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Cheese whey wastewater treatment by combined Electrocoagulation-Electrooxidation

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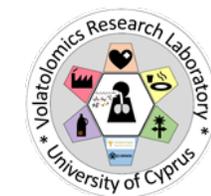
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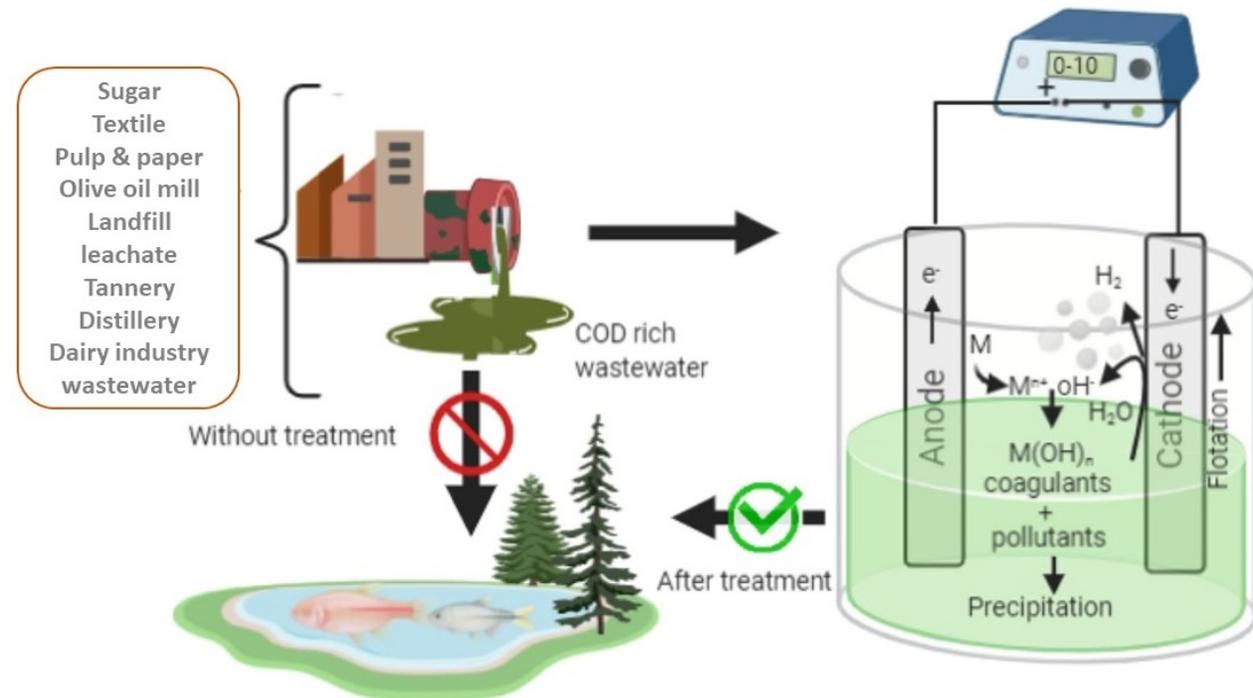
University of Cyprus
Department of Chemistry



Goat cheese whey wastewater

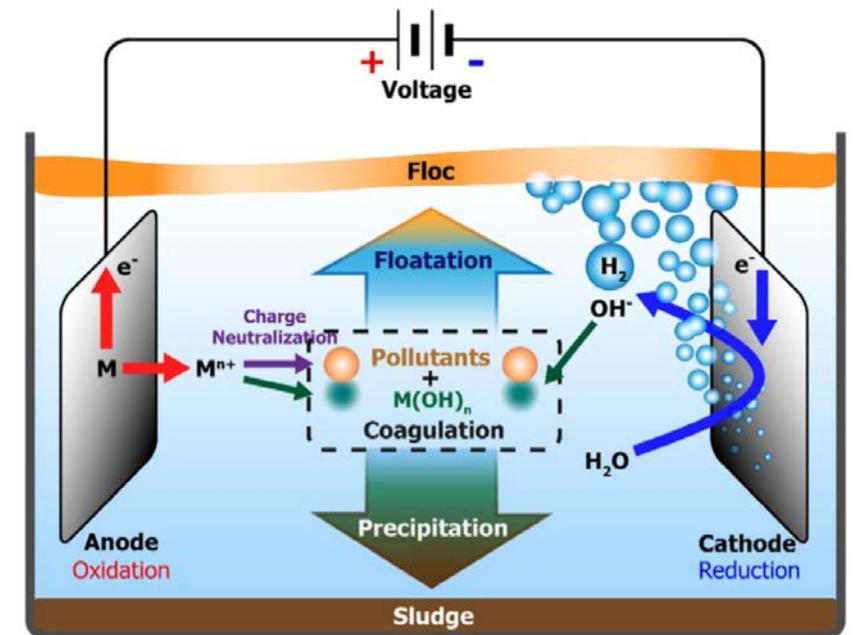


- Whey as a waste is characterized by high COD.
- It is a polluting by-product, which needs to be treated to protect the environment.
- In Europe, their direct discharges to surface waters are not permitted under the relevant European Directives 91/2717/EEC and 97/771/EEC, respectively.
- Electrochemical processes such as electrocoagulation (EC) and electrooxidation (EO) have been reported to be the most effective for COD removal.



Electrocoagulation (EC)

- Mainly Fe and Al electrodes are used, which are immersed in wastewater.
- Its effectiveness is due to the formation of insoluble metal hydroxides $M(OH)_3$ through redox reactions, which act as coagulants and are capable of destabilizing organic matter.



