





ЕВРОПЕЙСКИ СЪЮЗ ЕВРОПЕЙСКИ ФОНД ЗА РЕГИОНАЛНО РАЗВИТИЕ clean





# New glass ceramic by Fe Ni wastes with improved structure and properties

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## Content :

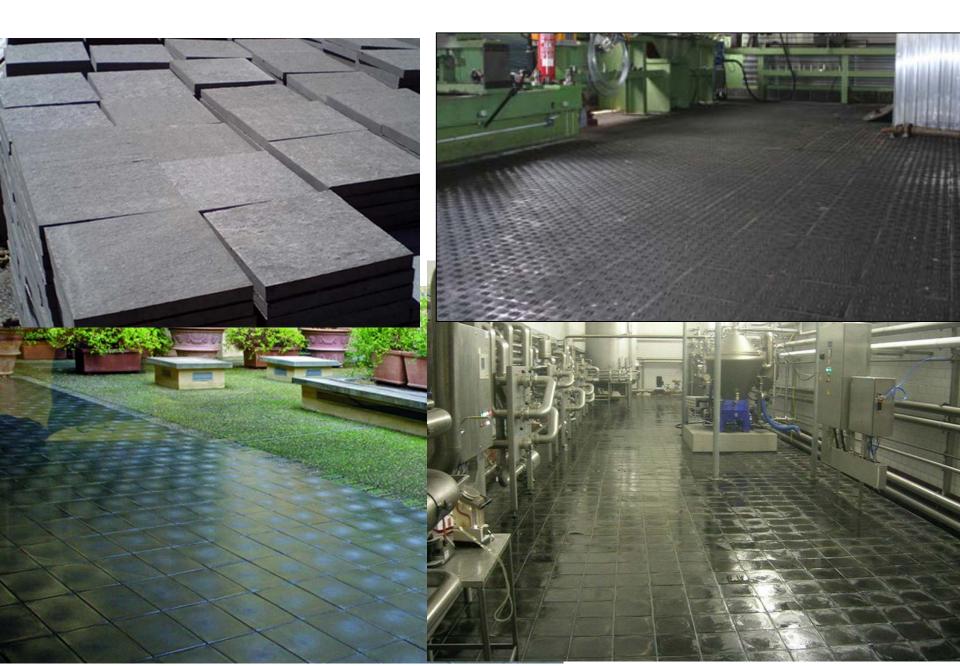
- Basal tiles and petrurgical products
- Used wastes and parent glass
- Evaluation of the crystallization heat-treatment regime
- Morphology and structure by SEM and TEM
- Properties of obtained glass-ceramic

Main techniques and apparatus: DTA-TG (Perkin Elmer Diamond), Density measurements (AccuPyc 1330), HSM (Misura 1400), XRD (PANalytical Empyrean), TEM (JEOL 200 kV), SEM-EDS (JEOL JSM 6390) and FESEM (Zeiss Merlin)

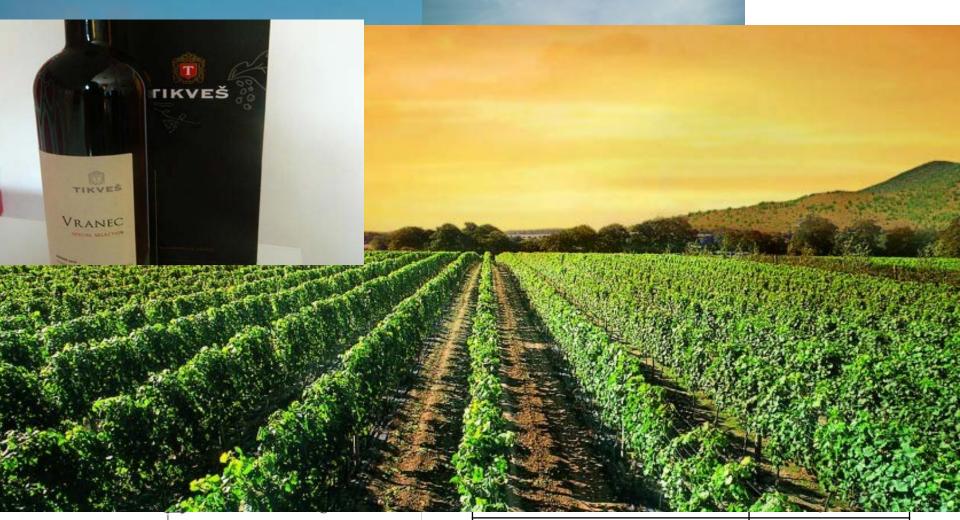
## Basalt road tiles – Roman Empire and Napoli – Italy (today)

