



Valorization of Plastic Waste with the Aid of Solar Hydrothermal Liquefaction

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Overview

- ❖ Introduction
 - ✓ Plastic waste Industry
 - ✓ Hydrothermal Liquefaction (HTL) concept
 - ✓ The role of water
 - ✓ Solar Hydrothermal Liquefaction (SHTL) concept
- ❖ Aim and methodology
- ❖ Preliminary HTL tests
 - ✓ Setup and experimental conditions
 - ✓ Experimental results
 - ✓ Indicative physico-chemical characterization
- ❖ Lab-scale HTL-CST coupling
- ❖ Conclusions and future work



Plastic Waste Industry

EUROPE (EU28+NO/CH), 2018



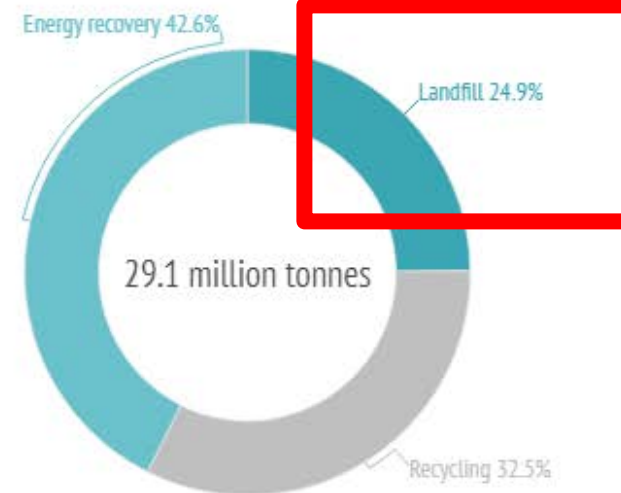
61.8 million tonnes

Plastics production

9.1 million tonnes

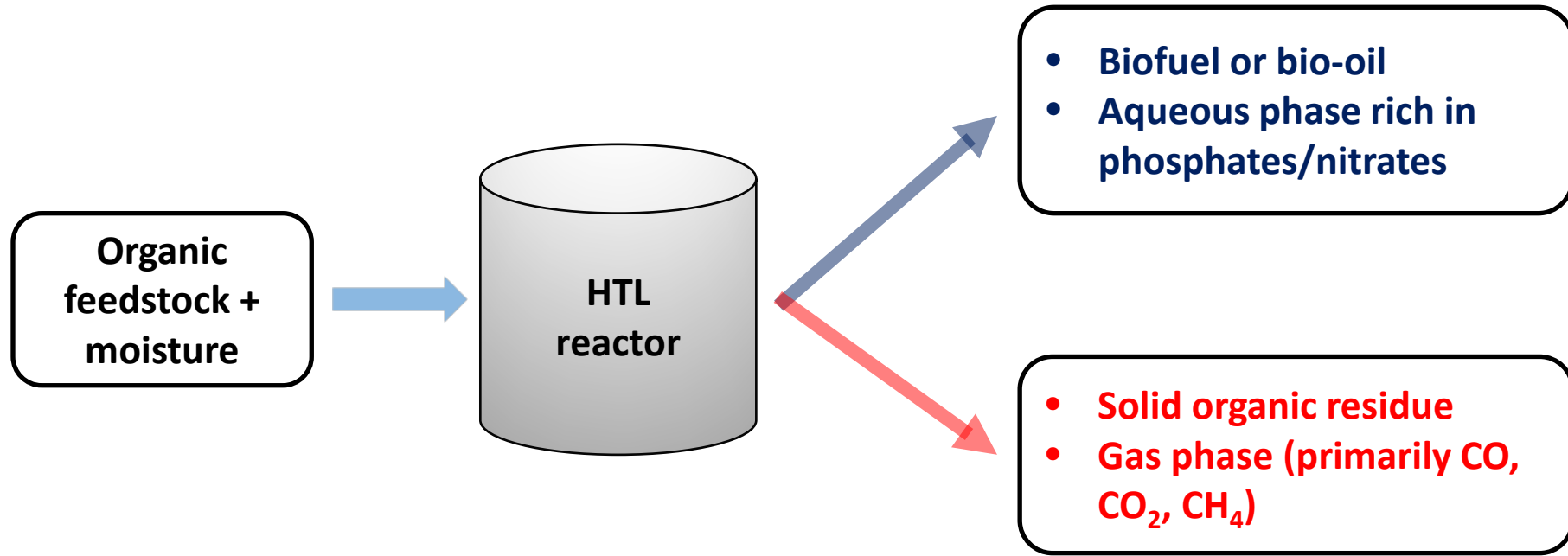
Plastic post-consumer waste collected to be recycled

