



# Unraveling HTL mechanisms of carbohydrates-proteins interactions

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# Overview

1

## Introduction

HTL in a nutshell



2

## Materials & Methods

Experiments  
organization and  
work-up



3

## Results

Main results



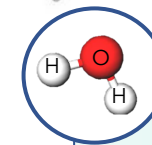
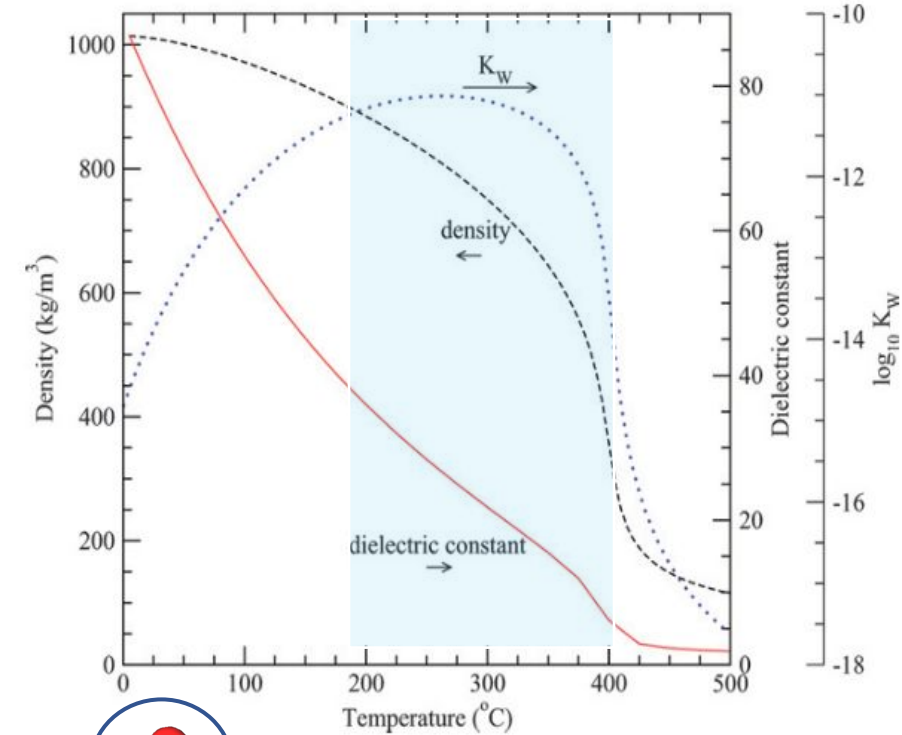
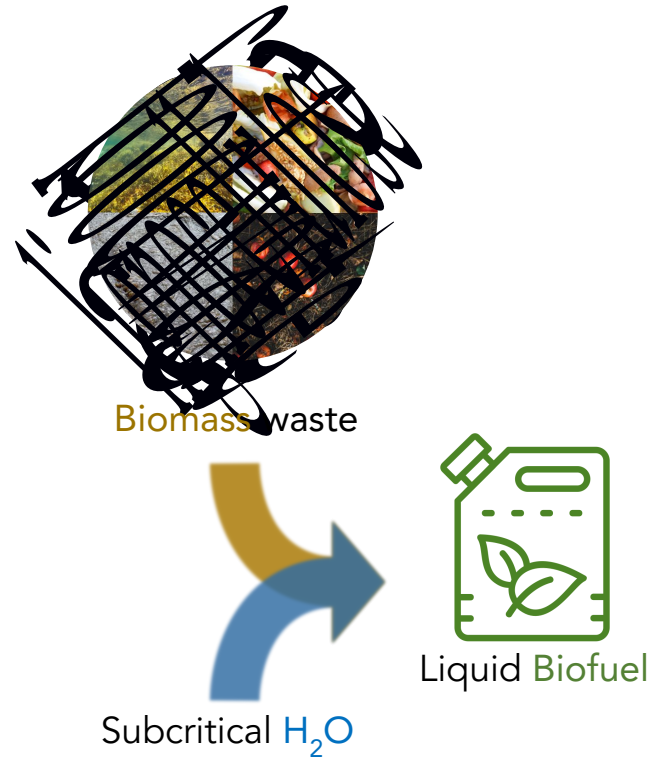
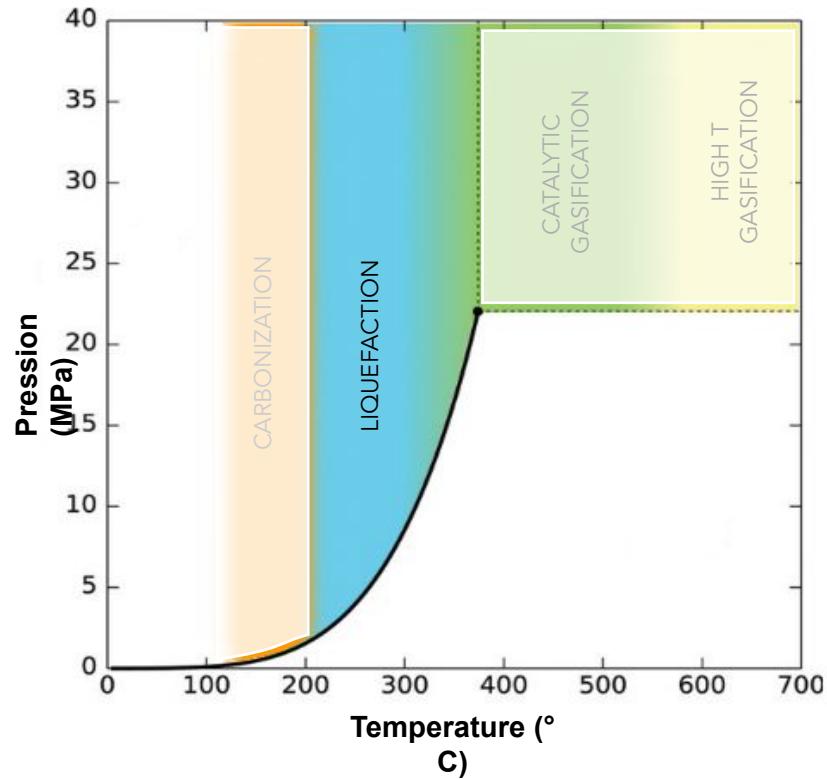
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## Conclusions

Conclusions  
Take-home messages



# Hydrothermal liquefaction (HTL)

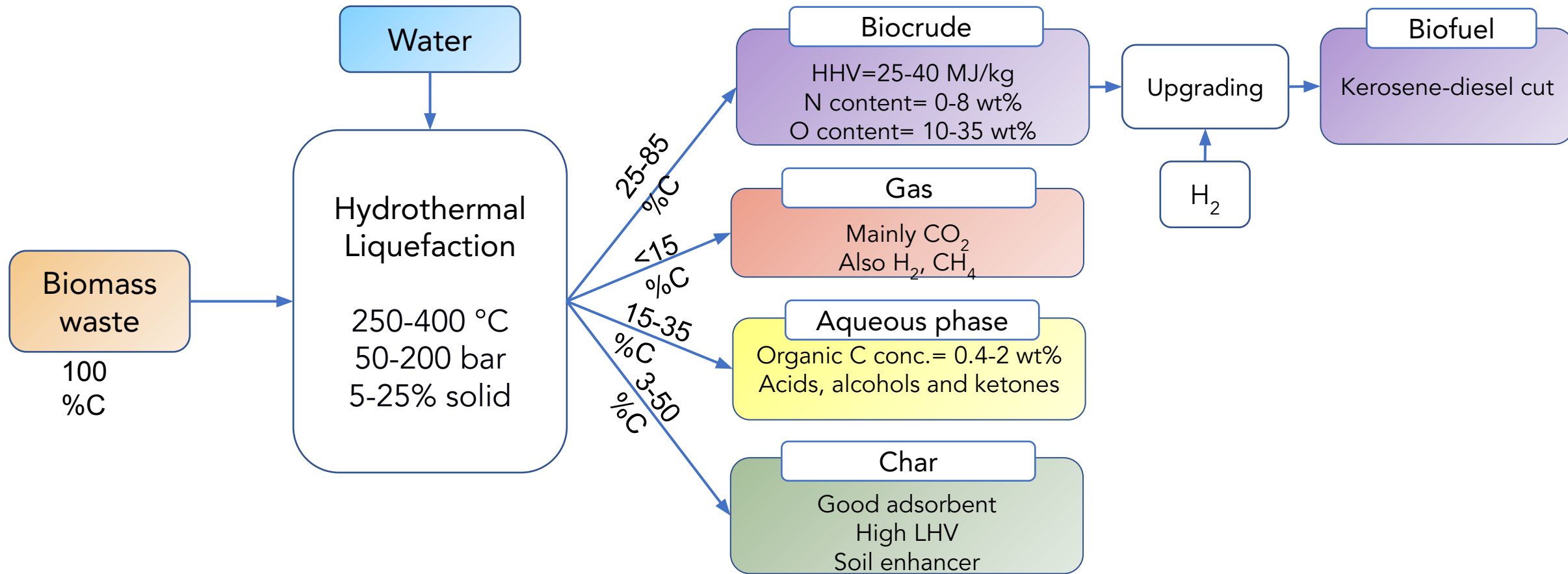


Non-toxic or hazardous  
Good biomass solubility  
Reactive

- ✓ 5-20 wt.% solid concentration
- ✓ No need for drying → high energy efficiency
- ✓ High flexibility

Peterson, Andrew A. et al. 2008. "Thermochemical Biofuel Production in Hydrothermal Media: A Review of Sub- and Supercritical Water Technologies." *Energy and Environmental Science* 1(1): 32–65.

# Hydrothermal liquefaction (HTL)

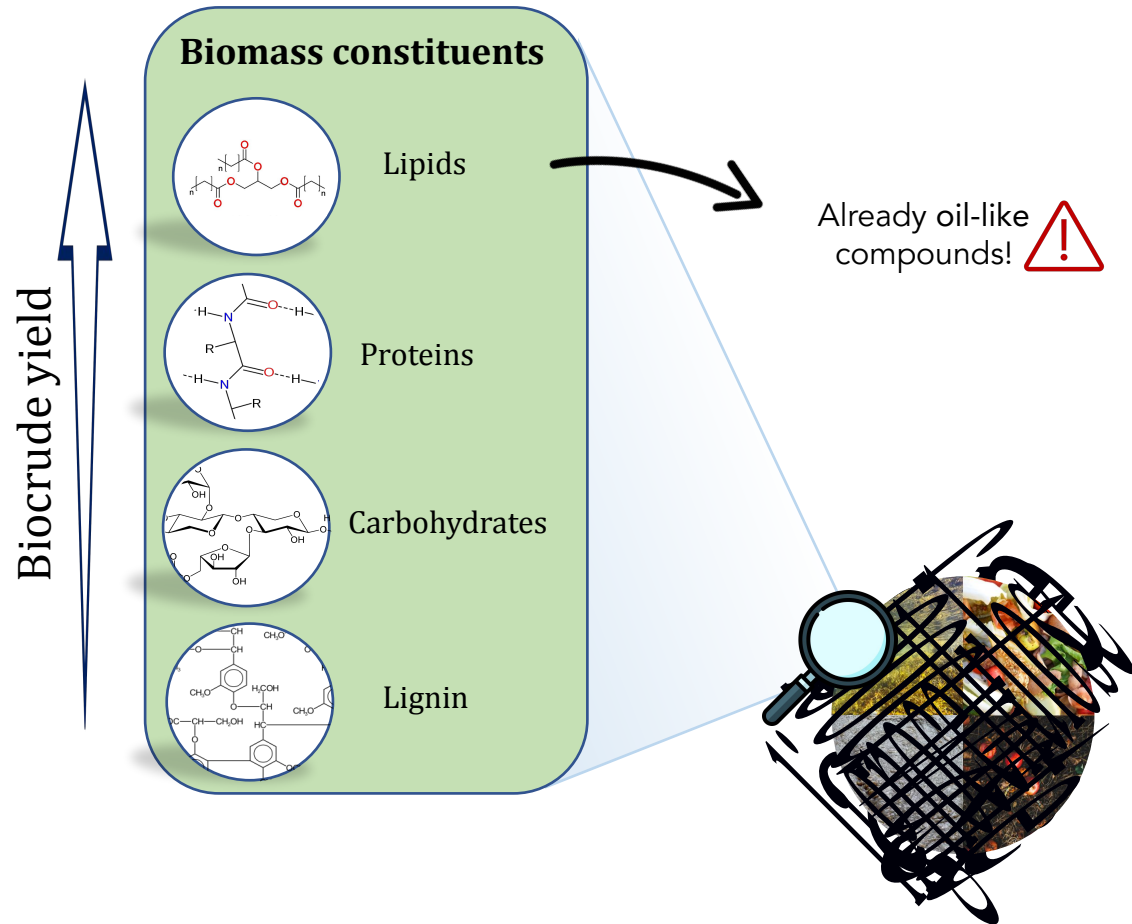


Lu, Jianwen et al., 2022. "Elemental Migration and Transformation during Hydrothermal Liquefaction of Biomass." *Journal of Hazardous Materials* 423(PA): 126961.