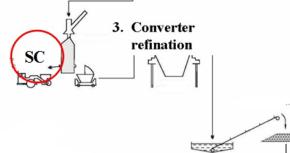


## **Products of the petrurgy - cast basaltic tiles**



#### Analysis of used wastes and parent glass





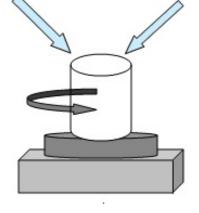
Dust (FD)	102 000
Electro-furnace slag (SEF)	1 135 000
Converter slag (SC)	109 000

Analysis of used wastes and parent glass

## Leaching test (TCLP) results for the used wastes (mg/l)

Solid waste

CH₃COOH



– CH<sub>3</sub>COOH,

 $- pH \leq 5$ ,

– **24 h**,

- **30 rpm**,

- *s*:/ = 1:20

	SEF	FD	SC	Applied limits
Zn	0.29	3.6	0.42	5
Ni	2.7	6.1	19.2	1
As	0.01	1.2	0.18	0.2
Cr	1.1	0.22	2.9	1.0

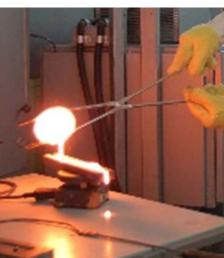
#### Analysis of used wastes and parent glass

#### **Chemical compositions**

	SEF	FD	SC	WM	GC	WG
SiO <sub>2</sub>	53	37.5	1.9	47.8	71.4	55
Al <sub>2</sub> O <sub>3</sub>	2	1.8	0.3	1.9	0.6	1.5
MgO	16.9	14.5	6.2	15.9	3.3	12.1
CaO	2.4	2.3	15.9	3.5	9.8	5.4
Cr <sub>2</sub> O <sub>3</sub>	2.5	1	0.7	2.2		1.5
СоО	0.1	0.1	0.1	0.1		0.1
NiO	0.1	2.7	0.45	0.3		0.2
Fe <sub>2</sub> O <sub>3</sub>	14	30	60	19		19.8*
FeO	9		19	8.8		
Na <sub>2</sub> O					13.3	4
K <sub>2</sub> O					1.3	0.4
Wt %	58	6	6	70	30	100

Melting procedure:

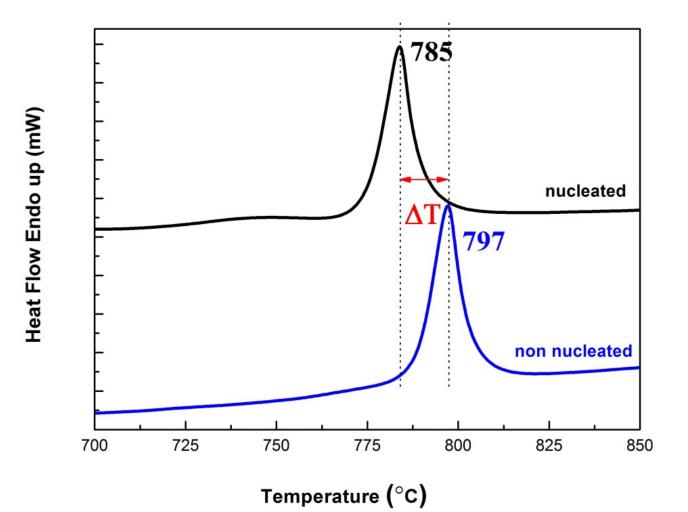
- ~ 0,5 kg glass
- 1400 °C
- 1.5 h



#### **Evaluation of the heat-treatment regime**

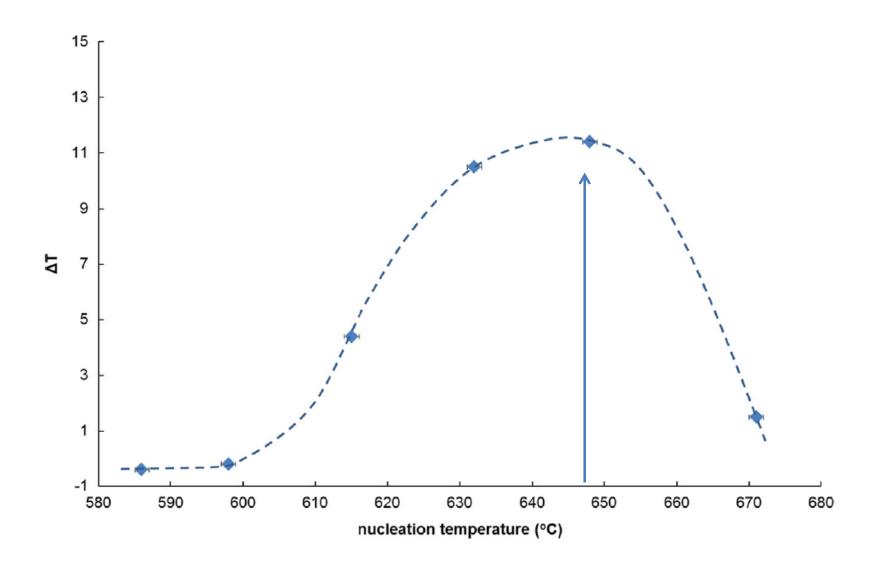
#### Evaluation of the optimal nucleation treatment by DTA

12-15 mg bulk samples, treated at 20 °C/min after various nucleation treatment (temperature and time) The maximum temperature shift (optimal nucleation) was obtained after **45-60 min at 650** °C.



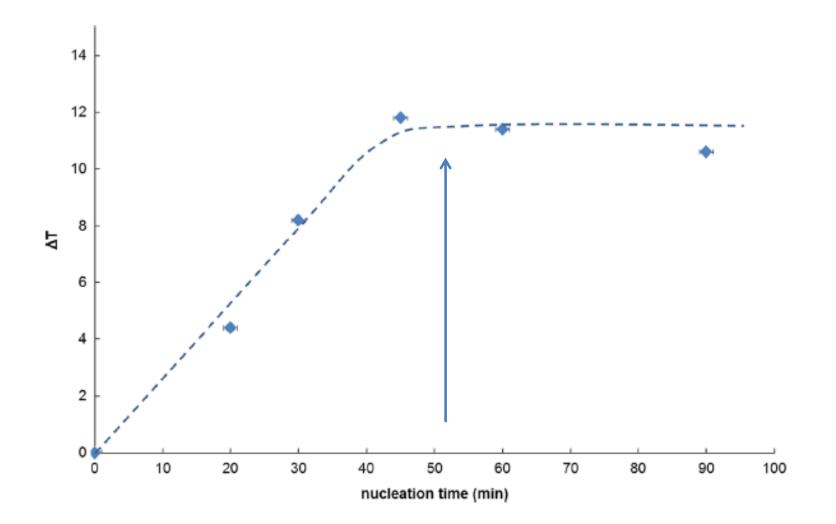
#### Evaluation of the optimal nucleation treatment by DTA

#### **Optimal nucleation temperature**



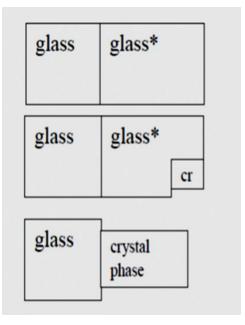
#### Evaluation of the optimal nucleation treatment by DTA