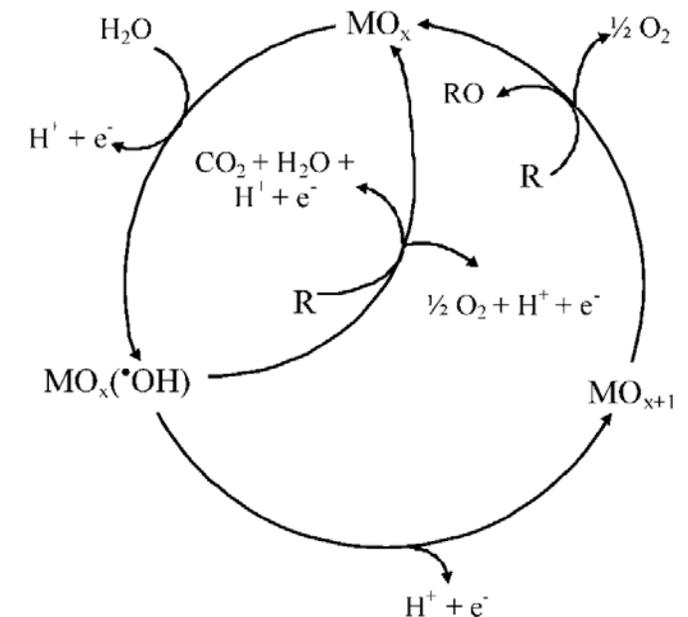
Magnisali, E., Yan, Q., & Vayenas, D. V. (2022) <https://doi.org/10.1002/JCTB.6880>

Electrochemical wastewater treatment

- The organic charge is destroyed due to the hydroxyl radicals ($\cdot OH$) formed on the anode surface by the oxidation of water.

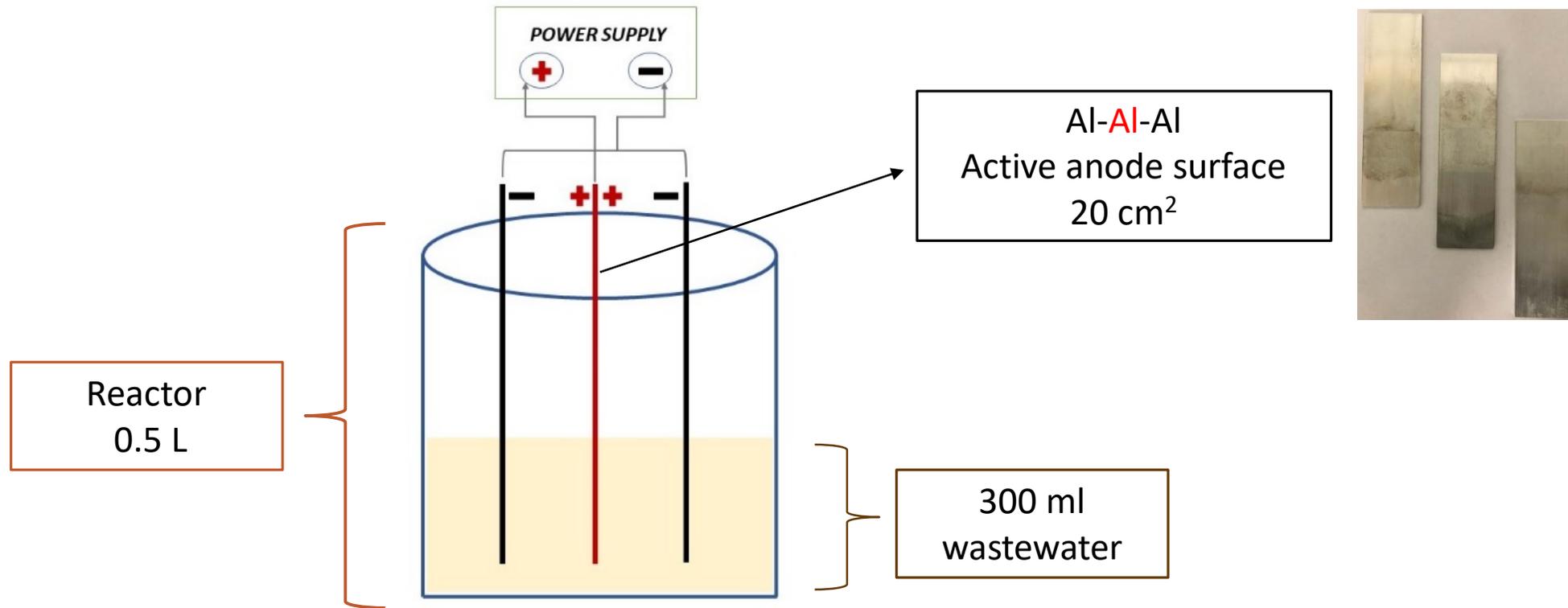


- The use of Boron-doped diamond (BDD) electrode, is able to completely and non-selectively mineralize organic pollutants (R) with high efficiency.

