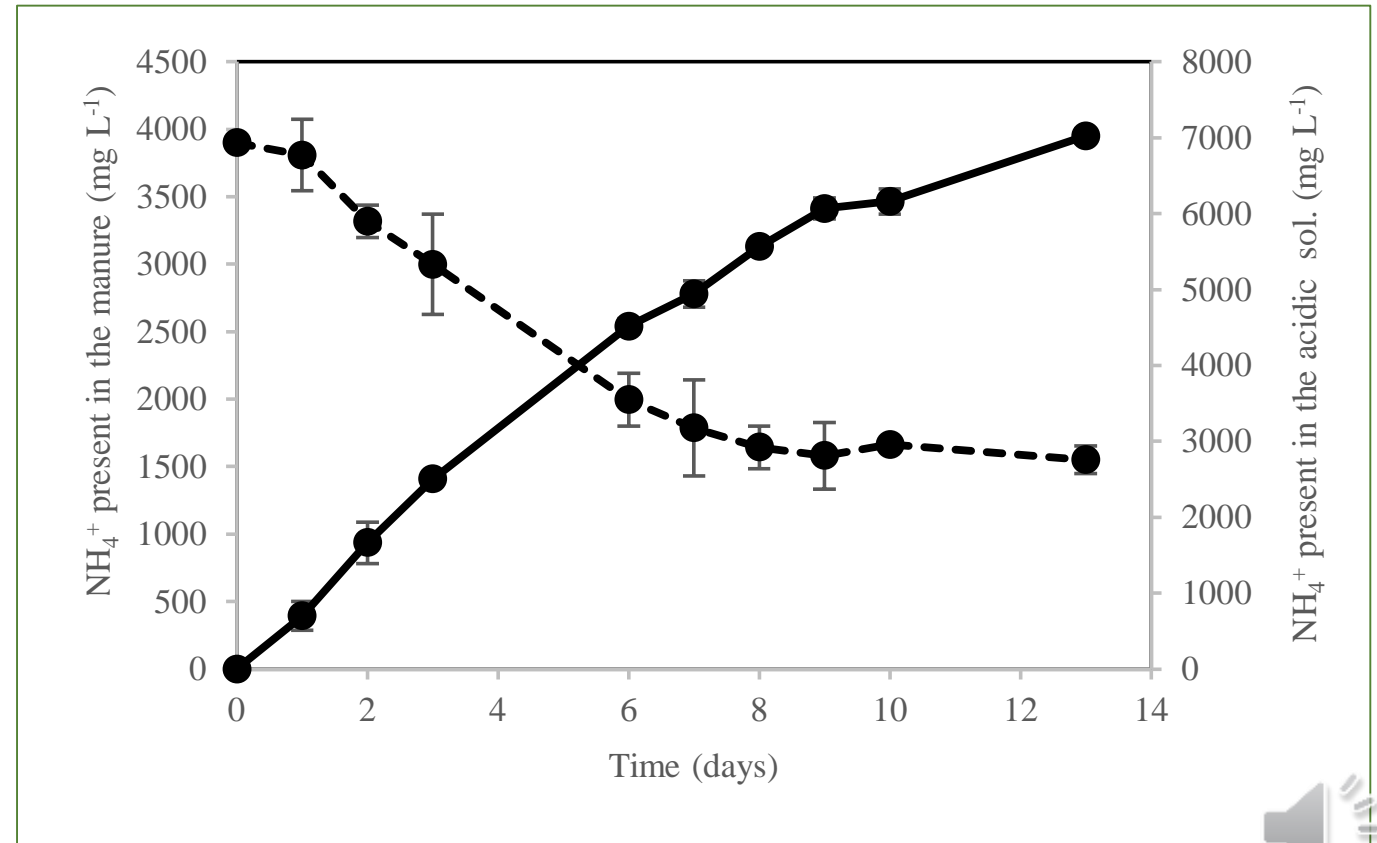


## RESULTS: FIRST PHASE

The first phase was carried out for 13 days.