#### **Optimal nucleation time**



#### Evaluation of the heat-treatment regime

#### Evaluation of the optimal crystallization treatment by density measurements

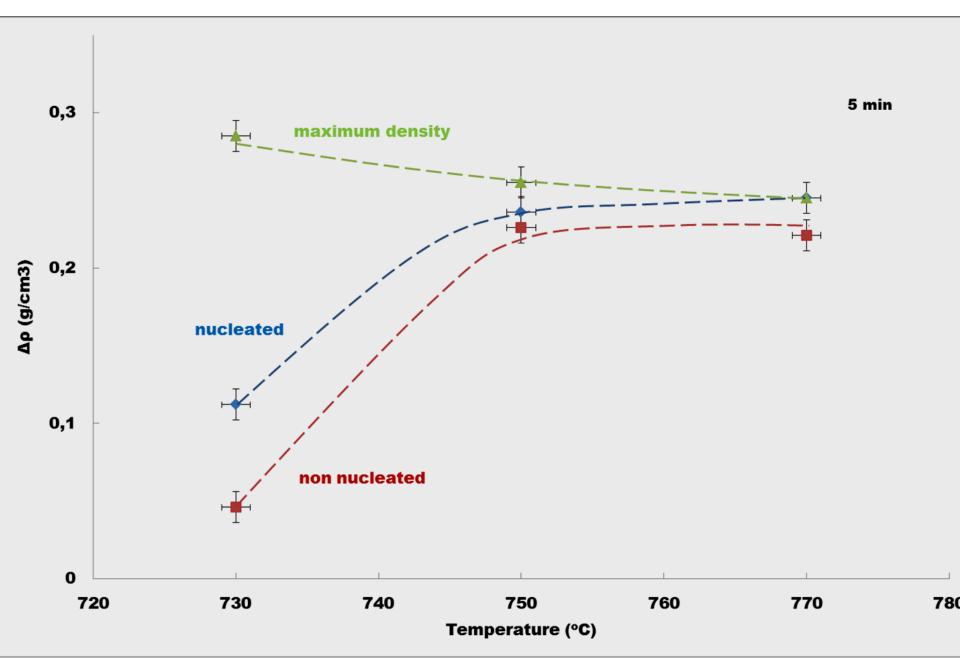


vol % crystal phase = 100 \*  $\frac{\frac{1}{\rho_{pg}} - \frac{1}{\rho_{gc}}}{k}$   $\rho_{pg}$  - parent glass density  $\rho_{gc}$  - glass-ceramic density  $k = \frac{1}{\rho_{g*}} - \frac{1}{\rho_{cr}}$   $\rho_{cr}$  - crystal phase density  $\rho_{g*}$  - density of a glass with composition of the formed crystal phase

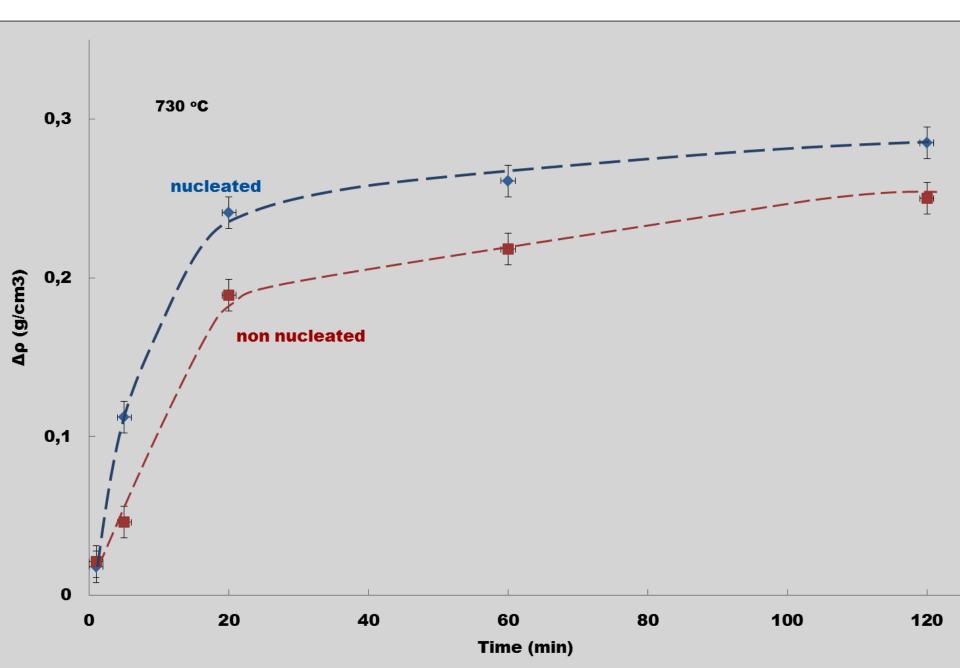
In the case of pyroxene crystallization $\rho_{g*}/\rho_{cr} \approx 0.84 \pm 0.1$ In the case of spinel crystallization $\rho_{g*}/\rho_{cr} \approx 0.82 \pm 0.2$ 

In our experiments:  $\rho_{pg} - 2.91 \pm 0.006 \text{ g/cm}^3$   $\rho_{gc} - 3.19 \pm 0.004 \text{ g/cm}^3$ , reached after **45-60 min at 730-740** ° **C** (corresponding to total crystallinity of 58 ± 2)