PLASTIC POST-CONSUMER WASTE TREATMENT



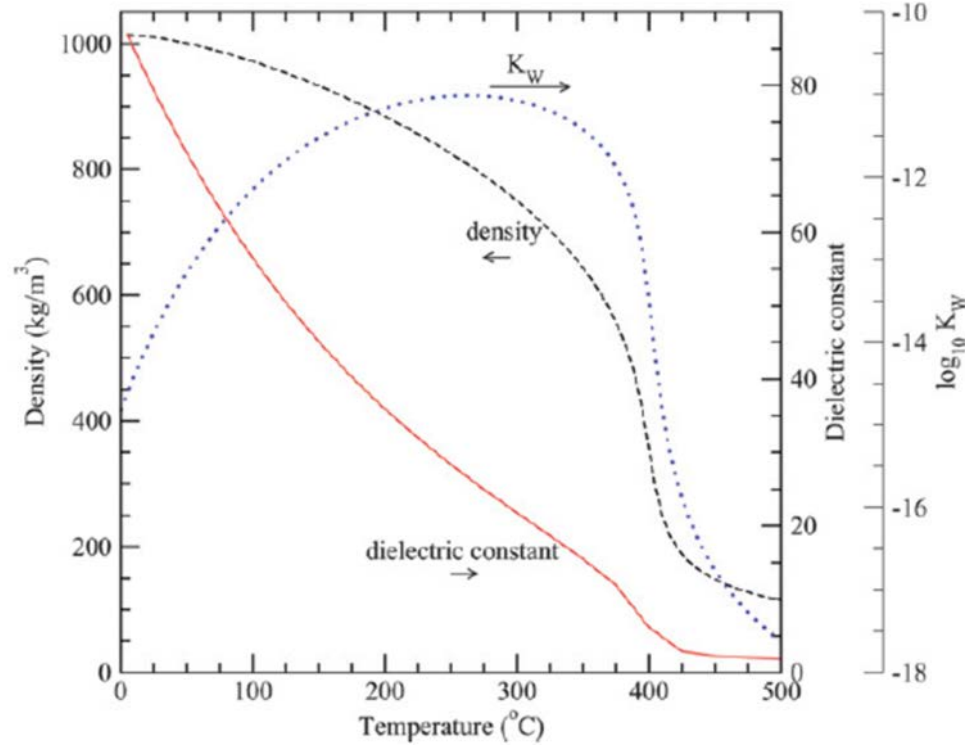


Hydrothermal Liquefaction (HTL) concept



- ❖ Thermochemical conversion of organic waste into added value products
- ❖ Suitable technique for organic waste/byproducts with high moisture
- ❖ Treatment at 250 - 500°C & 50 - 250 bar
- ❖ Use of a reducing gas and (optionally) a catalyst
- ❖ Mostly water (moisture) used as solvent in its subcritical/supercritical condition

The role of water



Temperature ↑ leads to

Density ↓

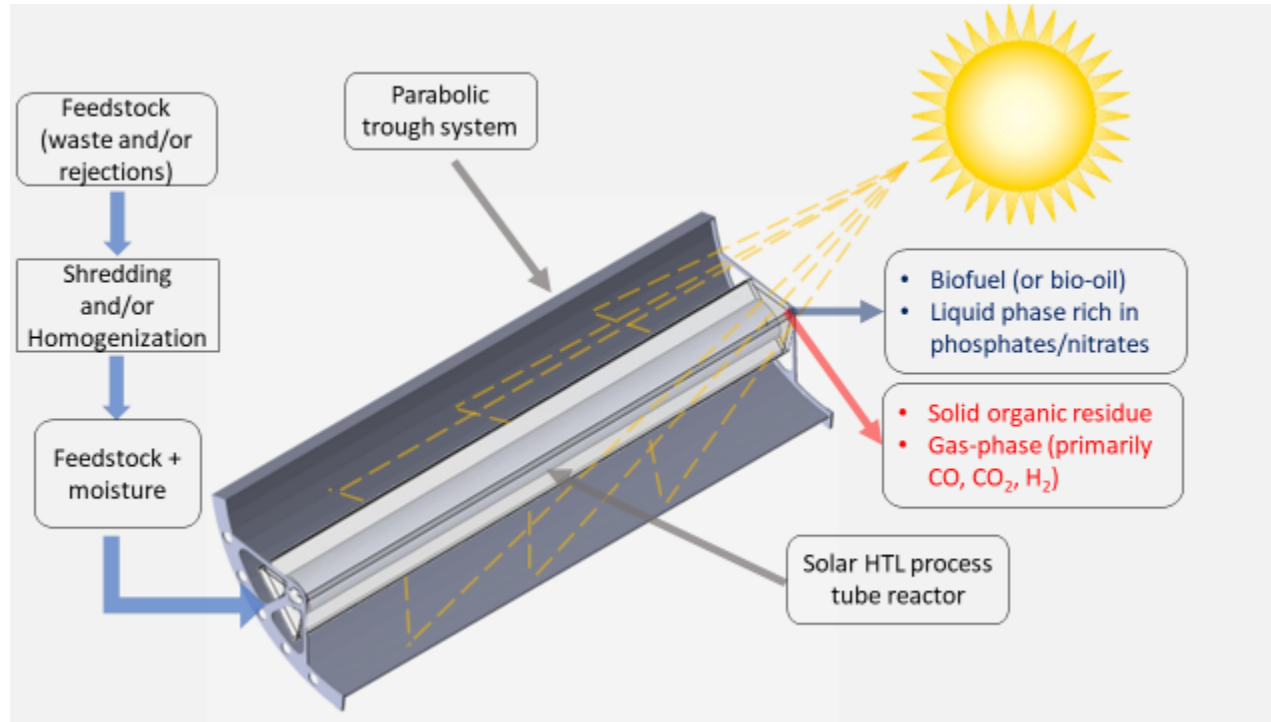
Dielectric constant ↓

Ionization constant (K_w) change

- ❖ Triple role as solvent, reactant and catalyst
- ❖ Non-polar solvent in appropriate pressure and temperature
- ❖ Avoidance of feedstock drying (energy demanding step)