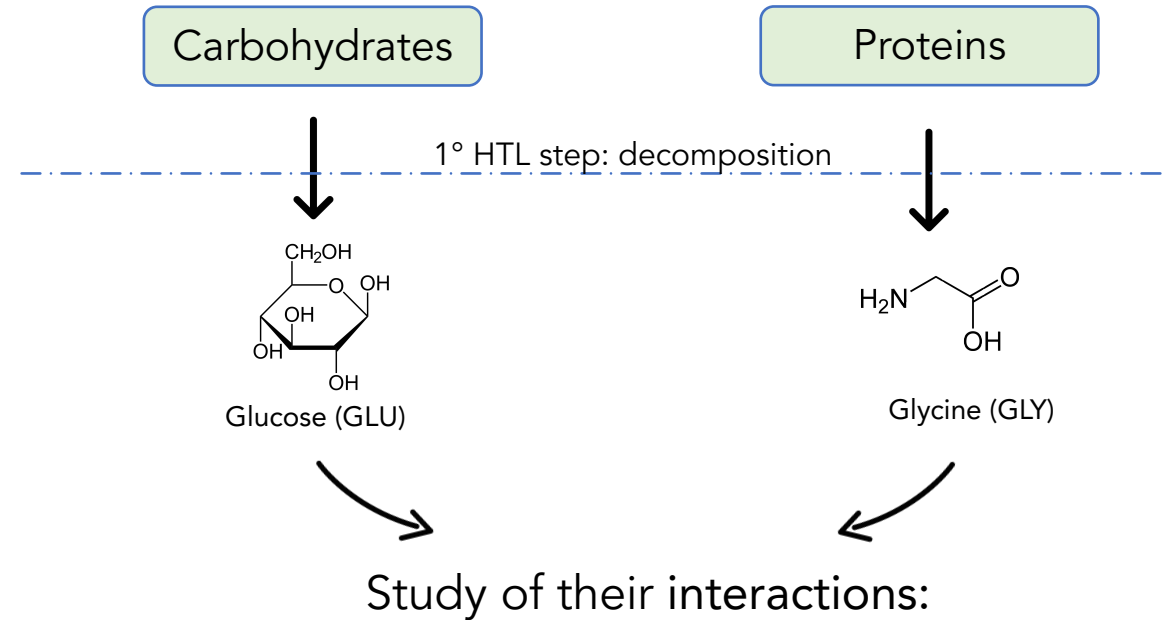
Panisko et al. 2015. "Characterization of the Aqueous Fractions from Hydrotreatment and Hydrothermal Liquefaction of Lignocellulosic Feedstocks." *Biomass and Bioenergy* 74: 162–71.

Mathanker et al. 2021. "A Review of Hydrothermal Liquefaction of Biomass for Biofuels Production with a Special Focus on the Effect of Process Parameters, Co-Solvents, and Extraction Solvents." *Energies* 14: 4916

# Aim of the work



Proteins-carbohydrates biomass represents «*lower value*» biomass



- ✓ Varying the starting composition (GLU, GLY, GLU-GLY)
- ✓ Varying the operative temperature (200-350 °C)

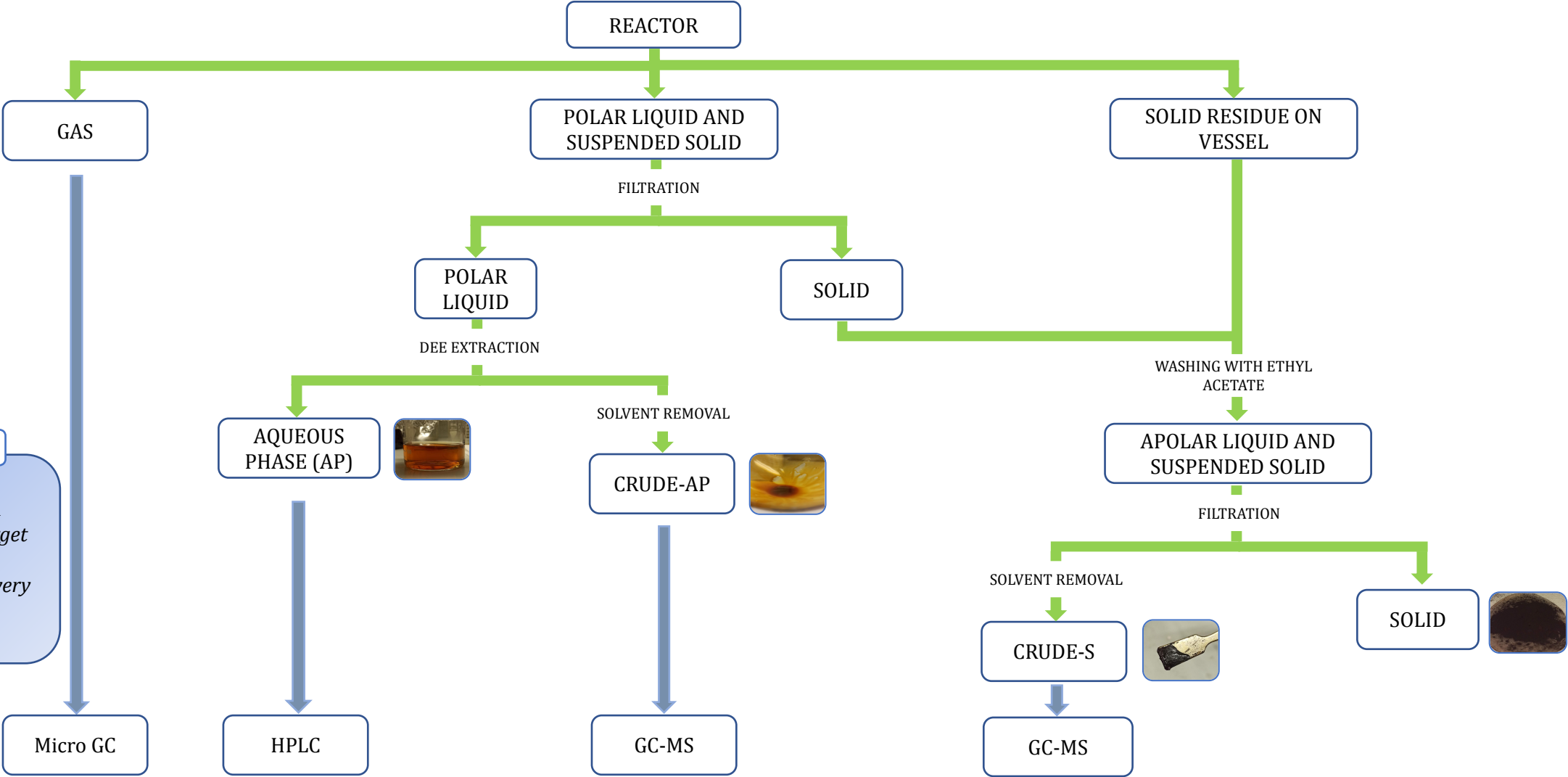
# Work-up

Commercial Parr reactor



Analyzers

**Micro GC:** gas composition  
**HPLC:** concentration of *target* compounds in **AP**  
**GC-MS:** concentration of *every* compounds in **OIL**



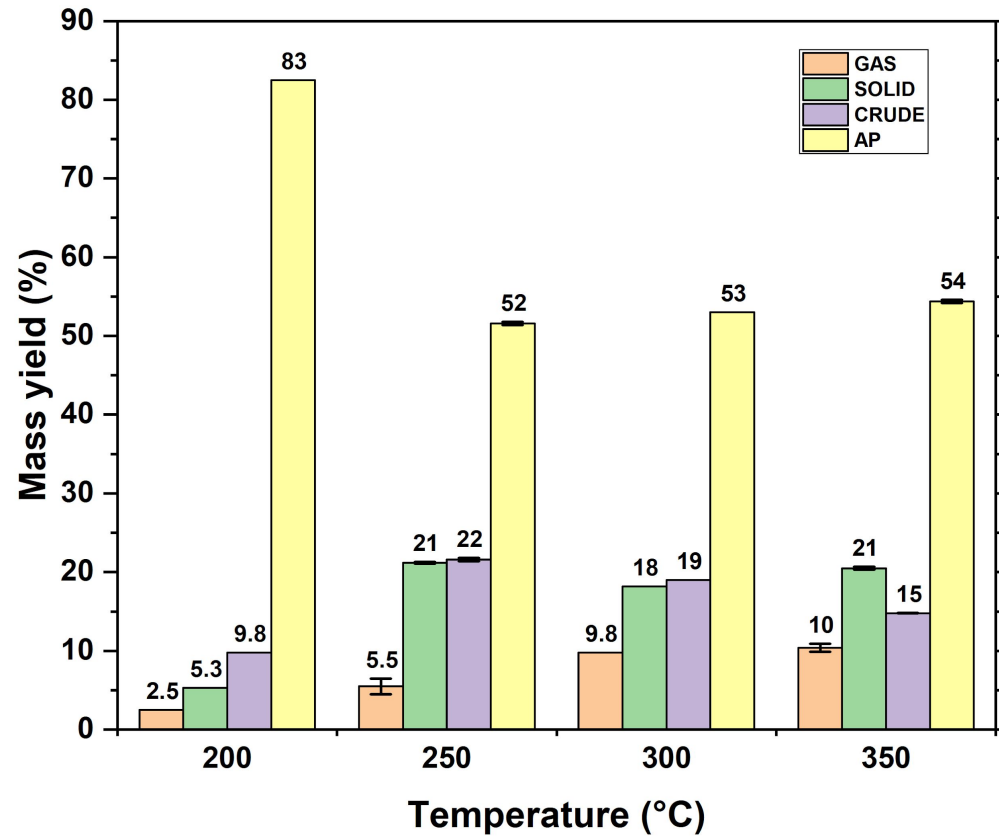
## Definitions

$$\text{Mass yield} = \frac{\text{phase mass}}{\text{glucose mass}}$$

$$\text{Mass yield AP} = 1 - \sum_{\text{other phases}} \text{Mass yield}$$

# Mass yields: glucose alone

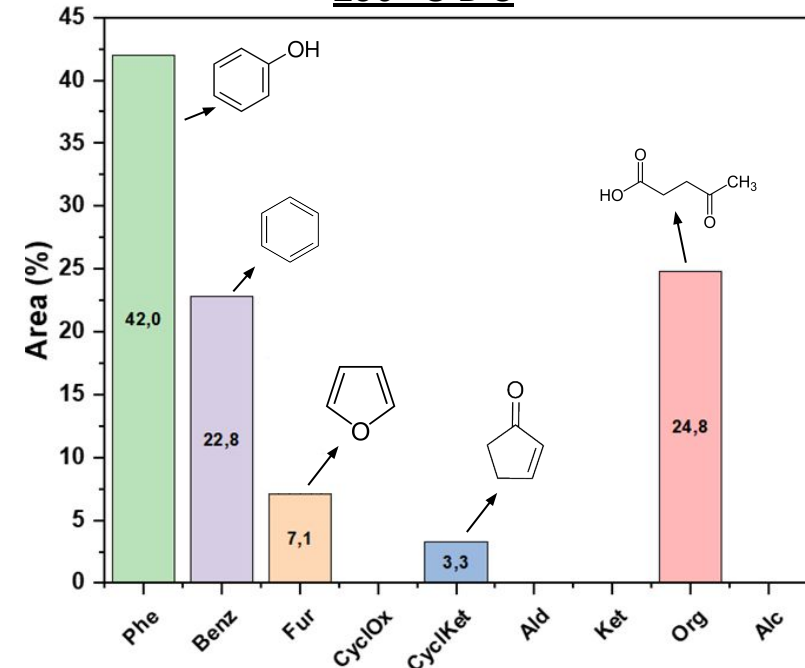
## 5% GLU



Reaction conditions: 5 wt.% glucose, 60 min, 2 barg initial pressure, 200 g feed, 570 ml vessel volume.

