Advances in Anaerobic Biotechnology: A Core Technology for the Production of Carbon-Neutral, Sustainable and Renewable Biofuels and Bioenergy from Waste

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Anaerobic Digestion & Biotechnological Extensions



Biogas

 $(CH_4, CO_2, trace gases)$

- Volatile fatty acids production & C-chain elongation
- Hydrogen (H₂) production "Dark fermentation"
- Struvite production (MgNP)
- Biochar production
- Bioelectrochemically-assisted anaerobic digestion
- Hydrogen (H₂) production Microbial electrolysis cell (MEC)
- Biogas upgrade using MEC ($CH_4 \ge 98\%$)
- Anaerobic membrane bioreactor

Liquid digestate, Biosolids



Bioelectrochemical Systems (BES)



- A Resistor (MFC) or applied potential (MEC)
- B Proton exchange membrane
- R1 Anode reactant (oxidation half reaction)
- P1 Anode product
- R2 Cathode reactant (reduction half reaction)
- P2 Cathode product

Microbial Fuel Cell (MFC) Produces electrical current

Microbial Electrolysis Cell (MEC) Produces hydrogen (H₂)

Microbial Electromethanogenesis *Produces methane* (CH₄)

Microbial Electrosynthesis (MES) Produces 2+ carbon compounds (e.g., acetate, methanol, etc.)



At 25 °C, 1 atm, pH 7.



Case I: Biomass-derived Biofuels

Overall Objective

H₂ production through the biotransformation of specific furanic and phenolic compounds using MEC technology





H₂ Production – Microbial Electrolysis Cell (MEC)

- <u>Acetate</u>: directly-utilized substrate
- <u>Fermentable, complex organic</u> <u>compounds</u>: fermentation required prior to exoelectrogenesis







Selected Furanic and Phenolic Compounds



- Widely found in hydrolysates and pyrolysates
- Inhibitory to ethanol- and H₂-producing microorganisms in dark fermentation
- Direct conversion to hydrogen in dark fermentation has very low yield



Current and H₂ Production (MEC)



- \checkmark The two furanic compounds and SA were productive substrates for H₂ generation
- \checkmark VA and HBA biotransformation resulted in low current and H₂

Zeng, Collins, Borole, Pavlostathis, Water Research, 2017

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Biotransformation Pathways



Zeng, Collins, Borole, Pavlostathis, Water Research, 2017



Bioanode Conversion -- MEC vs. Fermentation

Furanic and Phenolic Mixture		Sodium Acetate	MEC/Fermentation Studies
sCOD removal (%) Extent of biodegradation	49 - 61	76 - 87	≈ MEC-Domestic WW
Coulombic efficiency (%) (e ⁻ of cumulative current)/(e ⁻ of COD removed)	44 - 69	84 - 95	
H ₂ yield (%) (e ⁻ of H ₂)/(e ⁻ of COD removed)	26 - 42	55 - 58	> Dark Fermentation (~ 17%)
Cathode efficiency (%) (e^{-} of H_{2})/(e^{-} of cumulative current)	65 - 85	66 - 88	NA
Max H ₂ production rate (L/L-d)	0.09 - 0.13	0.08 - 0.14	< Dark Fermentation < MEC-Domestic WW



Bio-Electro-Refinery

Bio-Electro-Refinery: production of switchgrass bio-oil, chemical manipulation followed by phase separation, and chemical production in a BES using the acid-rich aqueous phase (NBOOP: neutralized bio-oil organic phase; NBOAP: neutralized bio-oil aqueous phase)





Case II: MEC Biocathode Conversion of Carbon Dioxide (CO₂) to Methane (CH₄)

Overall Objective

Develop and test a bioelectrochemical system (BES) designed to convert CO₂ to CH₄ for the purpose of increasing the energy content of anaerobic digester biogas (i.e., biogas upgrade)





Biocathode Performance – Effect of Inoculum

Biocathode methanogenic inocula: MM, mixed; EHM, pre-enriched hydrogenotrophic

EHM-inoculated



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Dykstra and Pavlostathis, Environ. Sci. Technol., 2017

 0.64 ± 0.19

 0.59 ± 0.03

Biocathode Performance – Effect of H₂S



Cathode Headspace H₂S (0-6% v/v)

INITIAL 3-DAY CH ₄ PRODUCTION RATE (L/m ³ -d)	100 80 60 40 20	- - - - -							
	v	0	1	2	3	4	5	6	
INITIAL CATHODE HEADSPACE H ₂ S (%)									

Headspace H ₂ S (% v/v)	CE (%)	CCE (%)
0	11	100
0		100
4	19	99
5	58	13
6	58	15

CE, Coulombic efficiency CCE, cathode capture efficiency

Two competing effects:

- Depression of CH₄ production (H₂S ≥4%): Inhibition of methanogens?
- Enhancement of CH₄ production (H₂S ≤3%): What is/are the process(es) involved?

Biocathode Performance – Effect of H₂S



The methanogenic biocathode is protected from sulfide inhibition by **biofilm** formation and a local higher pH at the cathode electrode surface.



Biocathode Performance – Effect of H₂S

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BES Performance – Cathode Potential



APPLIED POTENTIAL (V vs. SHE)

Cathode Potential	CH₄ Productio	Final Biocathode Biogas (%)		
(V vs. SHE)	n Rate (mmol/d)	CH₄	CO ₂	
-0.80	1.22	96	4	
-0.75	0.98	95	5	
-0.70	0.87	94	6	
-0.65	0.97	94	6	
-0.60	0.74	92	8	
-0.55	0.86	92	8	
-0.50	0.53	90	10	

Georgia

Dykstra and Pavlostathis, Water Research, 2021

- At a more positive applied cathode potential, the cell potential (driving force for electron transport) decreased and the anode potential decreased.
- At lower anode potentials, the transfer of electrons from a substrate to the anode is less energetically favorable.
- However, anode acetate removal did not reflect the biocathode CH₄ production rate, likely due to microbial acetate uptake and storage.



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Biogas Utilization – CO₂ Recycle/Zero-net Carbon Products





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