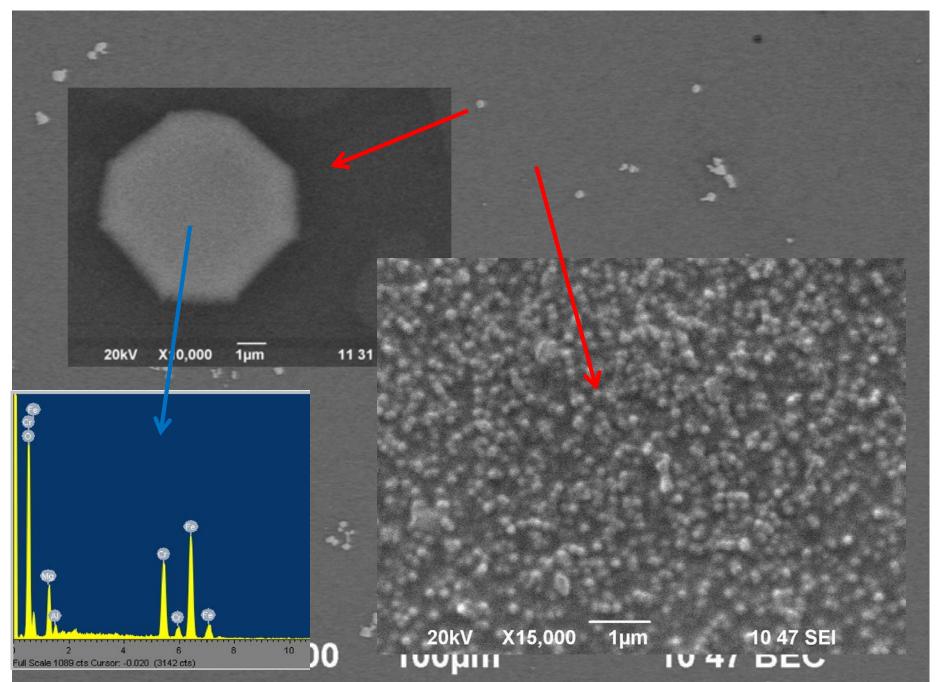
#### **Evaluation of the optimal crystallization treatment**