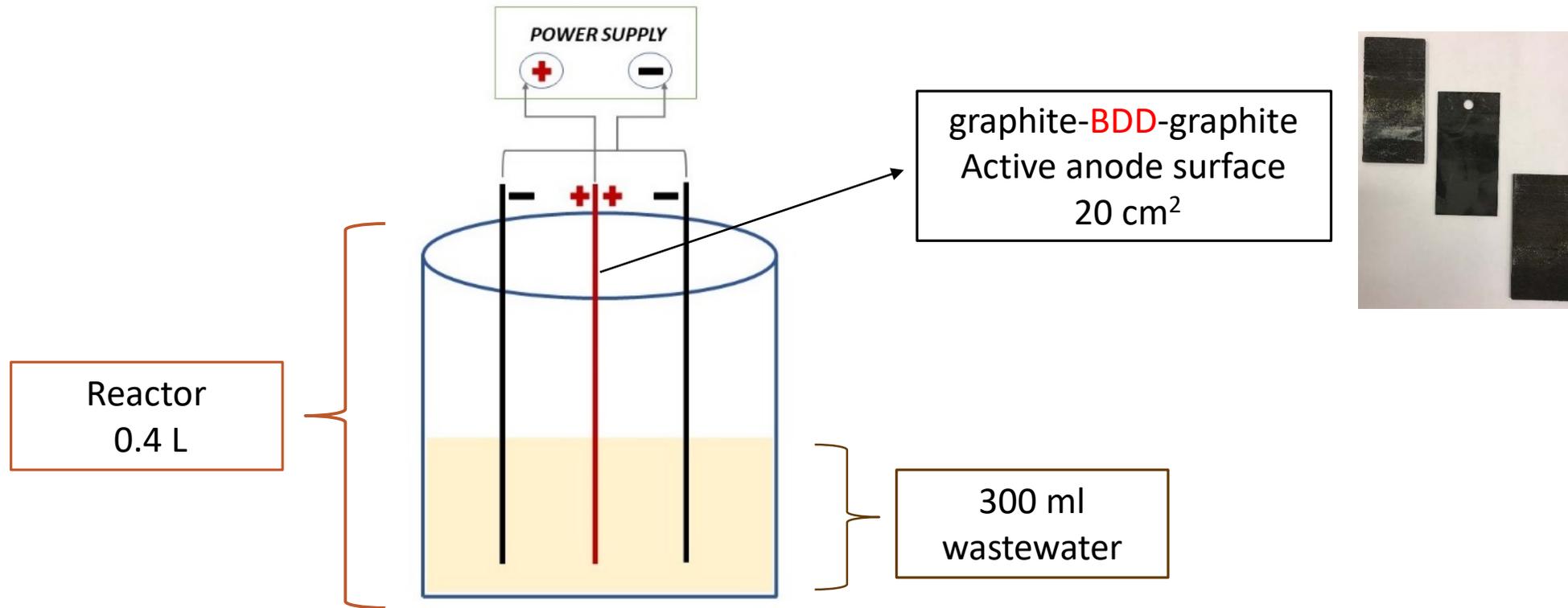


Panizza, M., & Cerisola, G. (2009) <https://doi.org/10.1021/CR9001319>

Experimental layout of the EC process



Experimental layout of the EO process



Optimization of EC and EO operating parameters

The main parameters that affect the efficiency of EC and EO treatment were studied:

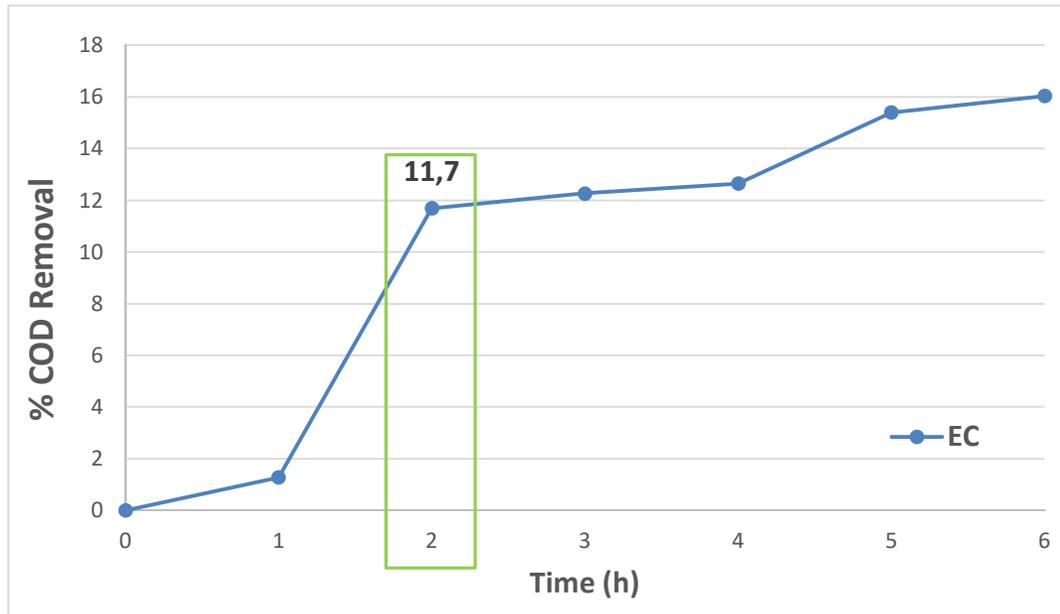
- 1) Operating time (h)
- 2) Agitation rate (rpm)
- 3) pH
- 4) Current density (mA/cm²)

The purpose was to find the optimal operating conditions for the highest efficient removal of COD from cheese whey wastewater