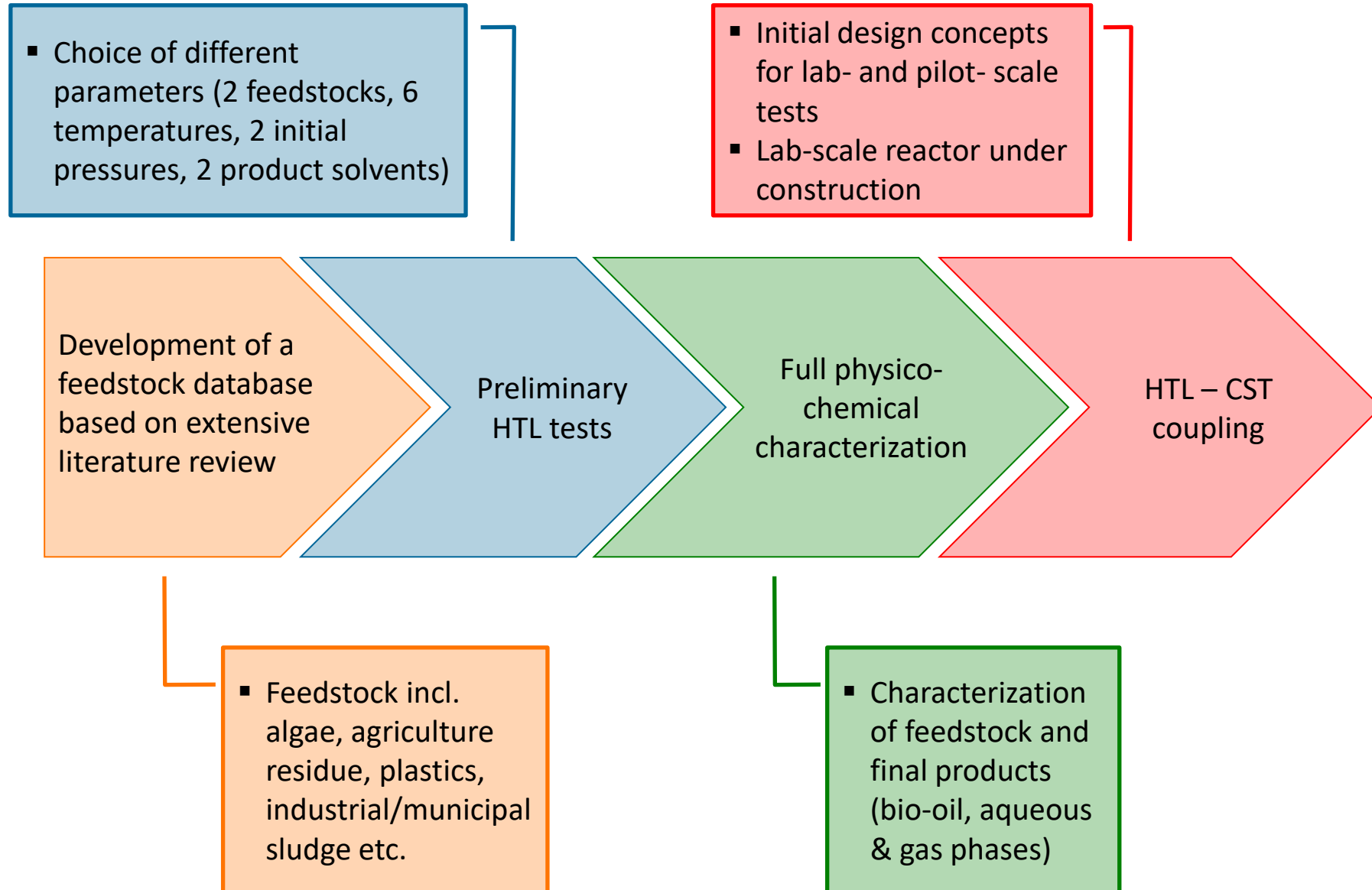
Solar Hydrothermal Liquefaction (SHTL) concept



- ❖ Solar energy as a power source
- ❖ Conceptual HTL & CST coupling flowsheet incl. preprocessing steps & outlet streams
- ❖ HTL tube reactor & parabolic trough system (used as an example)

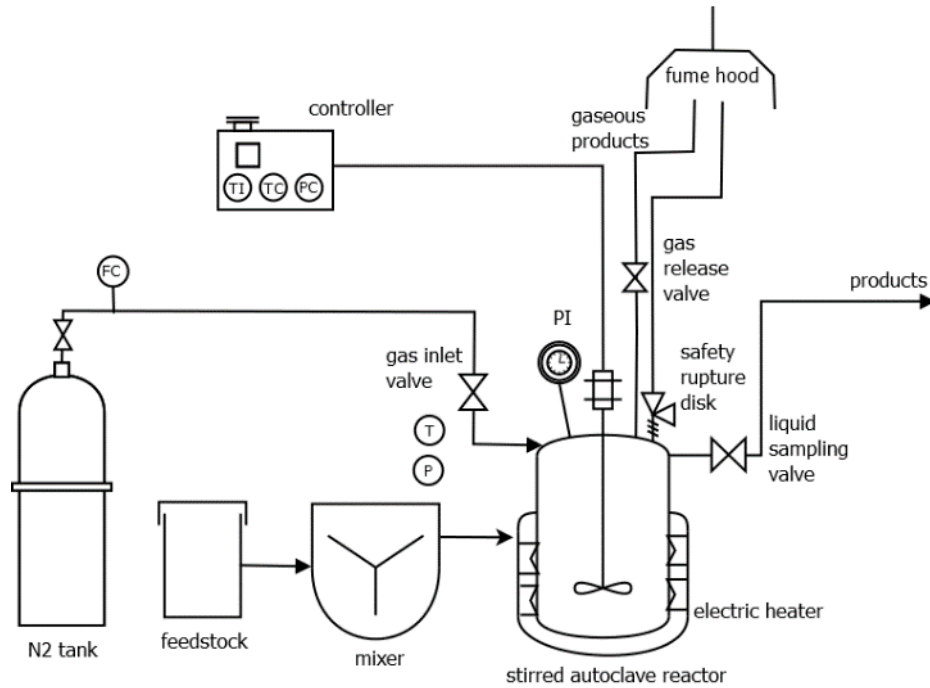


Aim of this work – methodology





Preliminary HTL tests – Setup



- ❖ Preprocessing of feedstock (shredding and/or mixing)
- ❖ 1.8 L HP/HT SS autoclave reactor
- ❖ Electronic controller for temperature & stirring speed regulation
- ❖ Use of an electric heater for reactor heating