#### **Evaluation of the optimal crystallization treatment**



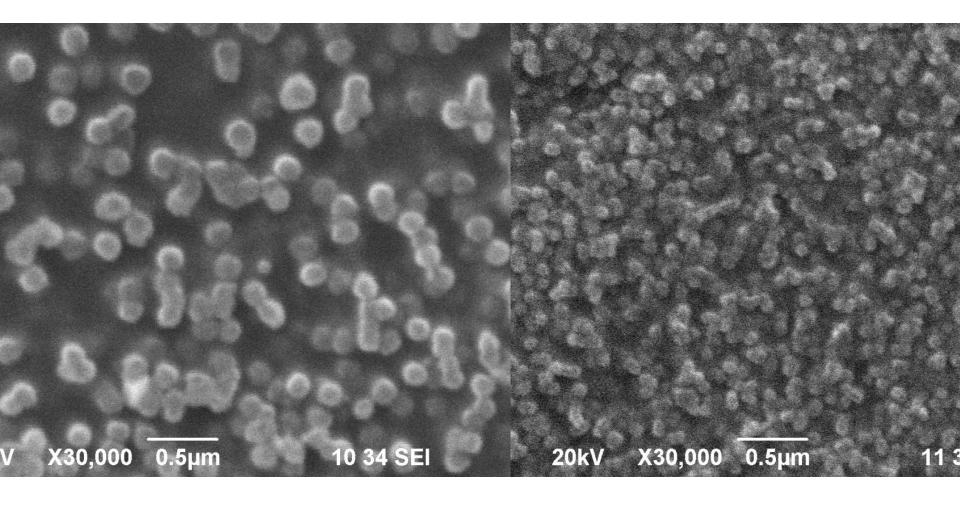
**SEM - 1** min at 730 °C



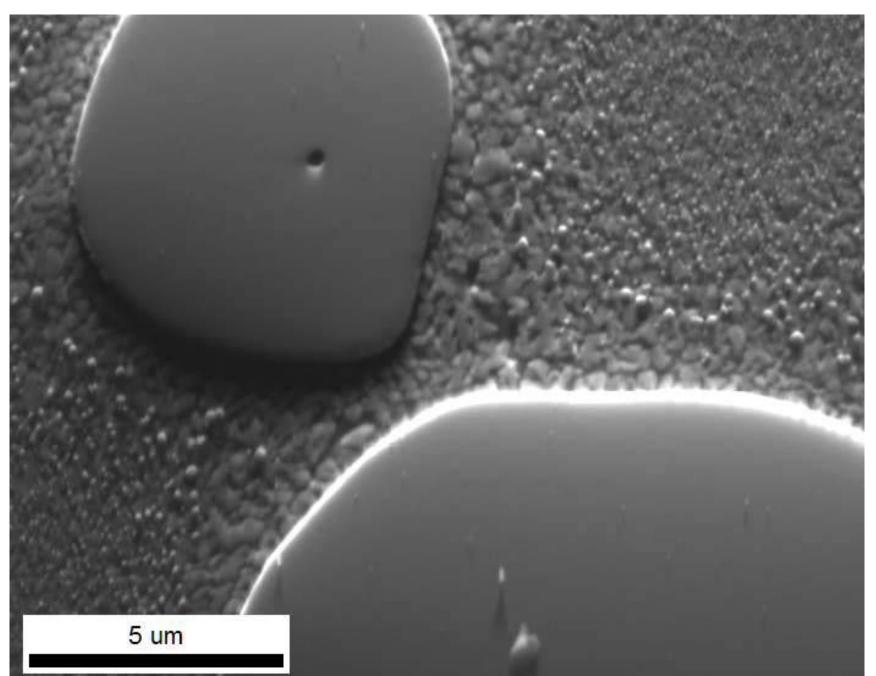
SEM

#### 1 min at 730 °C

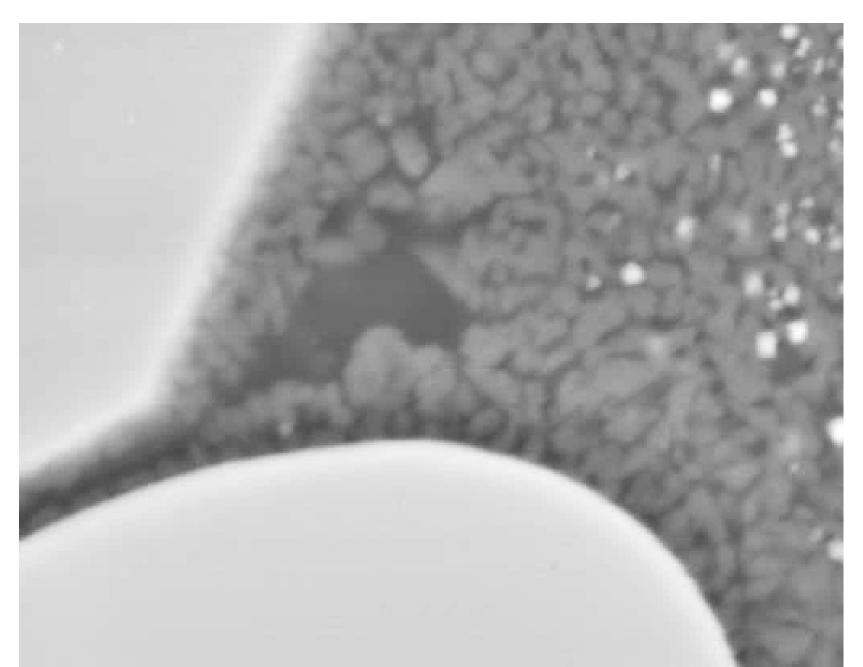
#### 60 min at 650 °C and 1 min at 730 °C



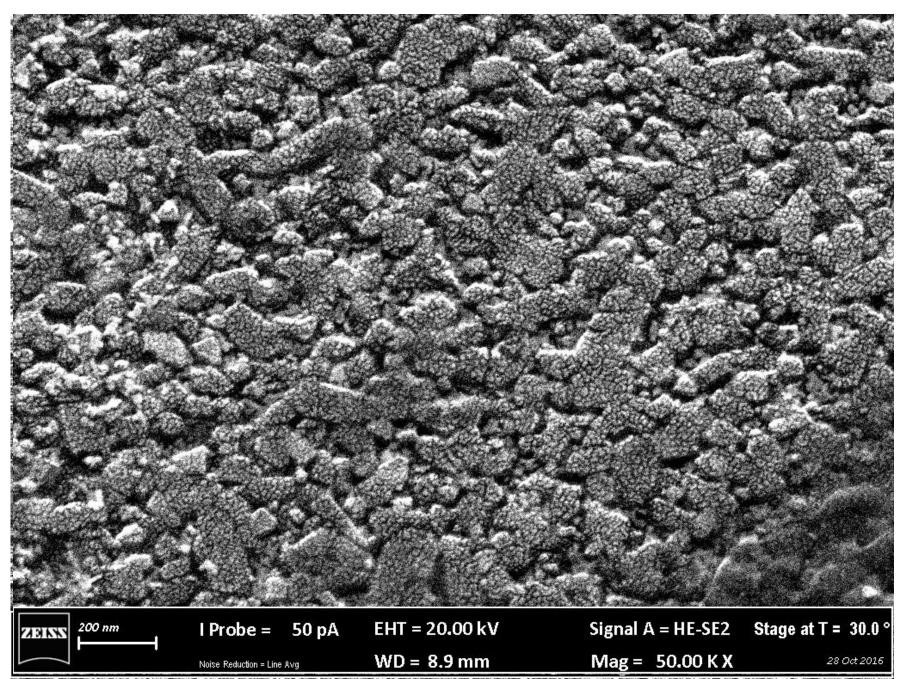
#### FESEM - 60 min at 650 °C and 60 min at 730 °C