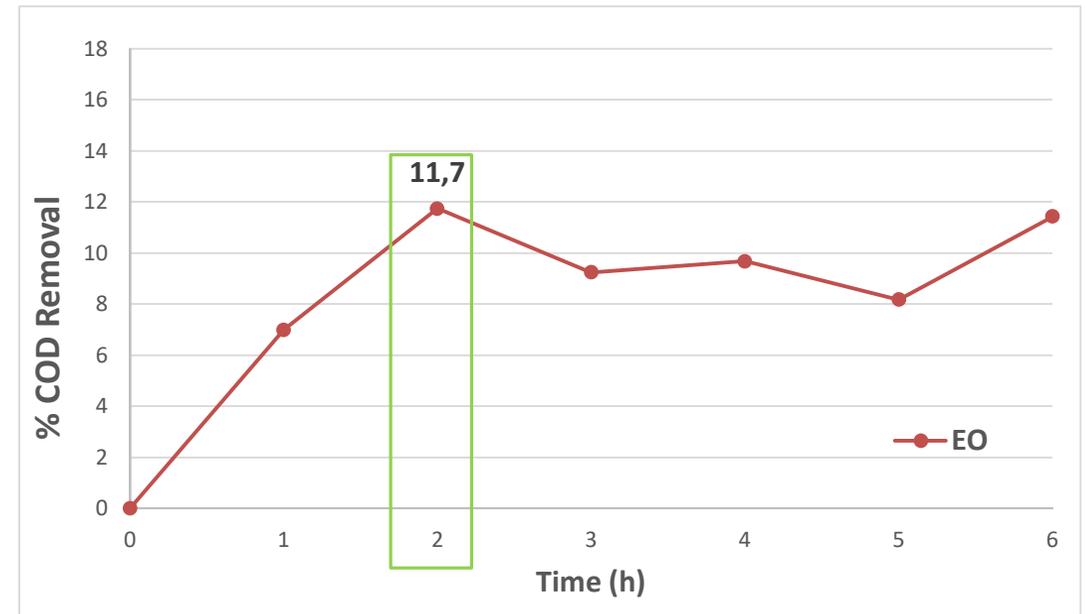
$$\% \text{ Removal} = \frac{C_{\text{initial}} - C_{\text{final}}}{C_{\text{initial}}} \times 100$$

➤ Operating time (h)

Electrocoagulation (EC)

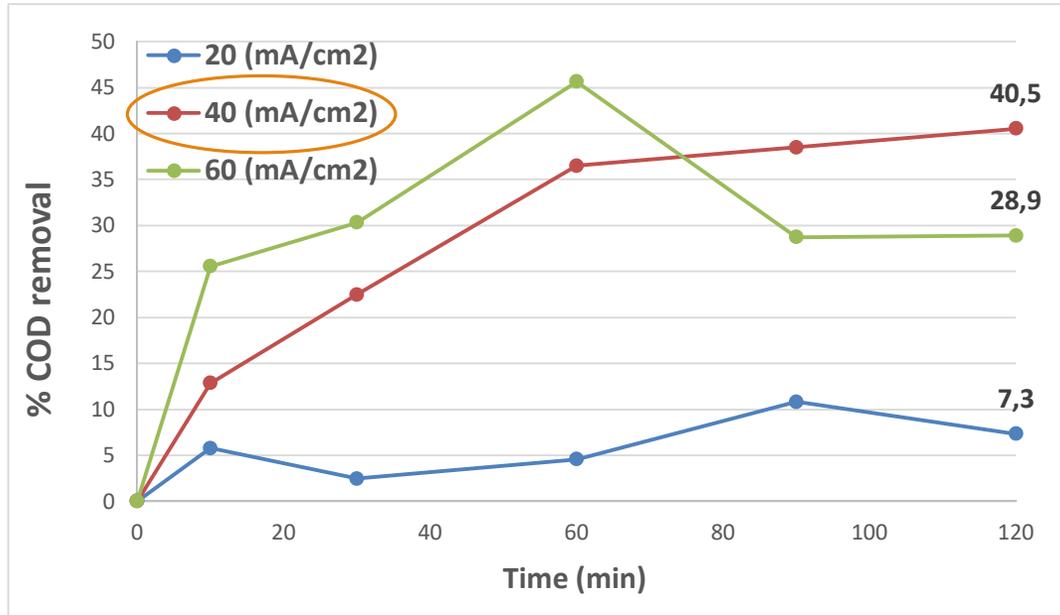


Electrooxidation (EO)



➤ Current density (mA/cm²)

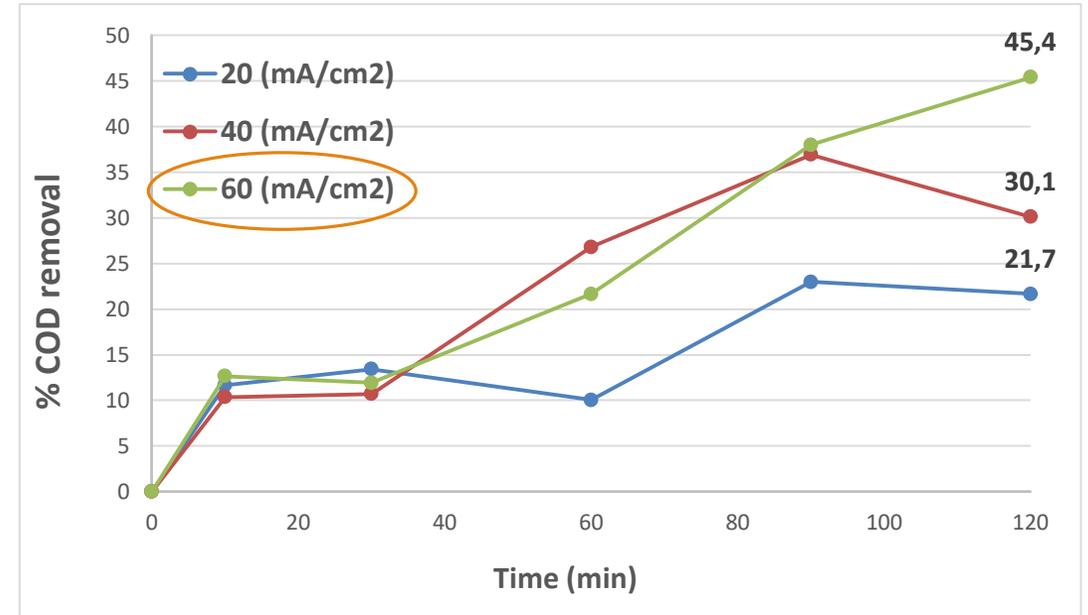
Electrocoagulation (EC)



The current determines the release rate of
 $\text{Al} \rightarrow \text{Al}^{3+} + 3\text{e}^-$