Preliminary HTL tests – Experimental conditions

Feedstock	Polypropylene flakes (PP), plastic waste mix (PWM)
Reducing gas	N ₂
Temperature (°C)	350, 370, 375, 400, 425, 450
Initial Pressure (bar)	1, 20
Residence time (min)	30
Stirring	Continuous
Water/waste ratio (wt%)	90/10

PWM



PP





Products separation



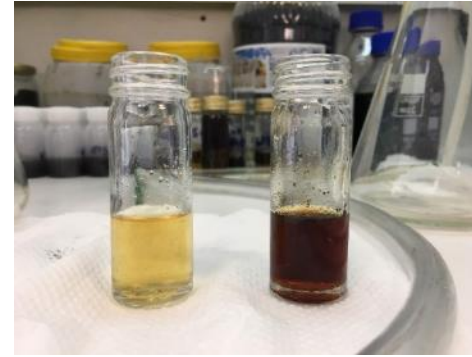
Gaseous products



Solvent addition



Filtration



Phase separation



Bio-oil

Products solvents:

- Dichloromethane (CH_2Cl_2 , **DCM**) \rightarrow b.p. 39.6 °C (polar)
- Acetone ($(\text{CH}_3)_2\text{CO}$, **DMK**) \rightarrow b.p. 56.1 °C (partially polar)
- Ethyl acetate ($\text{CH}_3\text{COOCH}_2\text{CH}_3$, **ETAC**) \rightarrow b.p. 77.1 °C (polar)

- Acetone was eliminated from further studies, unable to separate different phases

DCM



DMK





Preliminary HTL tests results (I)

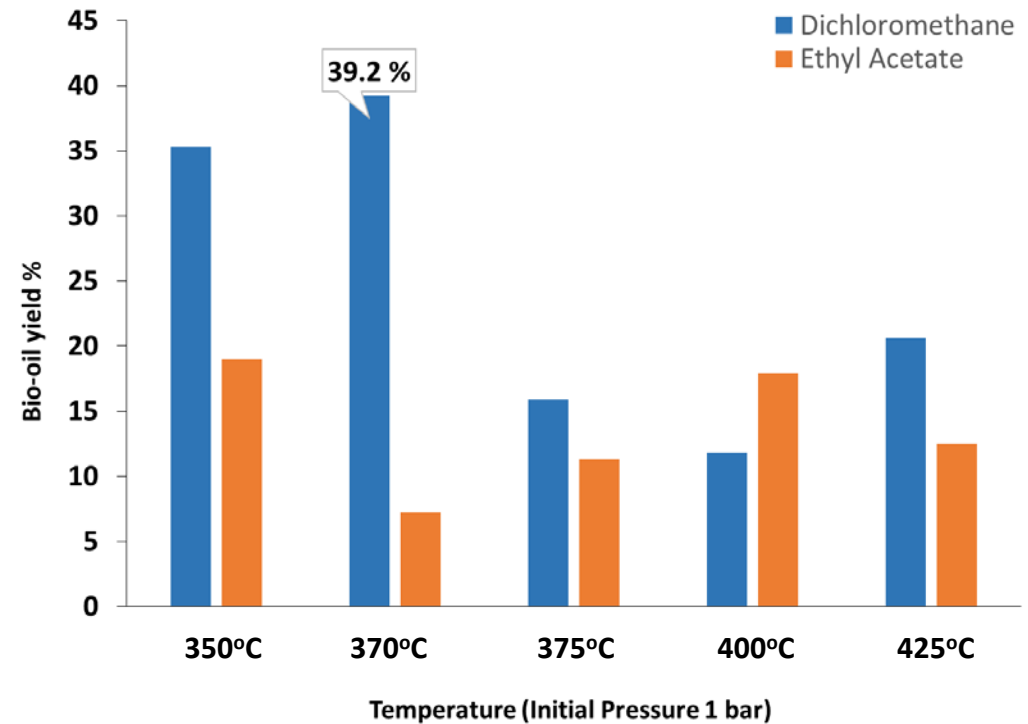
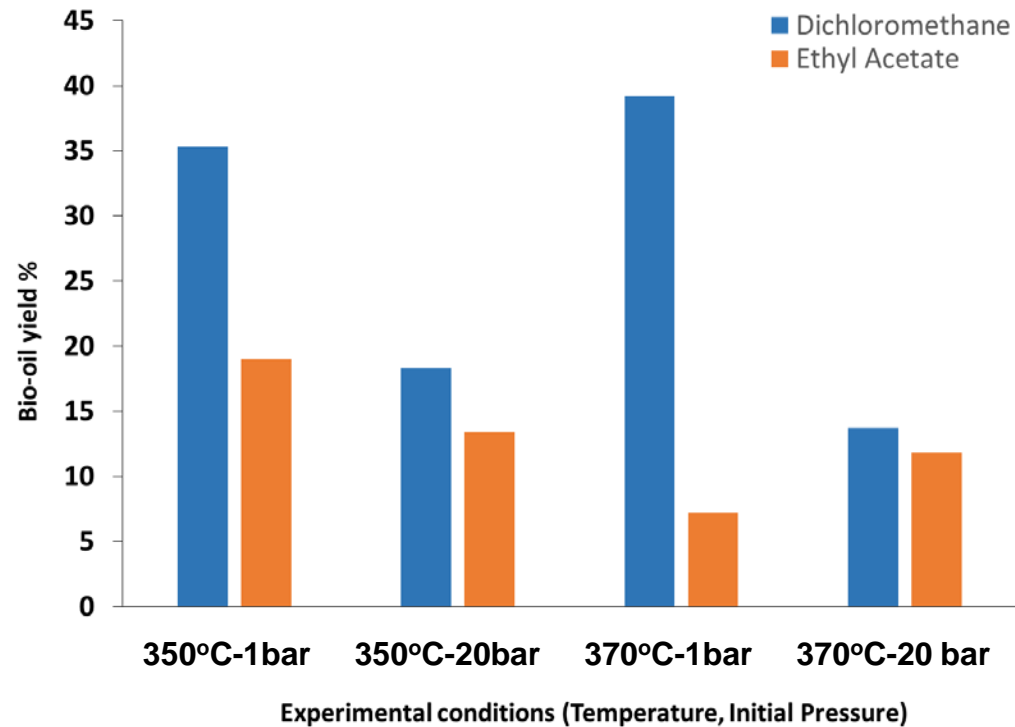
❖ Bio-oil yield calculation equation