- Increasing **GAS** → mainly CO<sub>2</sub>, lower CO and H<sub>2</sub> at high T
- Constant **SOLID**
- Max at 250 °C for **CRUDE** → mainly benzenes, phenols and levulinic acid  
Possible instability at high T

## 250 °C B-S



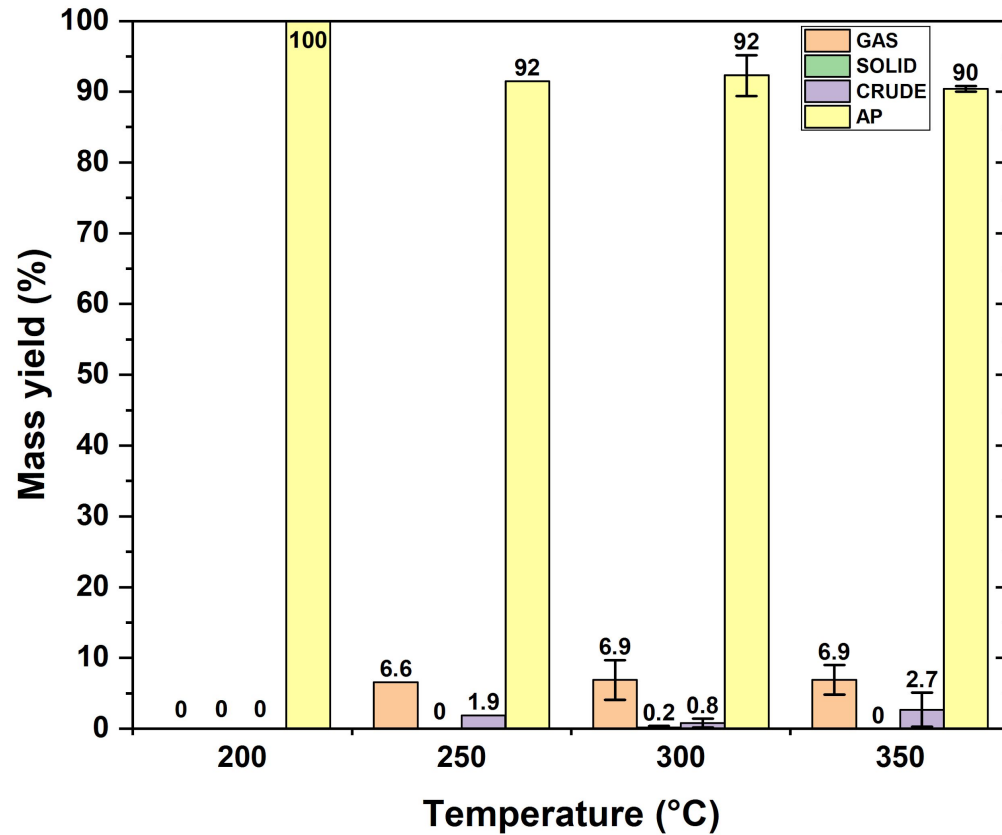
## Definitions

$$\text{Mass yield} = \frac{\text{phase mass}}{\text{glycine mass}}$$

$$\text{Mass yield AP} = 1 - \sum_{\text{other phases}} \text{Mass yield}$$

# Mass yields: glycine alone

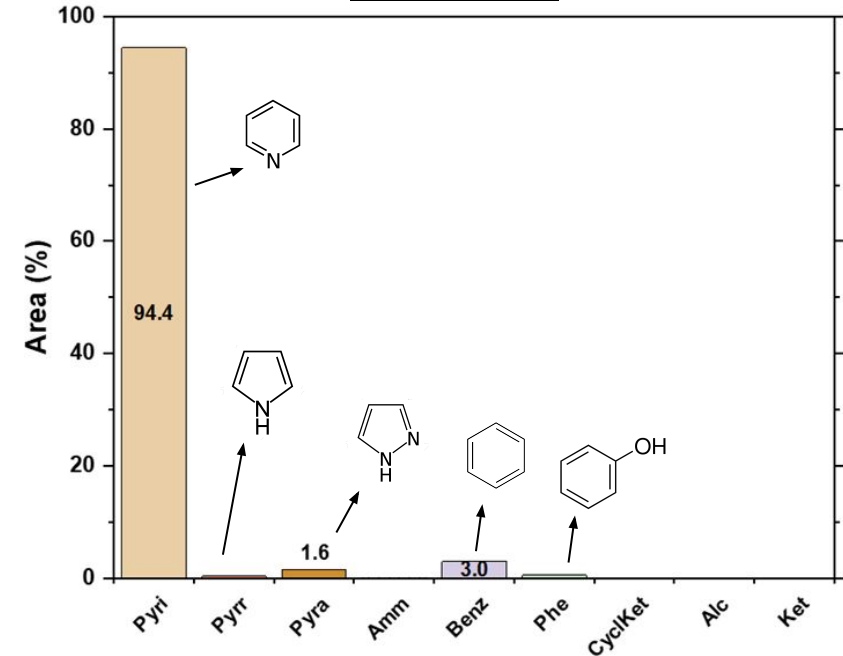
## 5% GLY



Reaction conditions: 5 wt.% glycine, 60 min, 2 barg initial pressure, 200 g feed, 570 ml vessel volume.

- Constant **GAS** → mainly CO<sub>2</sub>, lower H<sub>2</sub> and CO
- No **SOLID**
- Very low **CRUDE** → mostly pyridines c1ccncc1

## 350 °C B-AP





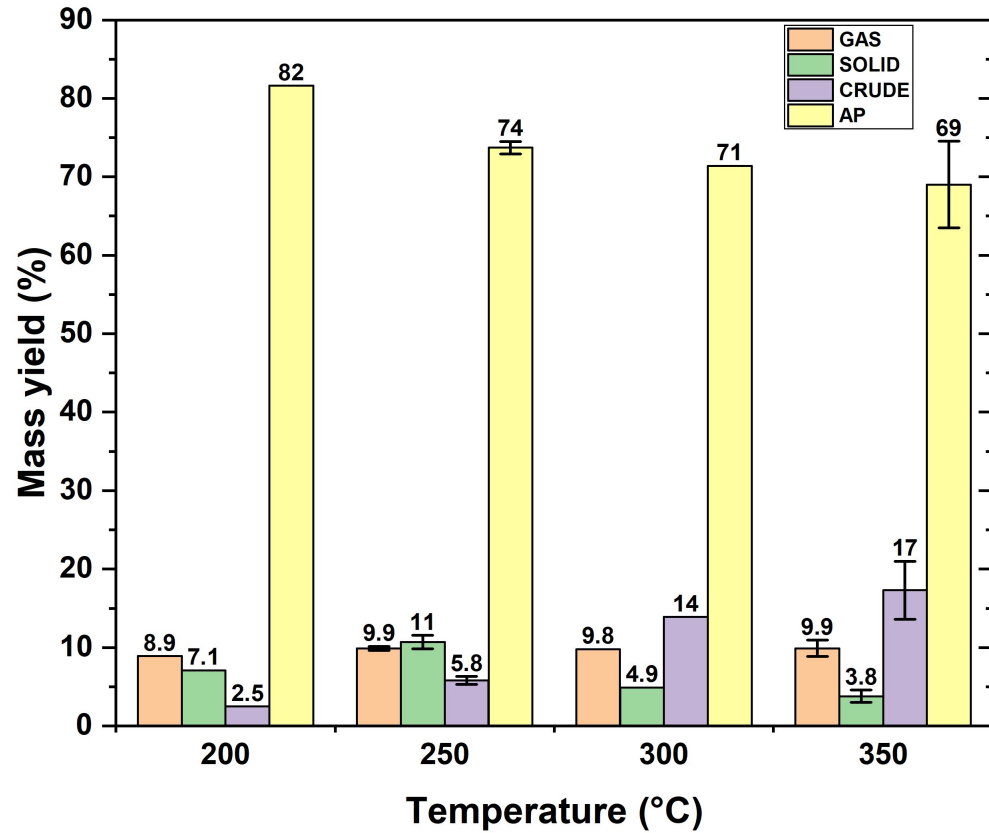
## Definitions

$$\text{Mass yield} = \frac{\text{phase mass}}{\text{glucose mass} + \text{glycine mass}}$$