$\uparrow \text{mA/cm}^2 \rightarrow \uparrow \text{e}^- \rightarrow \text{Al}(\text{OH})_3$

Electrooxidation (EO)

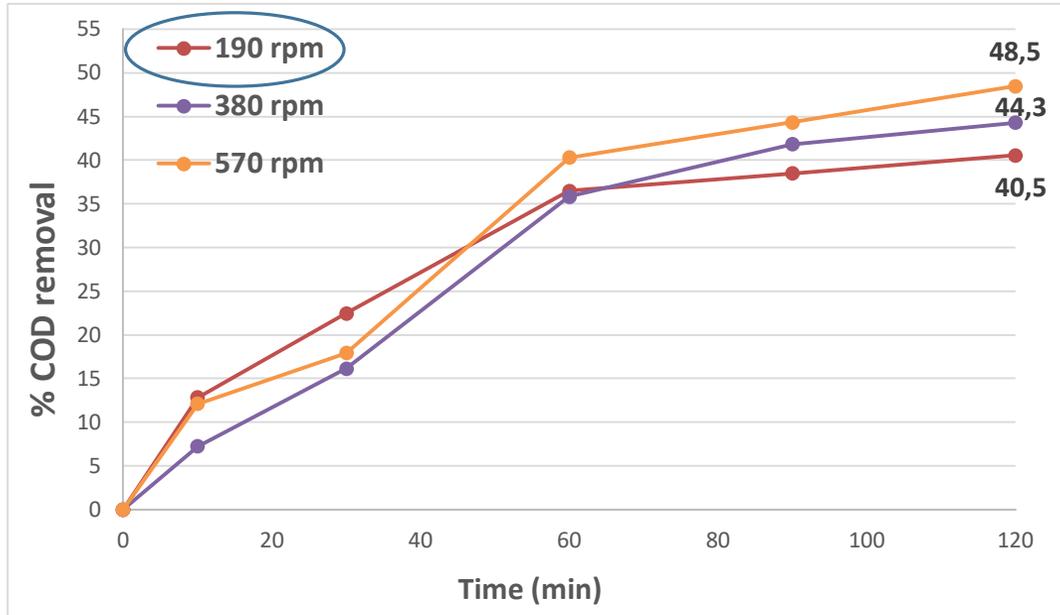


The current determines the formation of
 $\text{BDD} + \text{H}_2\text{O} \rightarrow \text{BDD}(\cdot\text{OH}) + \text{H}^+ + \text{e}^-$

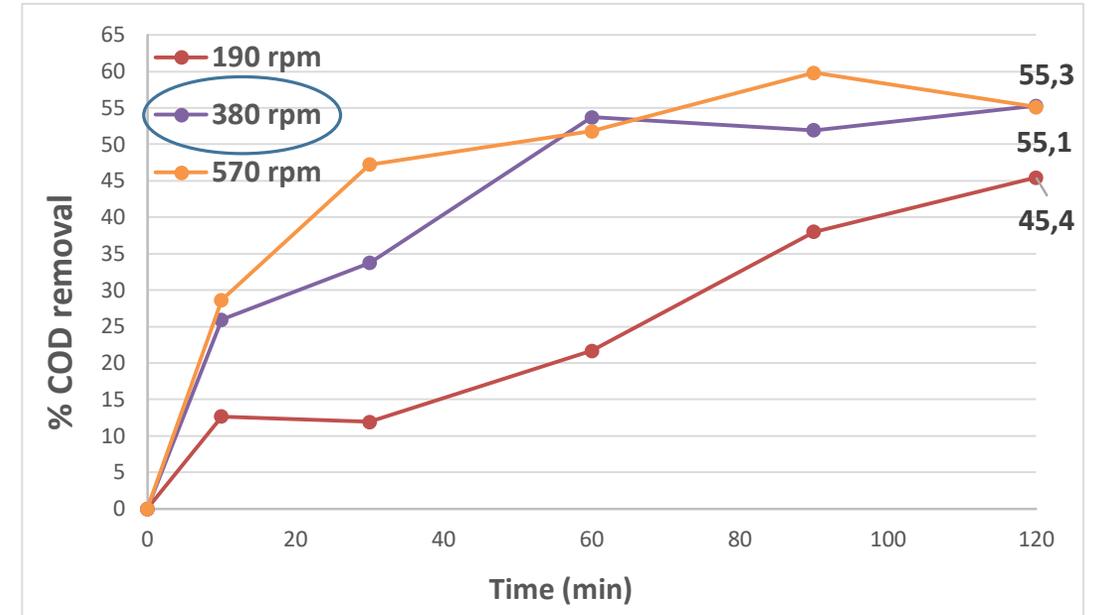
$\uparrow \text{mA/cm}^2 \rightarrow \uparrow (\cdot\text{OH})$

➤ Agitation rate (rpm)

Electrocoagulation (EC)