$$\text{Bio - oil yield}(\%) = \frac{W_{\text{bio-oil}}}{W_{\text{feedstock}}} \times 100$$

$W_{\text{bio-oil}}$: weight of bio-oil (g)

$W_{\text{feedstock}}$: weight of dried feedstock (g)

❖ PWM bio-oil yield (%) for all conditions & different solvents



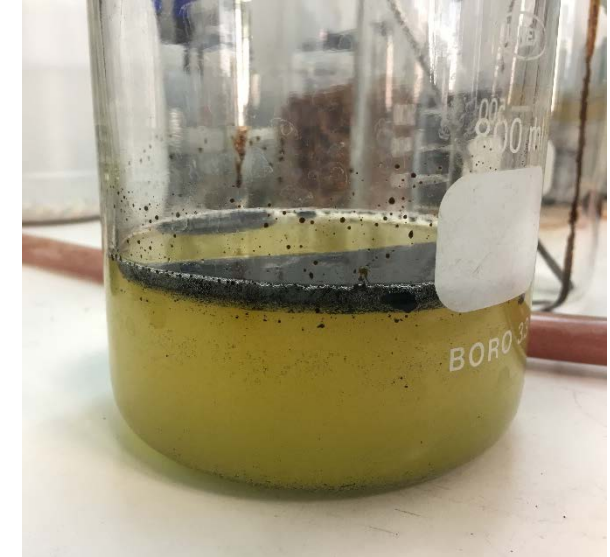
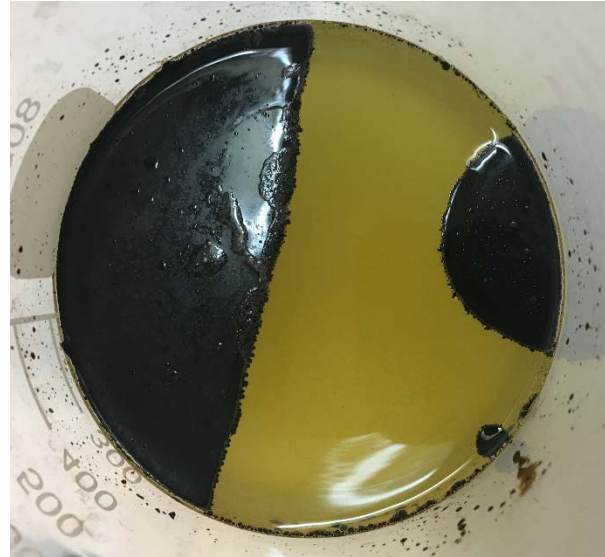


Preliminary HTL tests results (II)

- ❖ PP did not convert into bio-oil at temperatures below 425°C
- ❖ Bio-oil in PP case was separated from aqueous phase without use of solvent

Experimental conditions	Bio-oil yield (%)
400°C – 1 bar	0.0
425°C – 1 bar	53.8
450°C – 1 bar	45.4

* *Did not convert*



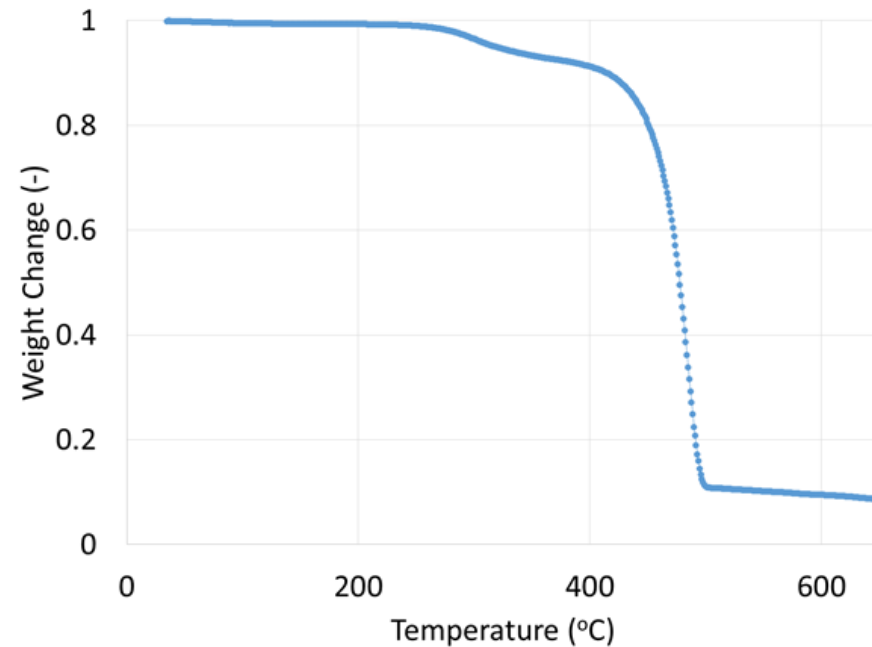
- ❖ Highest bio-oil yields observed in different experimental conditions, difficult determination of a specific experimental pattern → Bio-oil yield significantly different behavior depending on feedstock's main components