$$\text{Mass yield AP} = 1 - \sum_{\text{other phases}} \text{Mass yield}$$

# Mass yields: glucose-glycine mixture

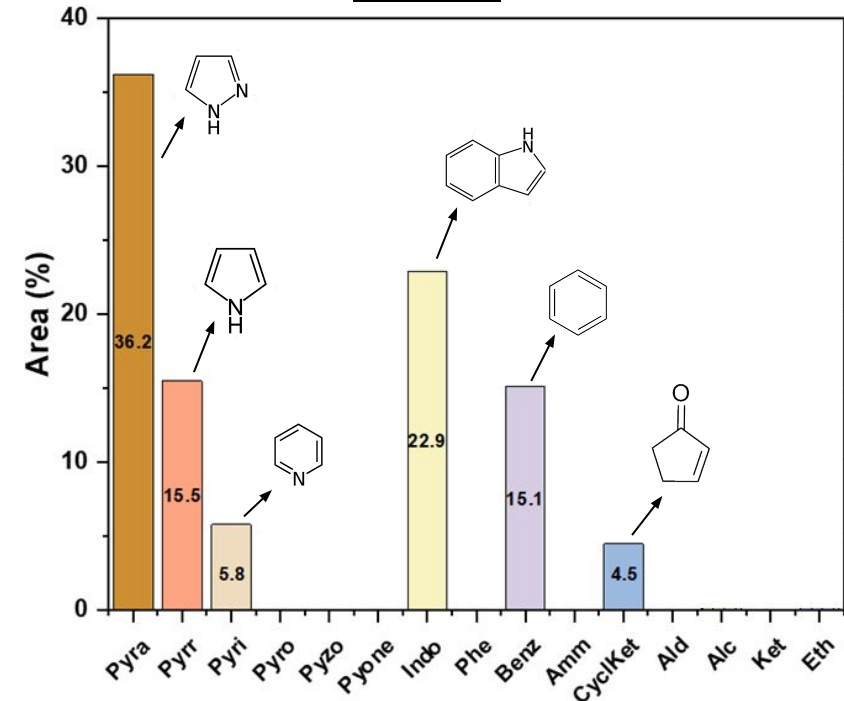
## 5% GLU - 5% GLY



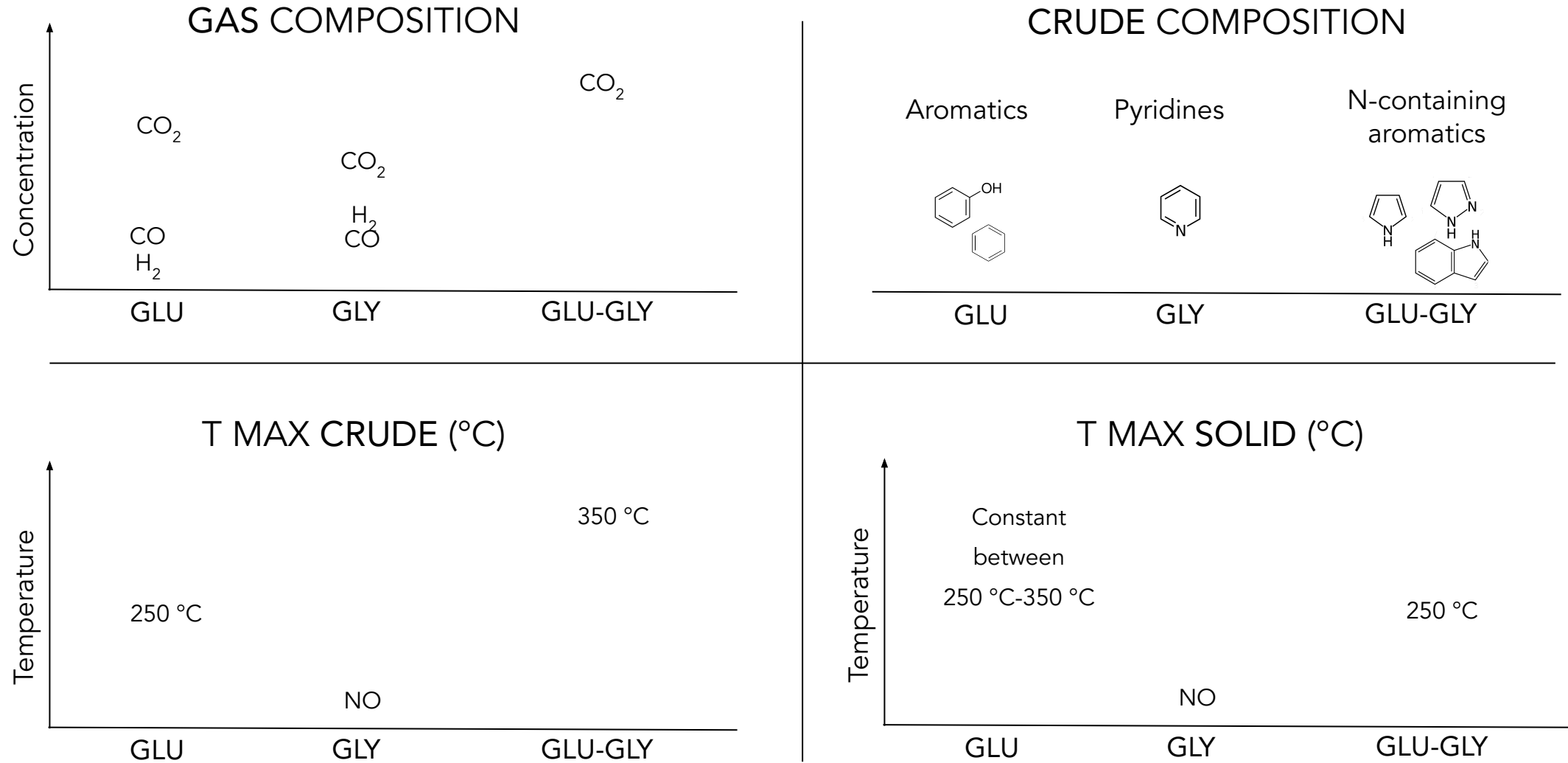
Reaction conditions: 5 wt.% glucose-5 wt.% glycine, 60 min, 2 barg initial pressure, 200 g feed, 570 ml vessel volume.

- Constant **GAS** → only CO<sub>2</sub>
- Max at 250 °C for **SOLID** → seems converted to **CRUDE**
- Increasing **CRUDE** → mainly nitrogen-containing aromatics

## 350 B-S



# Differences



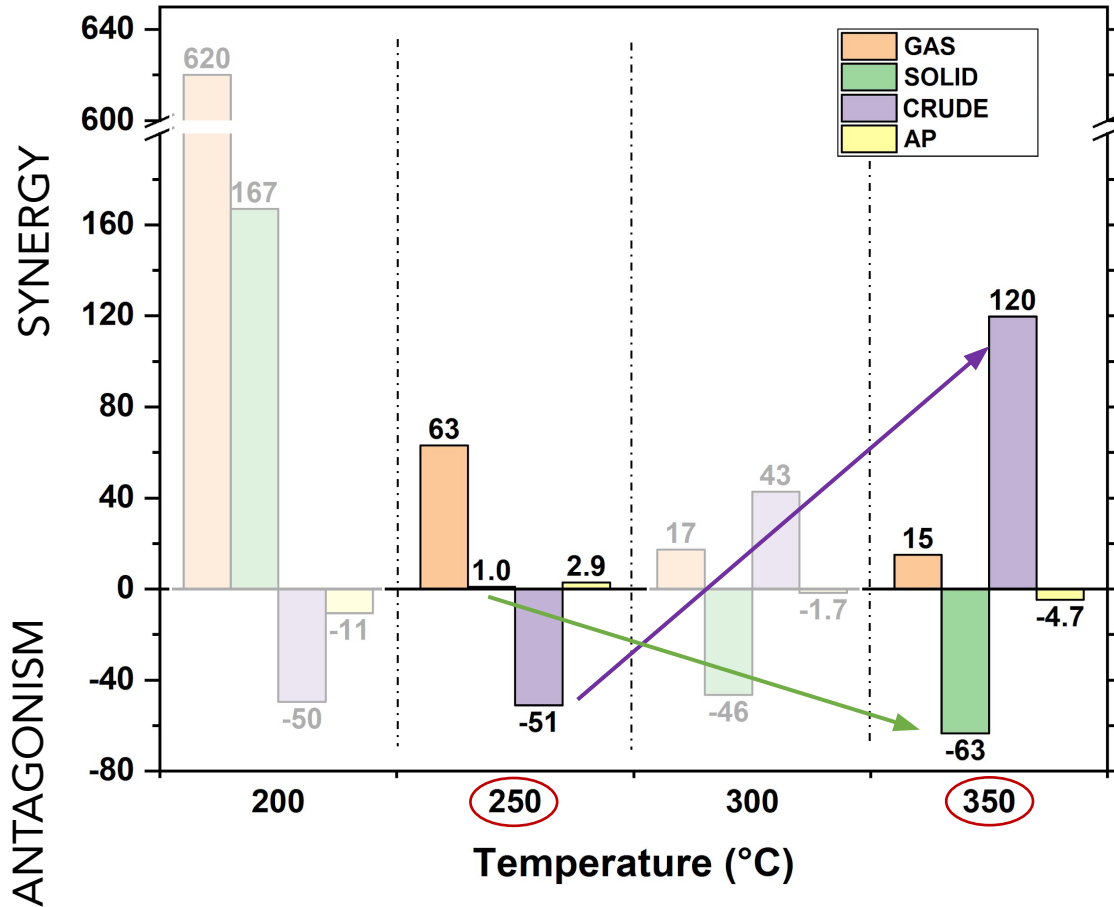
## Definitions

$$\text{Mass yield} = \frac{\text{phase mass}}{\text{reactant mass}}$$

$$\text{Mass yield AP} = 1 - \sum_{\text{other phases}} \text{Mass yield}$$

# Mass yields: synergy/antagonism

$$\text{Relative difference (\%)} = \frac{\text{yield}_{\text{experimental}} - \text{yield}_{\text{averaged}}}{\text{yield}_{\text{experimental}}}$$

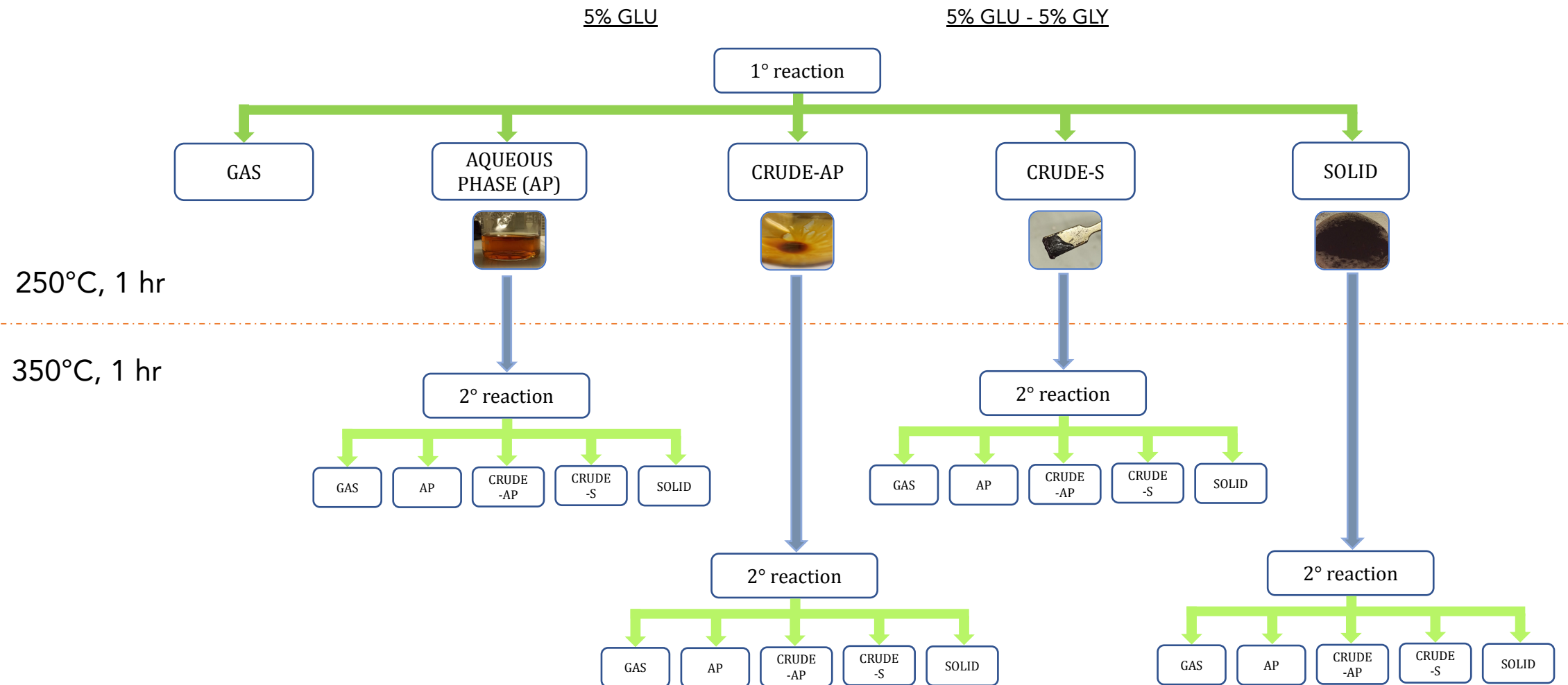


- **GAS** → much higher CO<sub>2</sub> production at 200 °C  
high reactivity already at 200 °C
- **SOLID** → strongly disfavored with temperature
- **CRUDE** → strongly favored at higher temperature