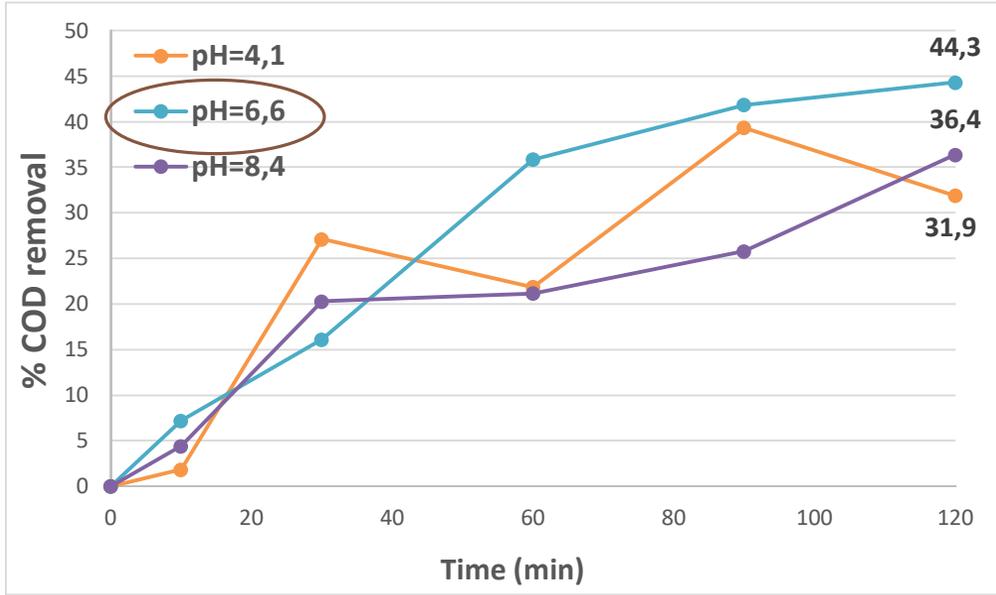


Electrooxidation (EO)

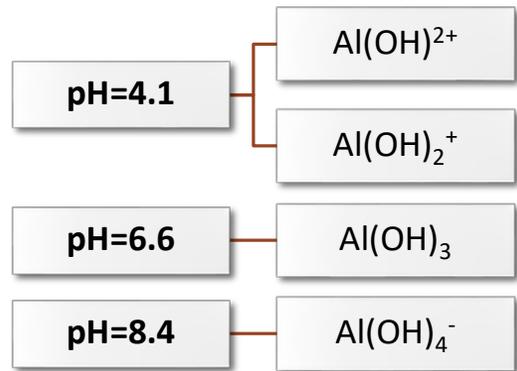


Parameter (n=3)	% Reducing weight of Al-Al-Al		
Agitation rate (rpm)	190 rpm	380 rpm	570 rpm
	0.5 %	0.7 %	1.1 %

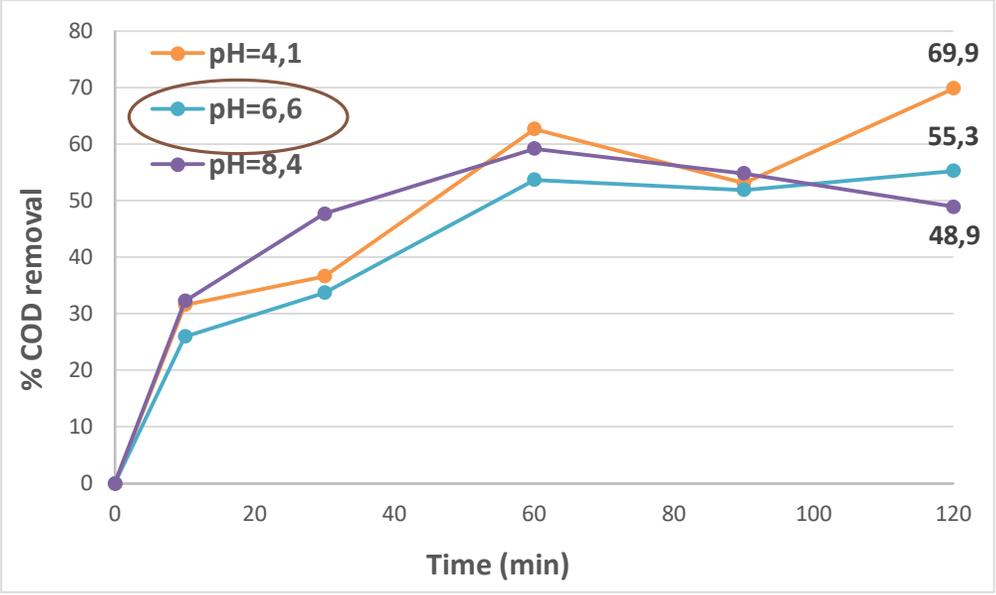
Electrocoagulation (EC)



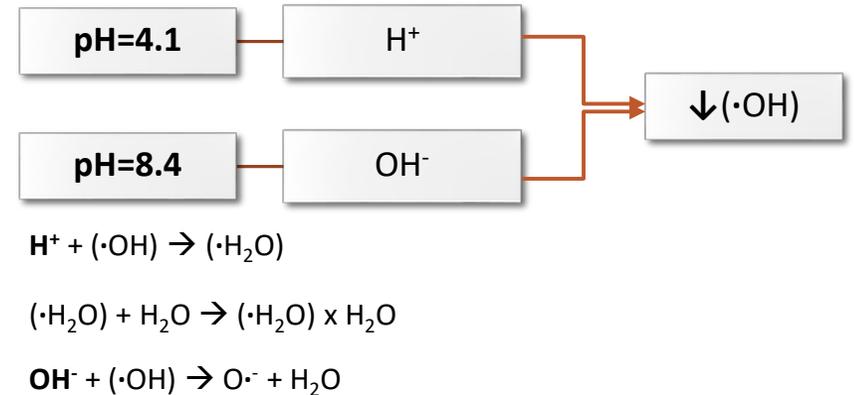
According to Yilmaz Nayir and Kara et al. (2017)



Electrooxidation (EO)



According to Stergiopoulos et al. (2021)



Combined of EC/EO processes

Optimal operating conditions:

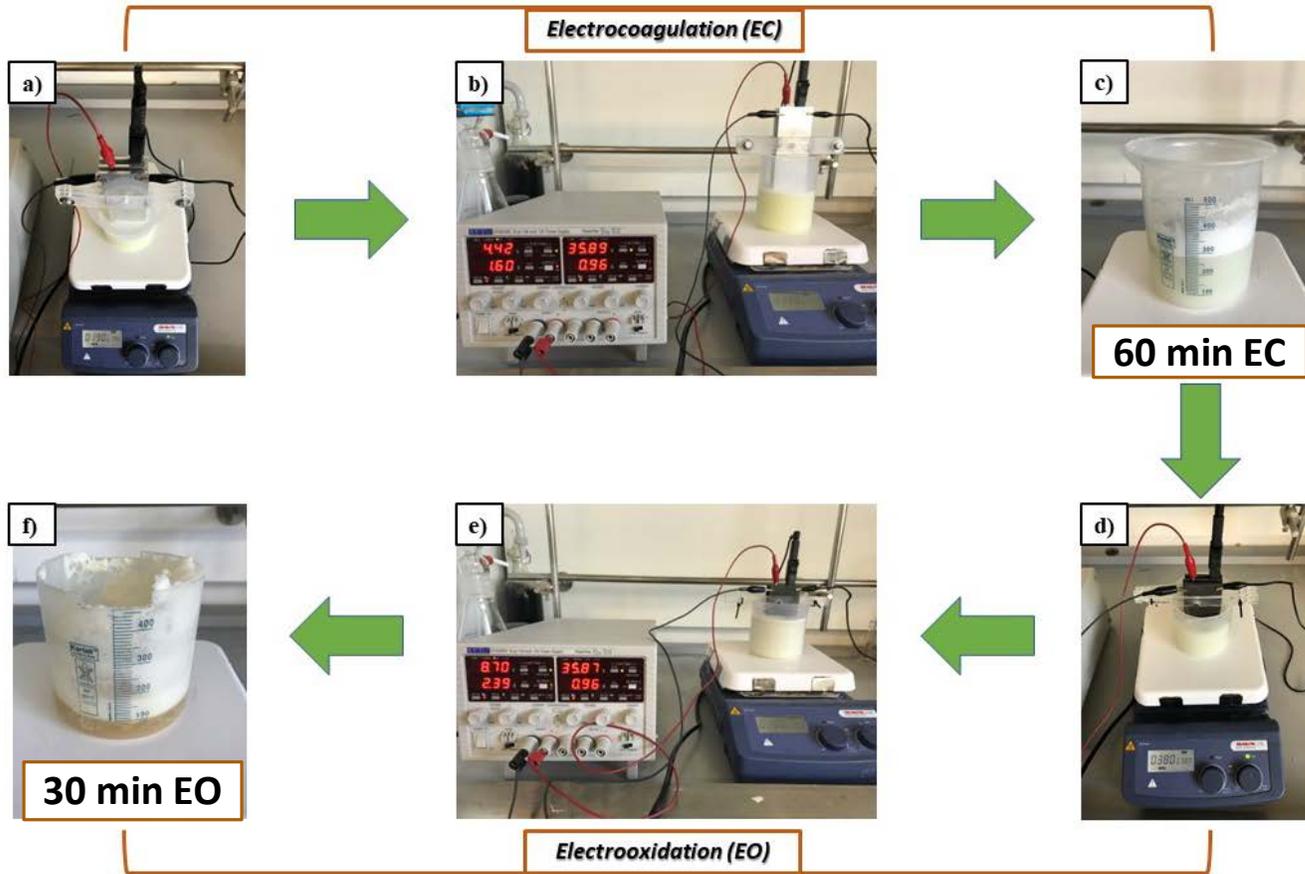
EC

- 40 mA/cm²
- 190 rpm
- without pH adjustment

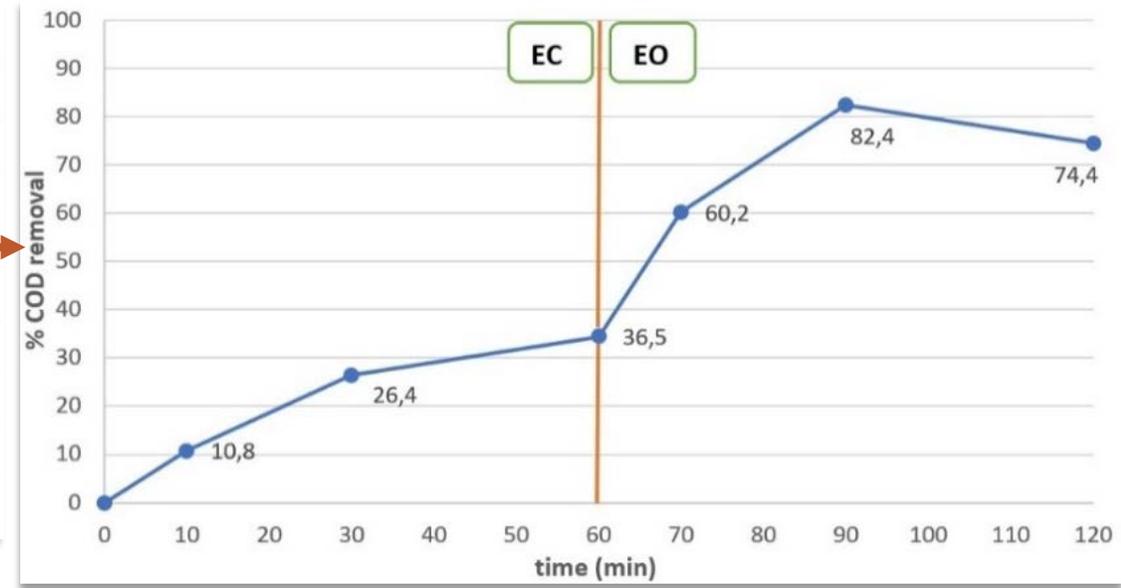
EO

- 60 mA/cm²
- 380 rpm
- without pH adjustment

EC principle



	EC followed by EO				
Efficiency %	Removal	Removal	Increase	Removal	Increase
Time (min)	60 min EC	90 min EC/EO		120 min EC/EO	
COD	36.5	82.4		74.4	
TN	25.2	38.3		45.8	
Cl ⁻	17.5		32.5		19.2
NO ₃ ⁻	31.0		20.8		31.4
PO ₄ ⁻³	11.7	12.7		14.5	
TPCs	56.5		74.2		86.4
Opti. Dens.	9.3	16.9		33.4	