Physico-chemical characterization

❖ Feedstock

Thermogravimetric analysis (TGA) of feedstock



- Negligible moisture content
- 300-500°C decomposition of heavy organics
- Low percentage of inorganics

Inorganics analysis of plastic waste

Inorganic element	Value (mg/Kg)
Chlorides	390
Zinc	0.12
Sulfates	280
Copper	0.05
Fluorides	0.4
DOC	170
TDS	840

- Inorganics remaining in aqueous phase could be used in other applications (e.g. as a fertilizer)



Physico-chemical characterization

❖ Products

Gaseous (GC)

Gaseous product	CO ₂	CH ₄	CO	C ₂ H ₄	C ₃ H ₆	C ₂ H ₆	C ₃ H ₈
Volume ^(*) %	6.04	1.6	1.3	0.2	0.19	0.13	0.12

* The rest gaseous product volume corresponds to N₂

Bio-oil (GC-MS): fatty acids, phenols & long chain alkanes

Indicative compounds of bio-oil (GC-MS):

Benzene, 1,1'-(1,3propanediyl)bis

Heptadecane

Octadecane

Nonadecane

Eicosane

Heptadecane, 2,6,10,15-tetramethyl

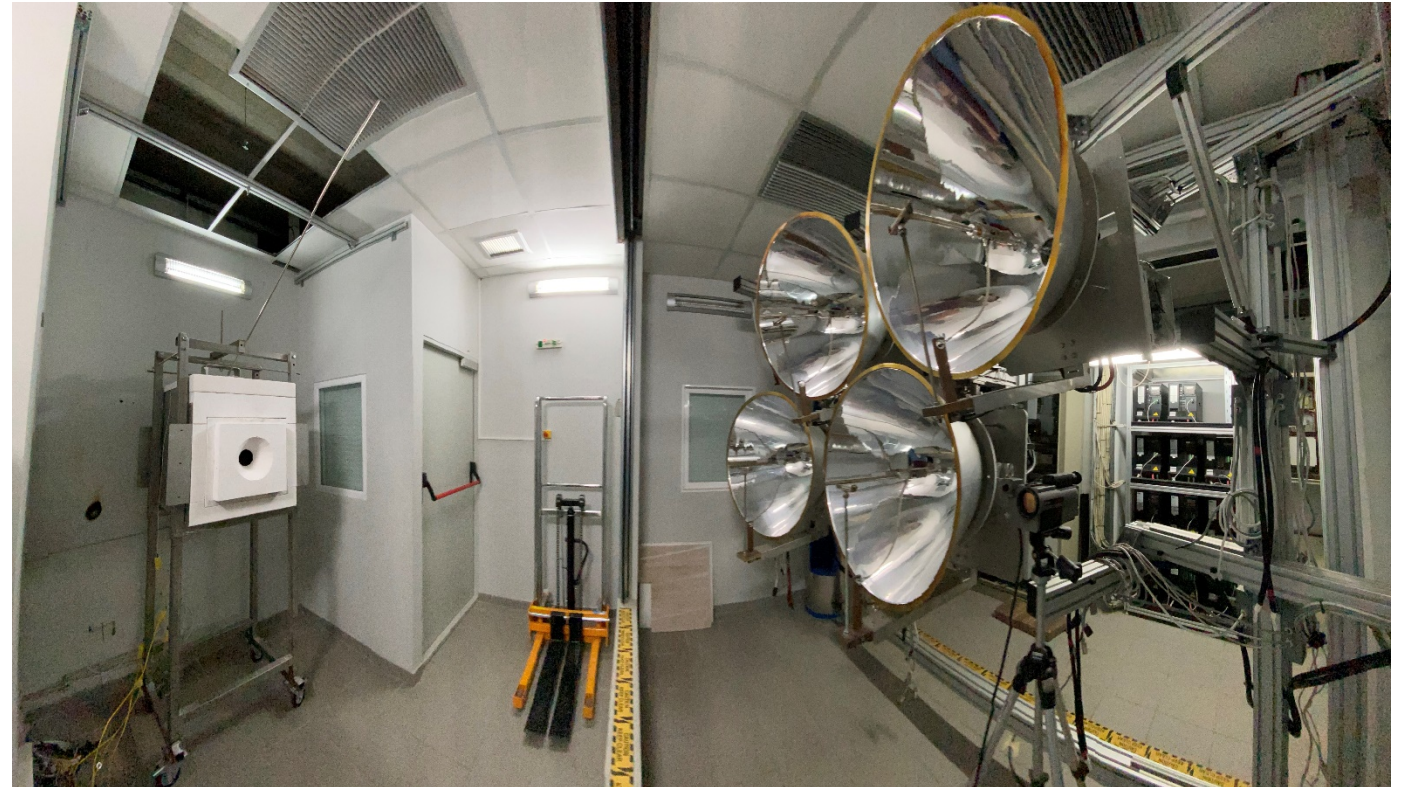
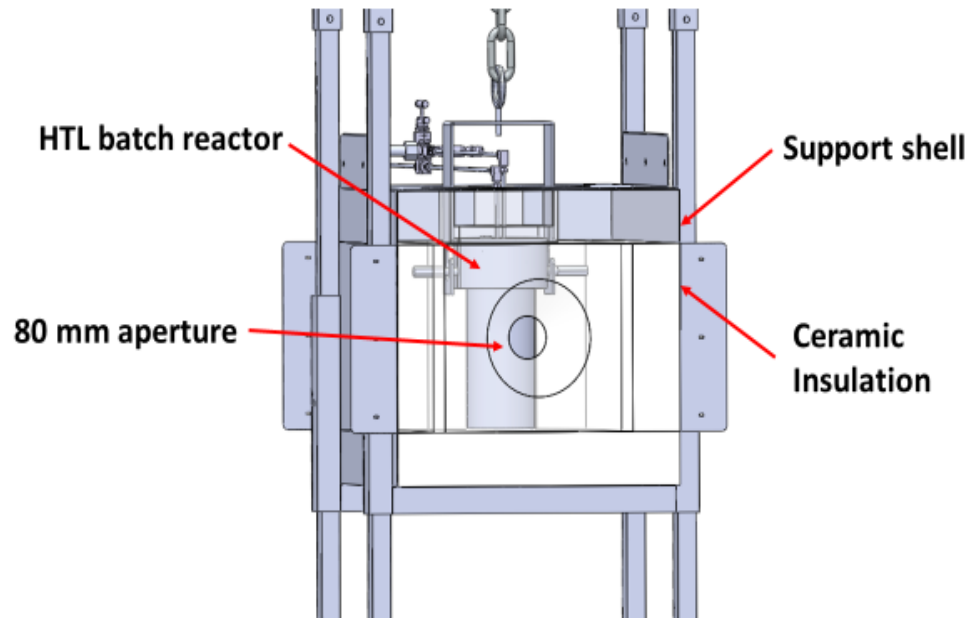
Heneicosane