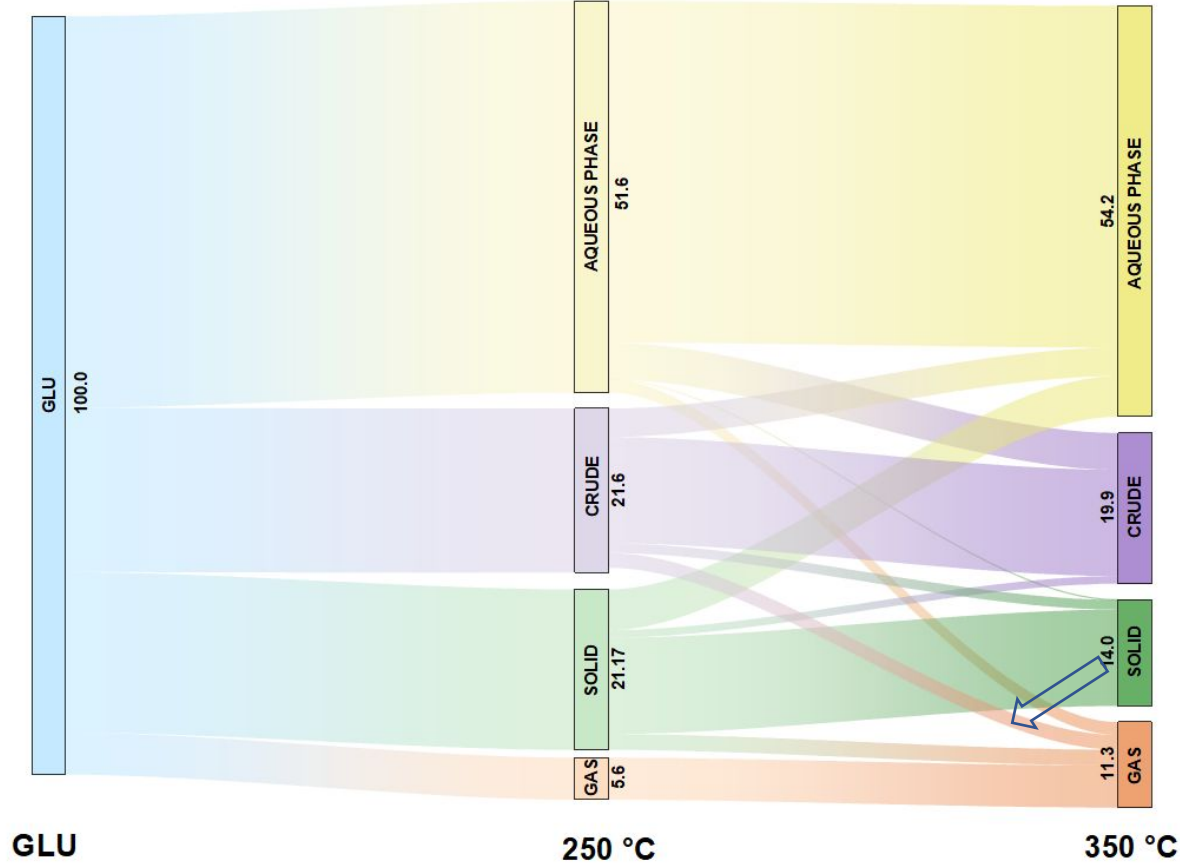
Exchange **SOLID** - **CRUDE** ?

# Work-up

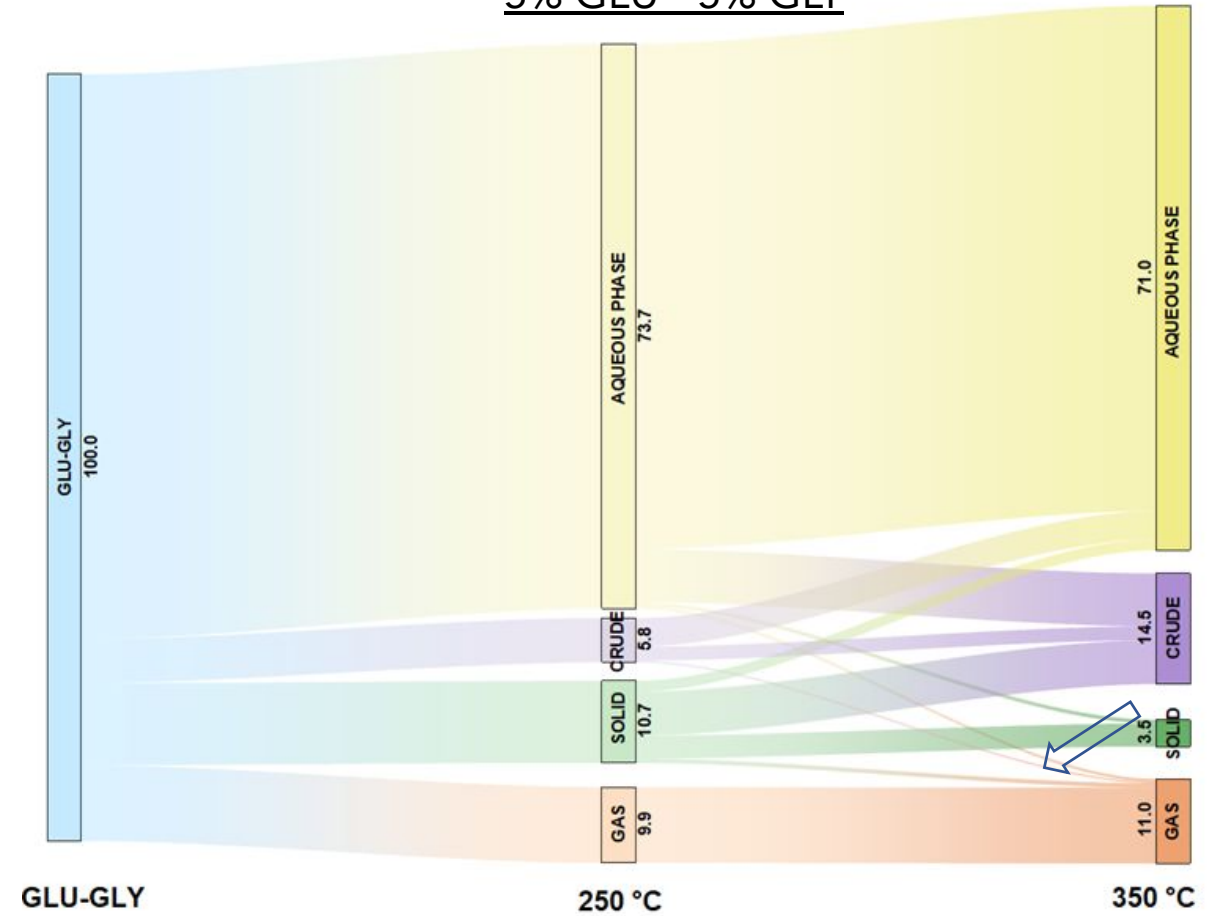


# Consecutive reactions

5% GLU



5% GLU - 5% GLY



- CRUDE degraded to AP
- SOLID doesn't produce CRUDE but AP-solubles

- CRUDE production from SOLID and AP

# Conclusions and perspectives

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## CONCLUSIONS:

- Study of glycine-glucose interactions at different temperatures
- Different mechanisms between glucose and glycine alone with respect to glucose-glycine
- Biocrude yield favored at higher temperature with mixture GLU-GLY
- Experimental observation of biocrude production from solid and aqueous phase for GLU-GLY

## TAKE-HOME MESSAGE:

- Better to process carbohydrates-rich and protein-rich waste together and at higher temperature



*Thank you for your  
kind attention*

Unraveling HTL mechanisms of  
carbohydrates-proteins interactions

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*Tito E., Zoppi G., Pipitone G., Monteverde Videla A.H.A, Pirone R., Bensaid S.*



# Appendix: macromolecules decomposition

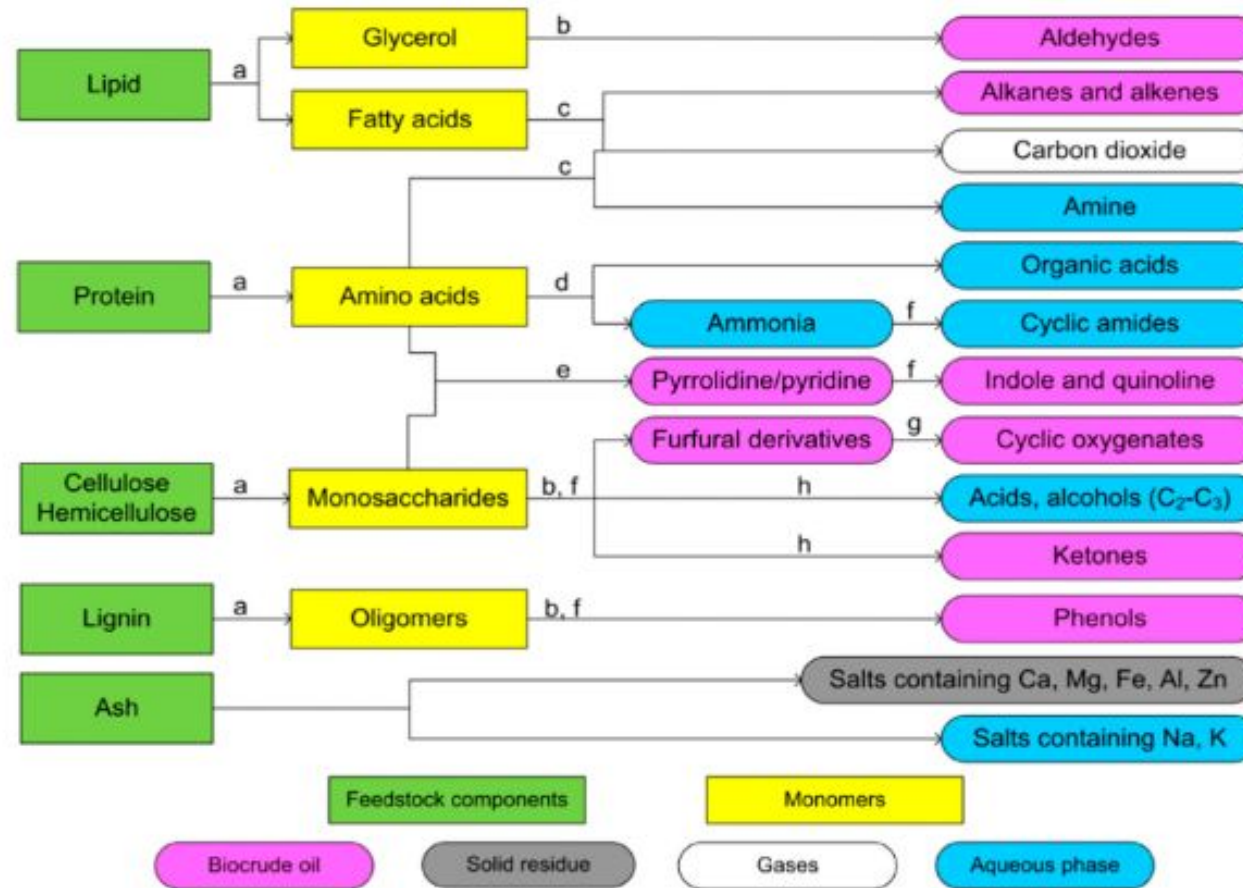
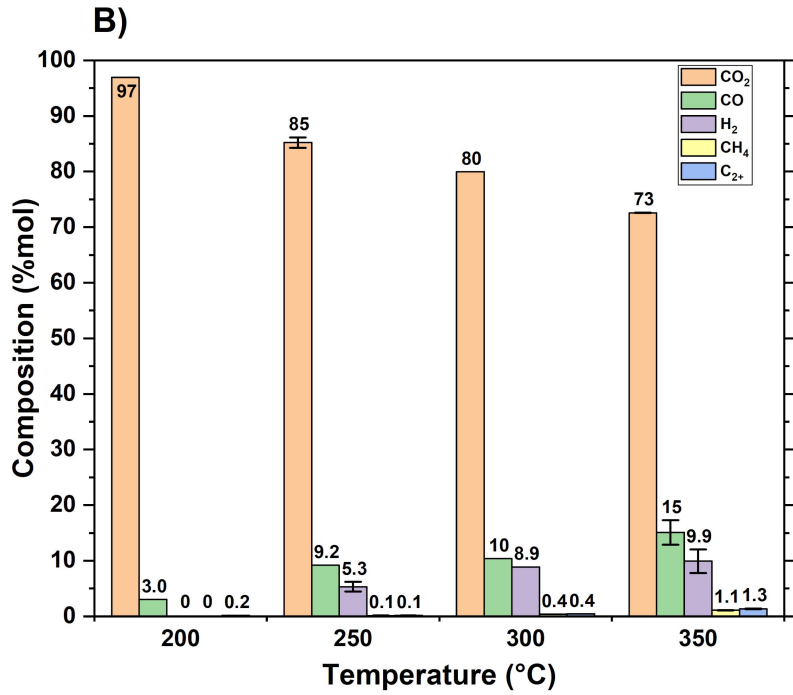


Fig. 2. Potential reaction pathways and resultant products for the conversion of biomass via hydrothermal treatment: (a) hydrolysis; (b) dehydration; (c) decarboxylation; (d) deamination; (e) Maillard reaction; (f) cyclization; (g) polymerization; and (h) decomposition (Lu et al., 2017). Copyright © 2017 Elsevier B.V.