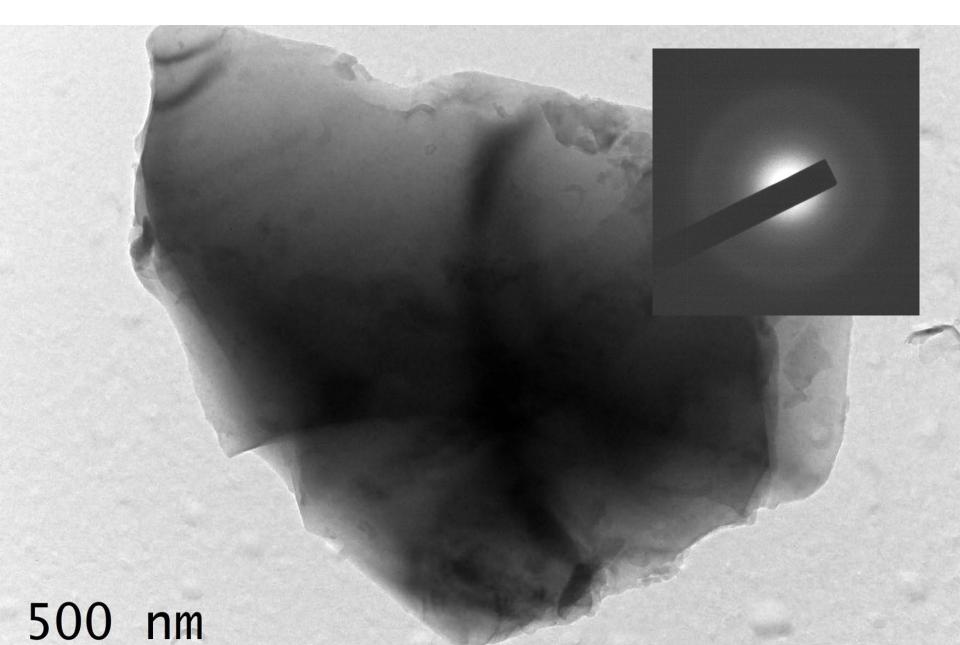
#### **FESEM** - 60 min at 650 °C and 60 min at 730 °C



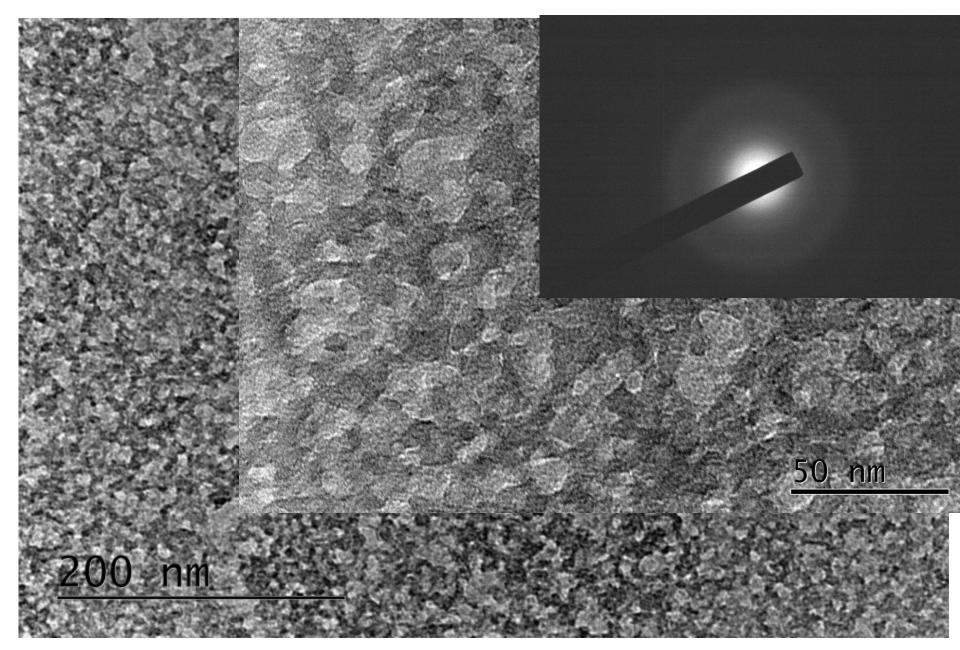
## Morphology and structure FESEM - 60 min at 650 °C and 60 min at 730 °C