1-Propene, 3-(2-cyclopentenyl)-2-methyl-1,1-diphenyl-



Lab-scale HTL-CST coupling

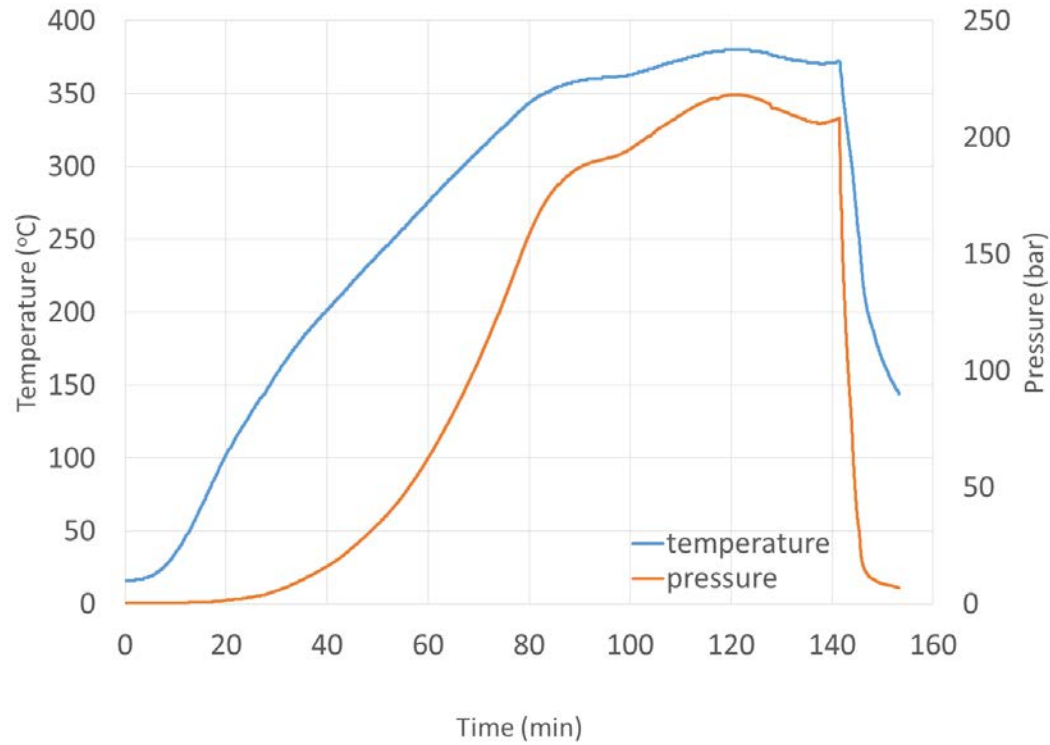
- ❖ Preliminary tests of solar HTL → Utilization of in-house Solar Simulator
- ❖ 4 ellipsoid reflectors with 6kW_{el} Xenon arc lamps
- ❖ 2L reactor arrangement designed and constructed for HTL-CST coupling tests