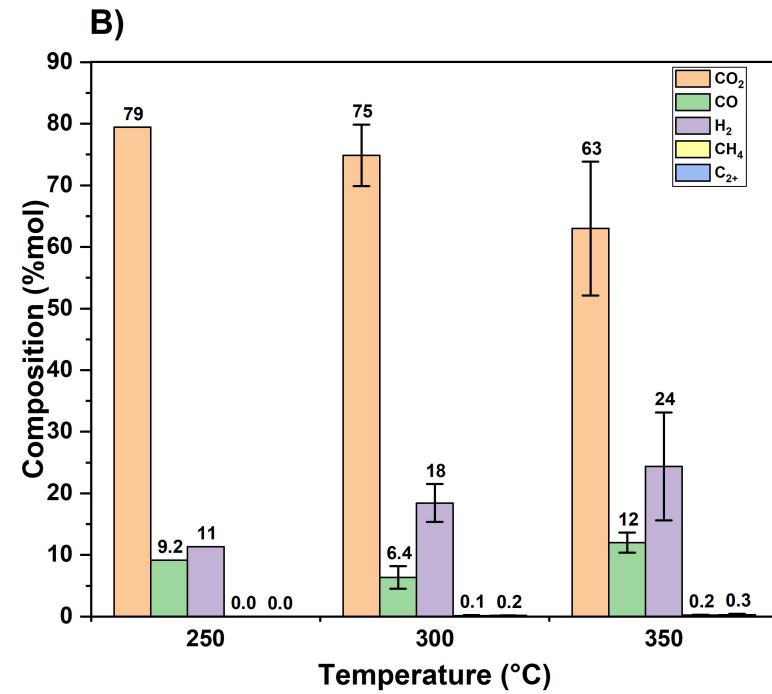
Leng, Lijian et al. 2020. "Bioenergy Recovery from Wastewater Produced by Hydrothermal Processing Biomass: Progress, Challenges, and Opportunities." *Science of the Total Environment* 748: 142383.



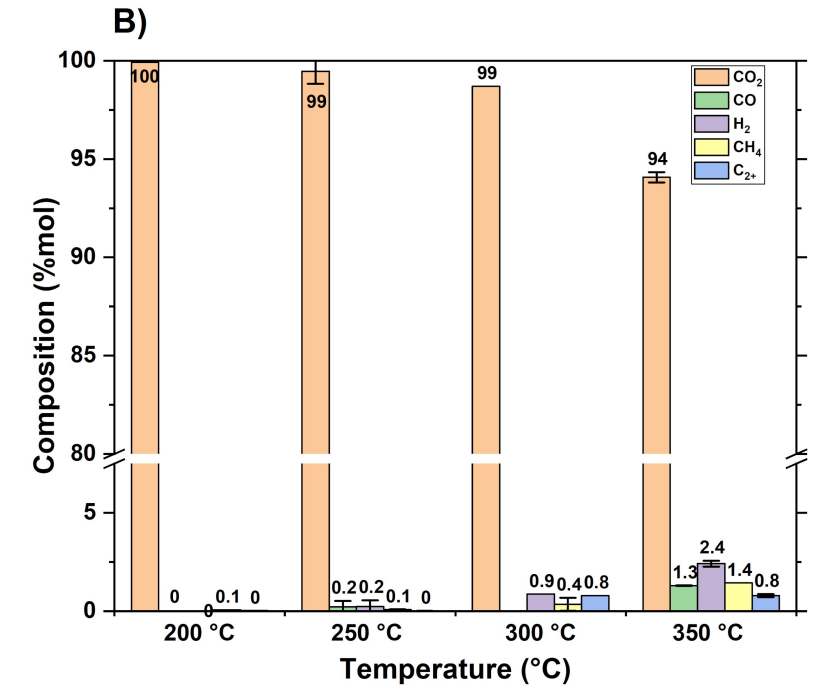
# Appendix: gas composition



GLU

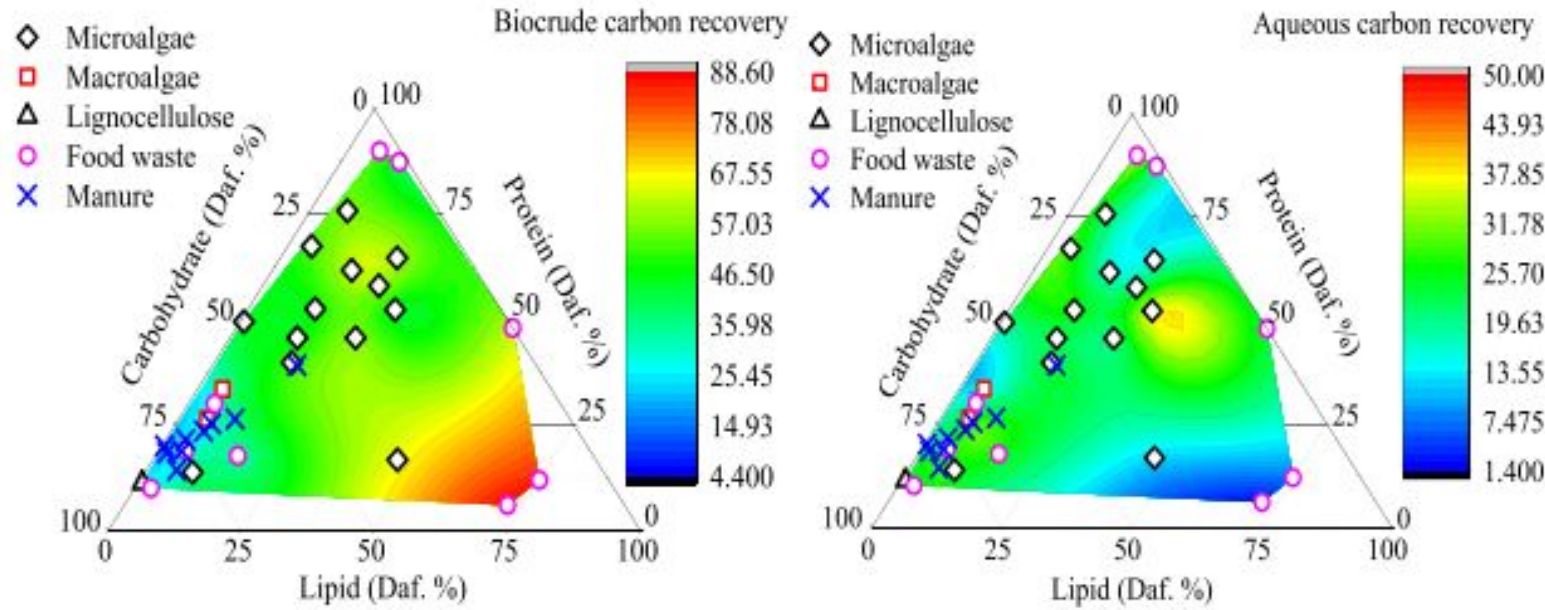


GLY



GLU-GLY

# Appendix: C and N distribution



**Fig. 4.** Carbon and nitrogen recovery of the biocrude and aqueous phase after the HTL of biomass. The lipid, protein, and carbohydrate content are based on the dry ash-free weight of the biomass. The data were obtained from previous studies involving the HTL of microalgae (Jazrawi et al., 2013; Li et al., 2018b; Watson et al., 2019; Huang and Yuan, 2016; Huang et al., 2016; Duan et al., 2018; Eboibi et al., 2014a; Christensen et al., 2014), macroalgae (Watson et al., 2019; Anastasakis and Ross, 2011; Duan et al., 2018), lignocellulose (Zhu et al., 2017), food waste (Déniel et al., 2016; Deniel et al., 2017; Aierzhati et al., 2019), and manure (Lu et al., 2017b, 2018b).

# Appendix: Maillard

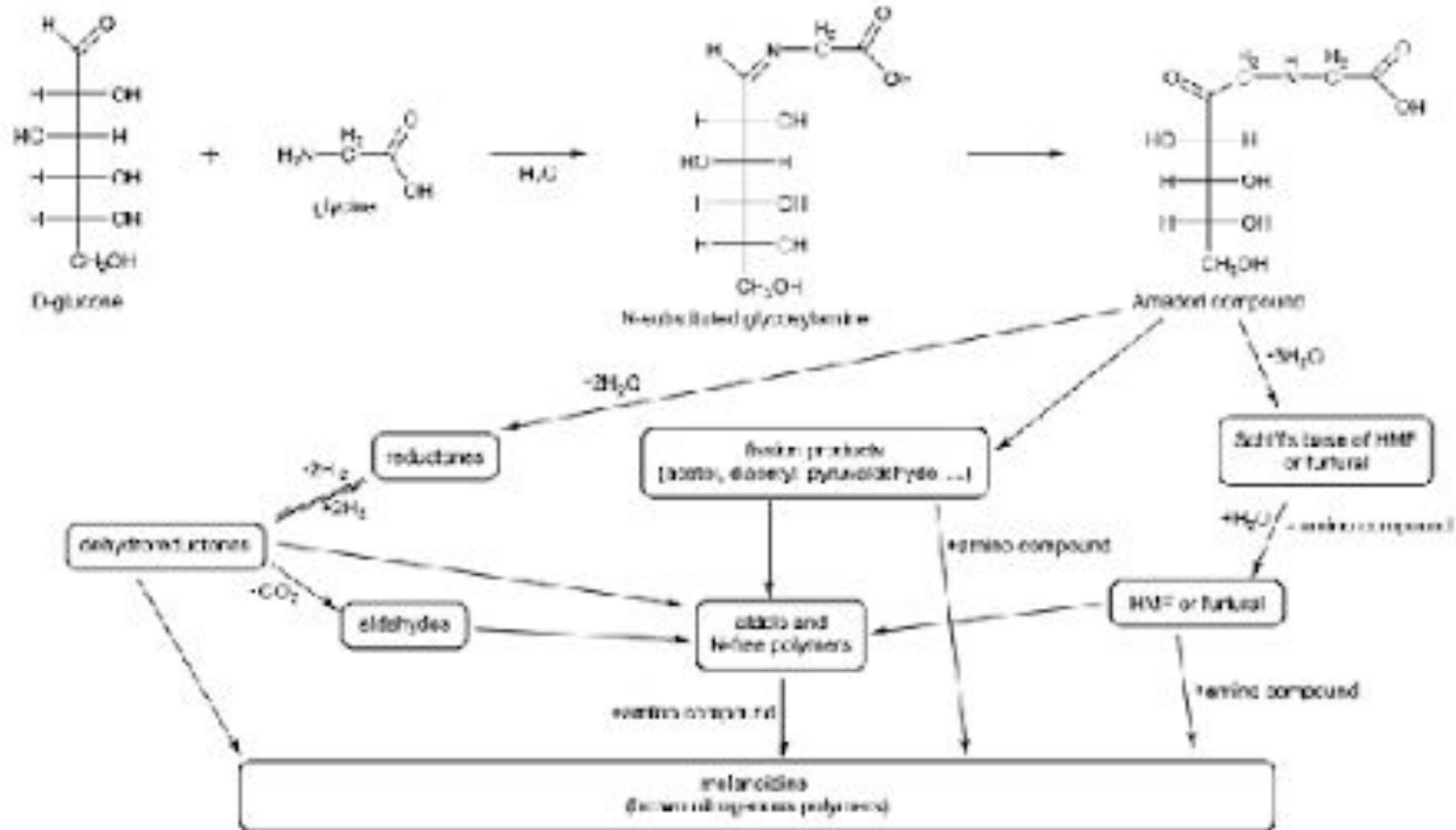
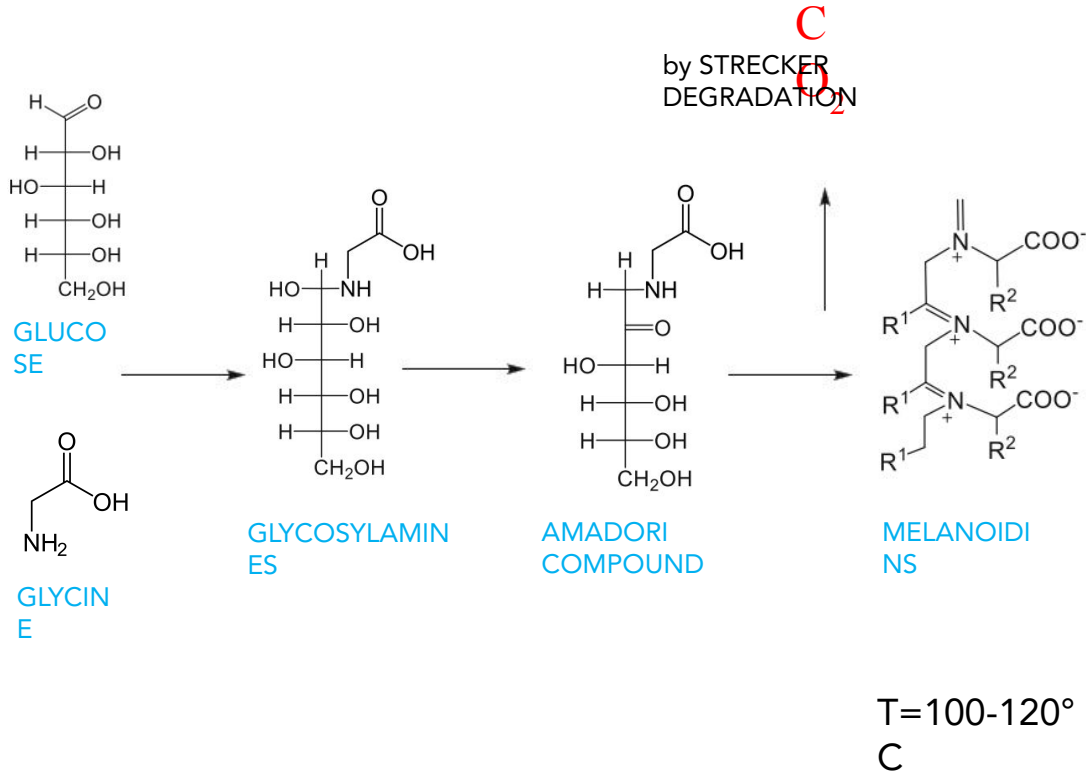
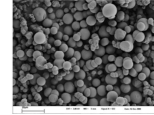