It was observed a decrease of  $\text{NH}_4^+$  in the manure, from the initial concentration of  $3900 \text{ mg NH}_4^+ \text{ L}^{-1}$  to  $1550 \text{ mg NH}_4^+ \text{ L}^{-1}$ , which means an  $\text{NH}_4^+$  removal of  $2080 \text{ mg}$  of  $\text{NH}_4^+$ , 60% of the initial amount of  $\text{NH}_4^+$

Recovery of 51% of the removed ammonia in the acidic solution

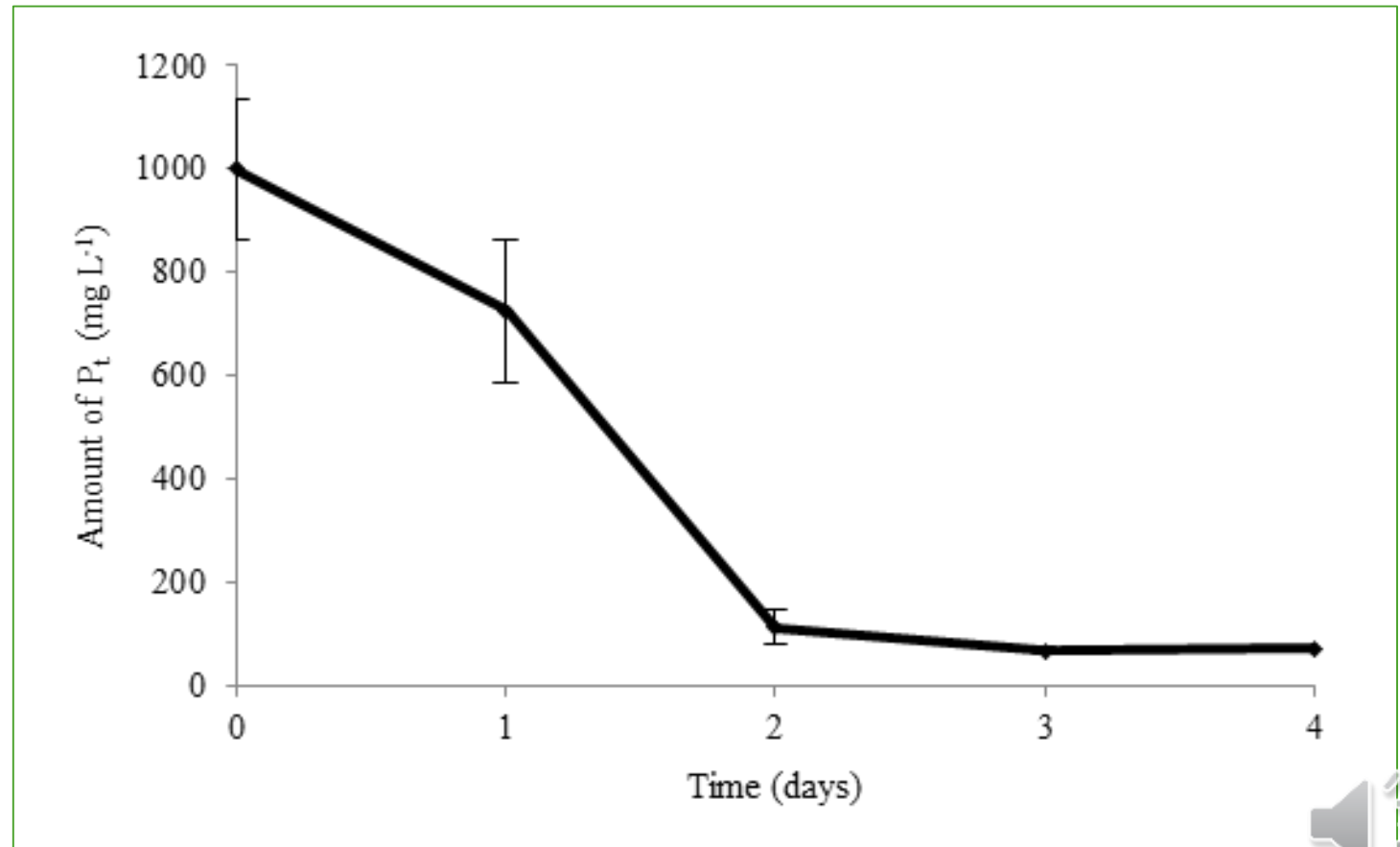


## RESULTS: SECOND PHASE

The second phase were conducted for 4 days.

The initial concentration of P of the manure was 1000 mg L<sup>-1</sup>, and during the experiment, the amount of P decreased reaching a final value of 110 mg L<sup>-1</sup>.

Recovery of 89% of the initial P into the electrolytic solution.



# CONCLUSIONS

**AMMONIUM SULPHATE SOLUTION :**  
NO NEED OF FURTHER TREATMENT  
EASY TO STORE AND SAFE  
APPLICATION (FERTIGATION) AS LIQUID BIO-  
BASED FERTILIZER



**P RICH SOLUTION :**  
PRECIPITATION AS STRUVITE  
EASY TO STORE AND SAFE  
APPLICATION AS  
SLOW-RELEASE BIO-BASED FERTILIZER





## CONCLUSIONS

The results of this study showed the **great potential** of combining gas-permeable membrane + ion exchange membranes for nutrient recovery and reuse





# Thank you

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