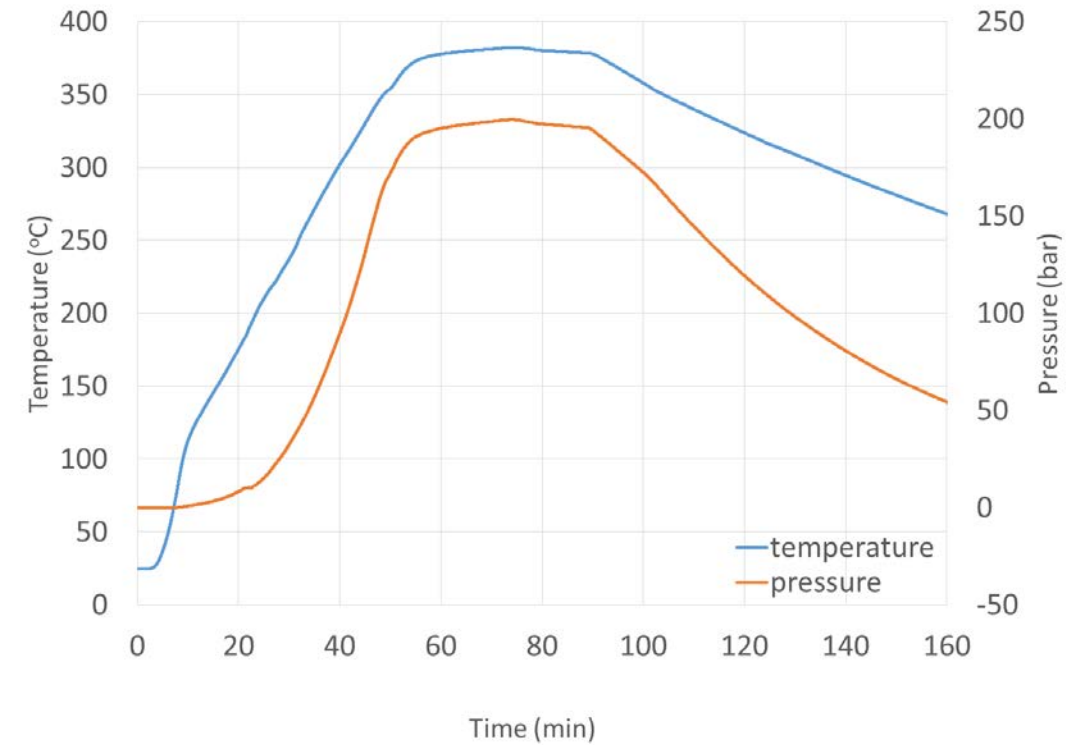


Proof of concept verification (II)

❖ Electric heater



❖ Solar Simulator



- The reactor in Solar Simulator reaches desired temperature and pressure in less than 50% of time compared to the one in the electric heater
- Successful conditions' stabilization using the Solar Simulator



Conclusions & Future work

- ❖ Verification of successful waste feedstock conversion into value added products → **Bio-oil yield up to ~ 50%**
- ❖ HTL-CST coupling significantly speeds up the process and greatly reduces energy consumption
- ❖ Highest bio-oil yields observed in different experimental conditions predominantly depending on feedstocks' main components

Future work

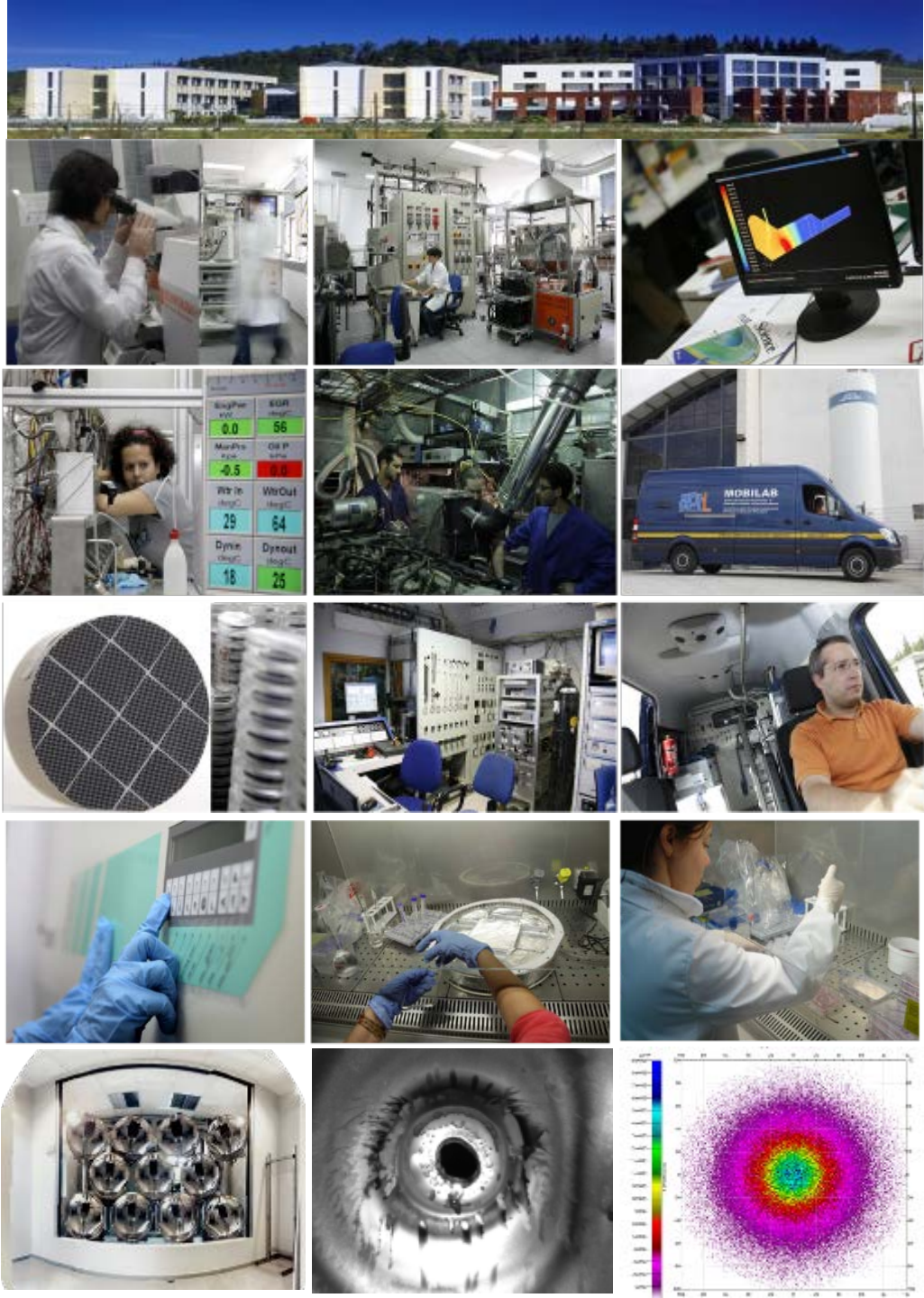
- ❖ Initial pilot-scale experiments utilizing solar energy to take place within coming months
- ❖ Experimental conditions expansion: **presence of catalyst**
- ❖ Study of additional feedstock materials incl. agricultural residues, manure & sludge



Acknowledgments

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Thank you for
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