Figure 1. Overview of the Maillard reaction network. Adapted from refs. 13 and 14.

Adapted from Ind. Eng. Chem. Res., Vol. 49, No. 5, 2010

# Appendix: Maillard

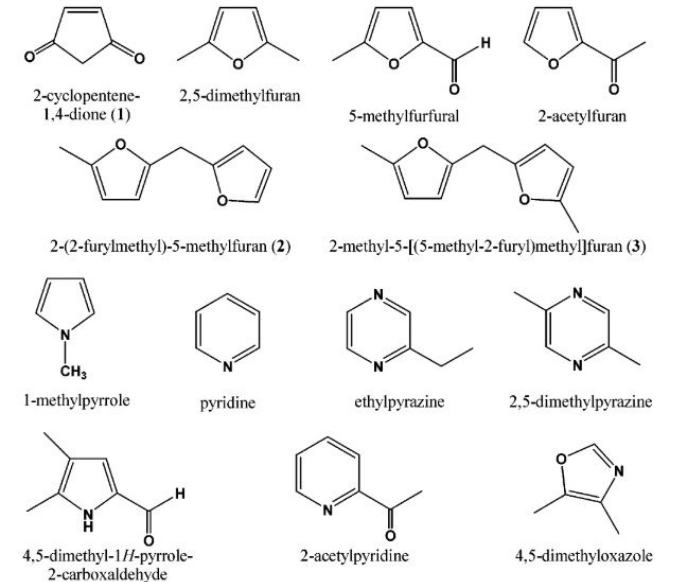
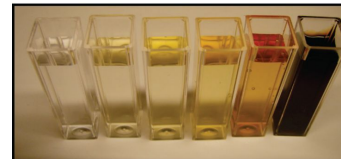


- Presence of pyridinic, pyrazinic, pyrrolic. No amines and few amides
- No trace of leftover glycine
- pyrazo-furanic matrix

CHAR/B-Non water soluble

A  
B  
Non DEE soluble

- MW > 10kDa (>40 unità GLU-GLY)
- Radical scavenger
- GLU-GLY backbone mainly intact
  - First aromatization
  - Amide, amine, pyrroles
  - Structure (furanic) due to GLU-GLU interaction



$T > 200-250^\circ$

- Decomposition product <sup>C</sup>

# Appendix: Glucose

Operating conditions

T=300 °C  
 Stirring=510 rpm  
 P start =2 barg  
 Time = **0 min**  
 V liquid=200 ml  
 V vessel =570 ml  
 5 wt% **glucose**- 5 wt% **glycine**

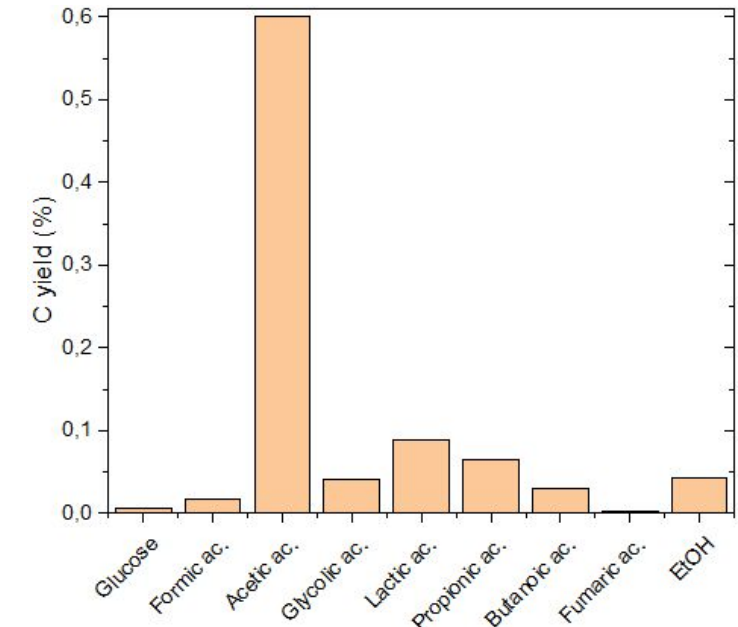
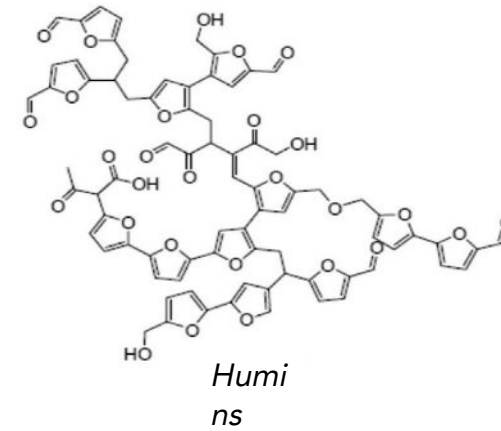
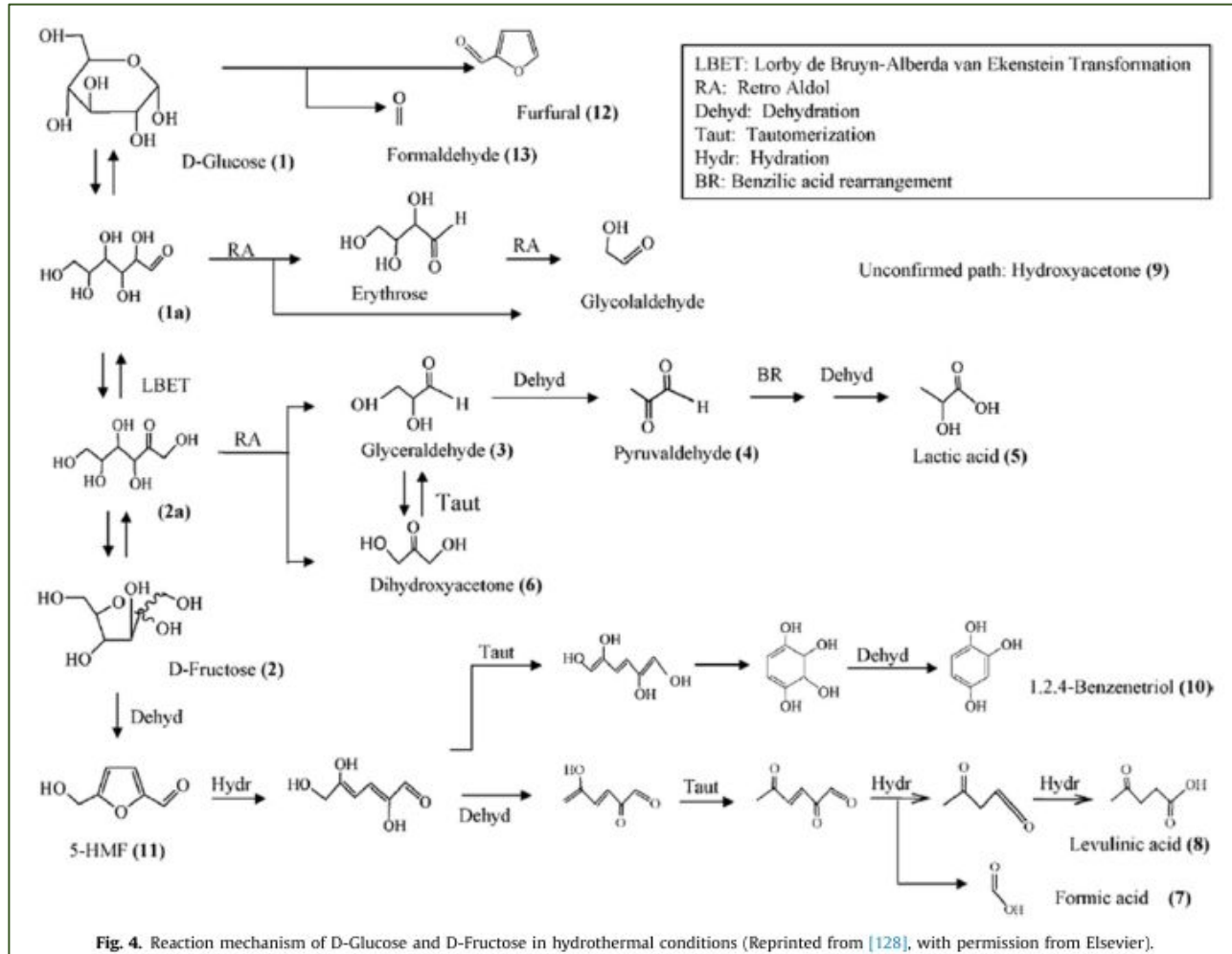
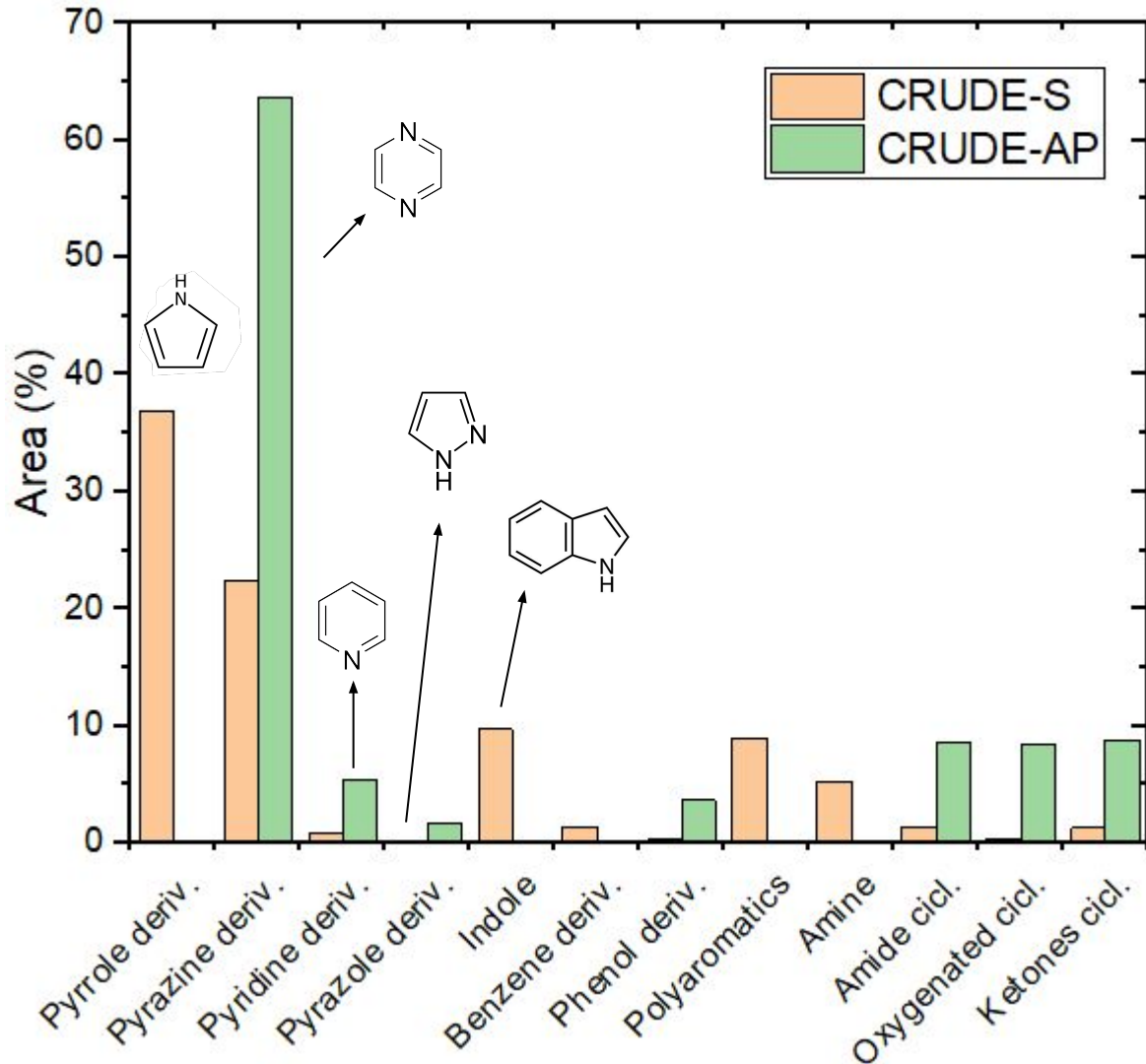