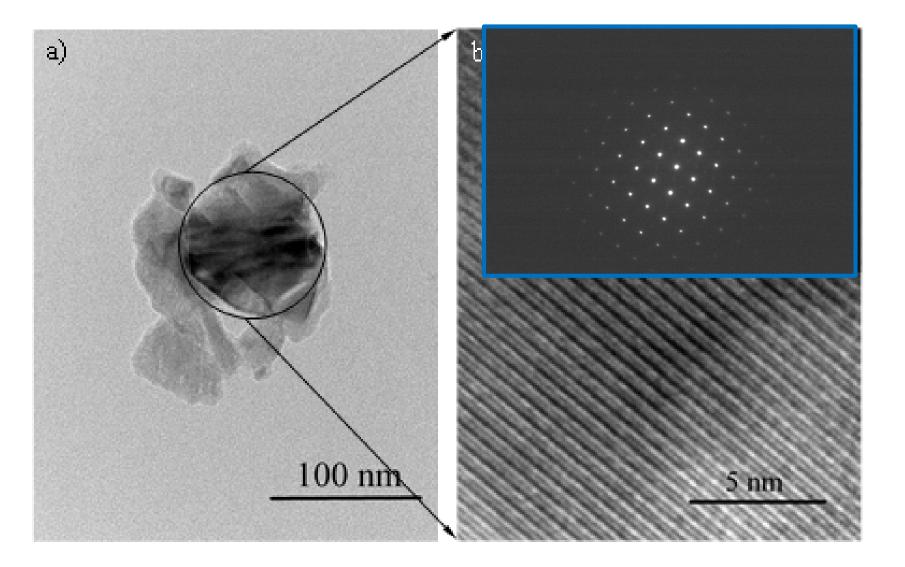
## **TEM – initial glass**



## **TEM** - 60 min at 650 °C and 1 min at 730 °C



#### **TEM** - 60 min at 650 °C and 60 min at 730 °C



## Leaching test (TCLP) results of the used wastes and obtained glass and glass-ceramic (mg/l)

	SEF	FD	SC	G	GC	Applied limits
Zn	0.29	3.6	0.42	<0.005	<0.005	5
Ni	2.7	6.1	<b>19.2</b>	0.311	0.098	1
As	0.01	1.2	0.18	0.009	0.003	0.2
Cr	1.1	0.22	2.9	<0.005	<0.005	1.0

## properties of obtained glass-ceramic

Properties	Slag Sitalls	FeNi GC
Density (g/cm <sup>3</sup> )	2.7-2.9	3.1
Coefficient of linear thermal expansion 20-400°C (10-7/C)	75-90	71
Bending strength (MPa)	80-110	120
<b>Compressive strength (MPa)</b>	250-350	280
Vickers Hardness (GPa)	7.5-9	9
Fracture toughness (MPa•m1/2)	1.2-1.5	1.6
Main technological parameters		
Melting Temperature (°C)	1450-1550	1350-1400
Nucleation step	1-2 h at 750-800 °C	0.5-1h at 650 °C
Crystallization step	2-3 h at 850-950 °C	0.5-1h at 750-760 °C

## **Conclusions:**

liquid-<sup>I;</sup>

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- The parent glass is obtained using huge amount of hazardour from ferronickel production at moderate melting temper
- The optimal low temperature crystallization regim about 650 °C and 30-45 min crystallization at by alternative fast methods: the nucleatic while the crystallization step – by pyc<sup>2</sup>
- The phase formation starts dur preliminary Fe-Mg-Cr spin epitaxial growth of pv<sup>r</sup> arrangement.

.eation at , was estimated ...ed by DTA analysis, inents.

ing with the precipitation of nese crystals act as centers for s is formed a part of the glass-ceramic

The main s<sup>+</sup> 30-85 vol %) is a result of the tendency for binodal . It leads to the formation of secondary magnetite J-70 nm size and to crystallization of main phase in glass-.ed by a new pyroxene with crystal size of 200-300 nm.

Jass-ceramic is characterized by a high chemical durability and -d mechanical characteristics.