Fig. 4. Reaction mechanism of D-Glucose and D-Fructose in hydrothermal conditions (Reprinted from [128], with permission from Elsevier).

M. Déniel et al. / Renewable and Sustainable Energy Reviews 54 (2016) 1632–16521639

# Appendix: GC-MS



## Operating conditions

T=300 °C  
Stirring=510 rpm  
P start =2 barg  
Time = **0 min**  
V liquid=200 ml  
V vessel =570 ml  
5 wt% **glucose**- 5 wt% **glycine**

- Maillard reaction took place
- More pyrrole in crude CRUDE-S
- More pyrazine in crude CRUDE-AP
  - Different solubilities

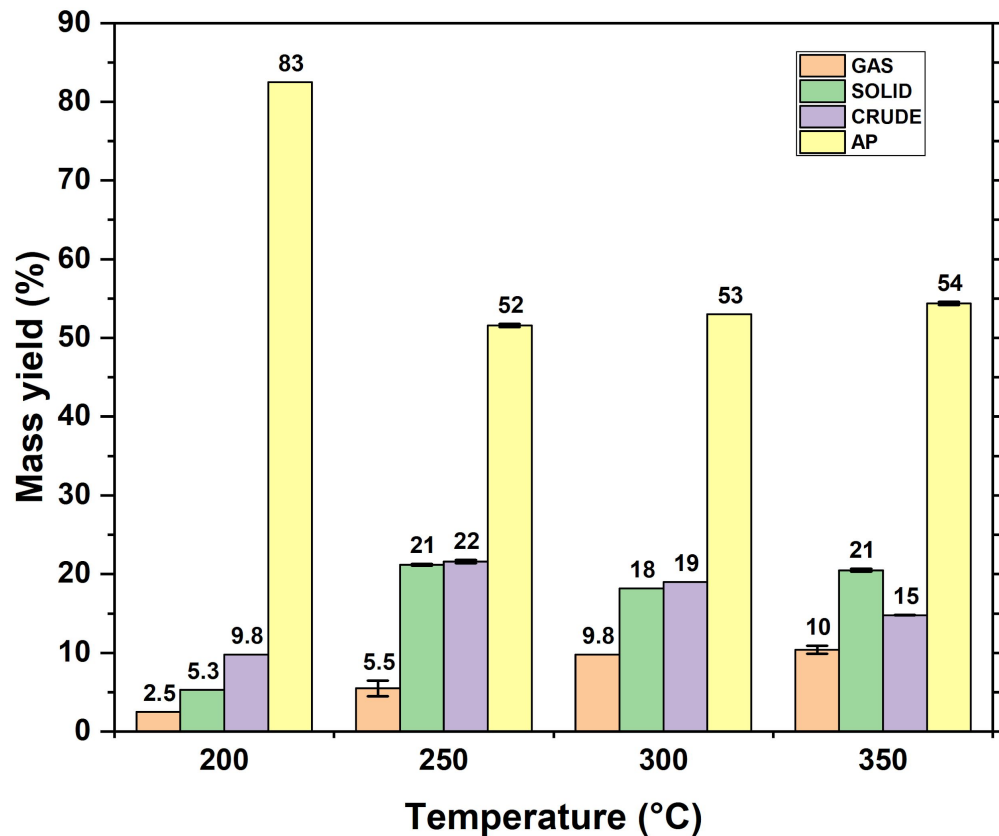
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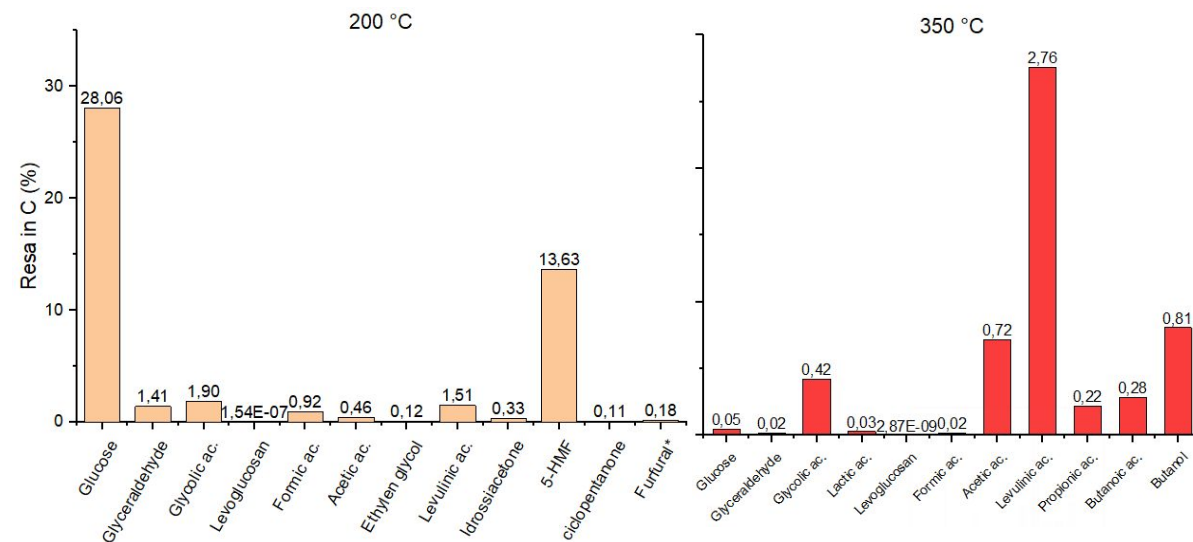
# Mass yields: glucose alone

## 5% GLU



Reaction conditions: 5 wt.% glucose, 60 min, 2 barg initial pressure, 200 g feed, 570 ml vessel volume.

- Increasing **GAS** → mainly CO<sub>2</sub>, H<sub>2</sub> and CO at high T
- Constant **SOLID** → humins formation
- Max at 250 °C for **CRUDE** → mainly benzenes, phenols and levulinic acid  
Possible instability at high T
- Residual glucose at 200 °C in **AP**



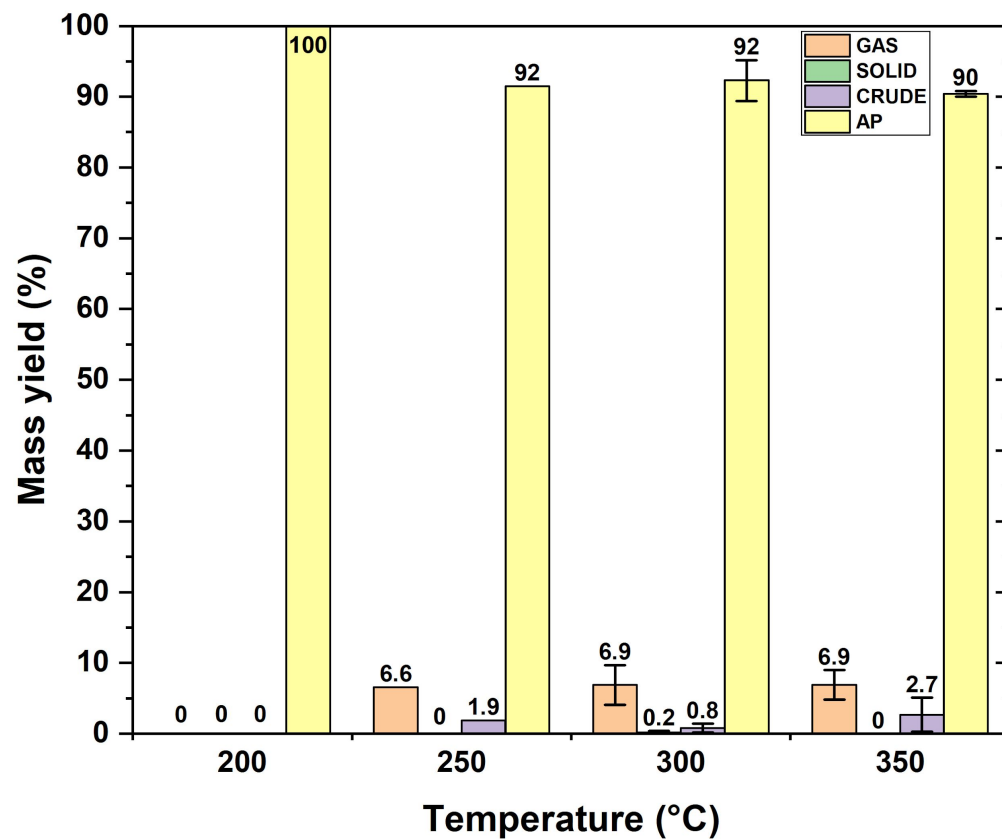
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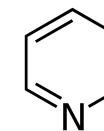
# Mass yields: glycine alone

## 5% GLY



Reaction conditions: 5 wt.% glycine, 60 min, 2 barg initial pressure, 200 g feed, 570 ml vessel volume.

- Constant **GAS** → mainly CO<sub>2</sub>, higher H<sub>2</sub> and CO than GLU
- No **SOLID**
- Very low **CRUDE** → mostly pyridines
- High residual of **AP-soluble compounds**





# Differences

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	<b>GLU</b>	<b>GLY</b>	<b>GLU-GLY</b>
Gas composition	Mainly CO <sub>2</sub> , Less H <sub>2</sub> , CO less	Mainly CO <sub>2</sub> Less H <sub>2</sub> , CO	Only CO <sub>2</sub>
Crude composition	Aromatics	Pyridines	N-containing aromatics
T max crude (°C)	250	No	350
T max solid (°C)	Constant	No	250

**STRONG DIFFERENCES**