Type of wastewater	Electrochemical system	Electrode anode/cathode	Current density (mA/cm ²)	Time (min)	pH	COD removal (%) *	References
Cheese whey wastewater	EC	Fe/Fe	60	20	5	86.4	Tezcan Un et al. (2014)
Artificial dairy wastewater	EC	Al/Al	-	30	7.1	61.0	Tchamango et al. (2010)
Dairy wastewater	EC	6xAl	31.5	60	7.2	98.8	Bazrafshan et al. (2013)
Dairy wastewater	EC	8xFe	5	15	7	58	Valente et al. (2012)
Cheese whey wastewater	EC/EO	Al/Fe	25	120	5	82	Yilmaz Nayır & Kara et al. (2017)
Dairy wastewater	EC/EO	BDD/Fe	25	300		89	Chakchouk et al. (2017)
		Ti/Ti	140	75		70	

* The % removal efficiency of COD is calculated based on its initial concentration, so the comparative results also depend on the initial content of wastewater in COD.

HiSorb TD-GC/MS sample analysis

The developed HiSorb TD-GC/MS method was used to analyze the VOCs of cheese whey wastewater before and after the optimal electrochemical processes (EC, EO and EC/EO).

$$\text{Semi - quantified} = \frac{A_c}{A_{IS}} \times C_{IS} \text{ (mg L}^{-1}\text{)}$$

$$\text{Compound category (\%)} = \frac{\sum_{category} \frac{A_c}{A_{IS}}}{\sum_{total VOCs} \frac{A_c}{A_{IS}}} \times 100$$



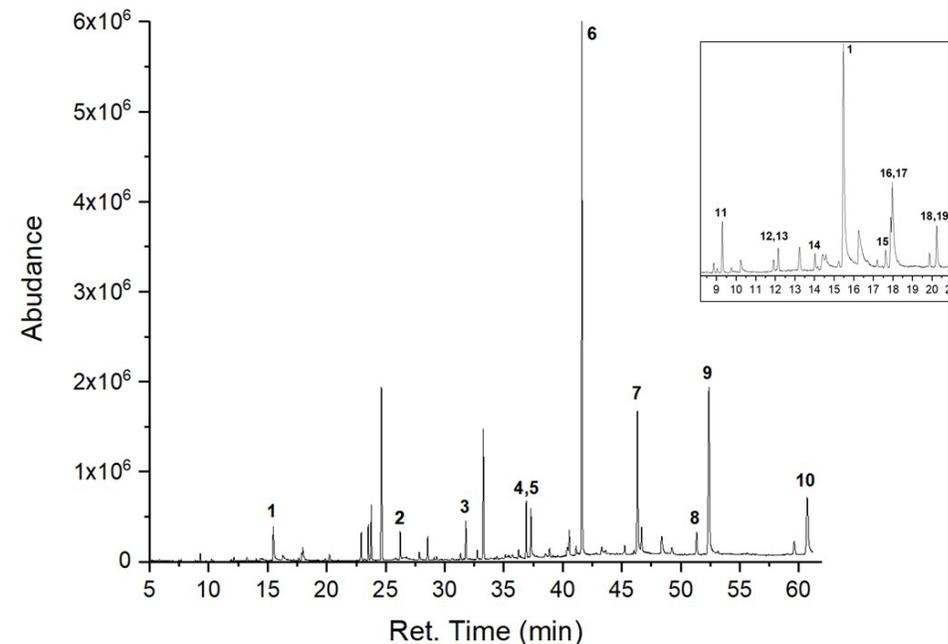
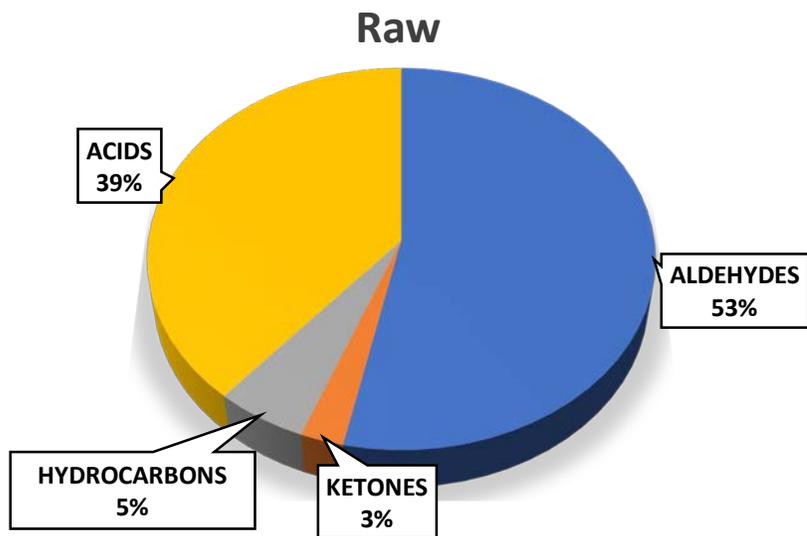
HS-HiSorb



TD-GC/MS

☐ ***VOCs of raw cheese whey wastewater (n=9)***

Categories of VOCs	Number of emitted VOCs	Structure	Σ mg L ⁻¹ (\pm CL)
Aldehydes	14	<chem>R-CHO</chem>	1.379 (\pm 0.955)
Acids	9	<chem>R-COOH</chem>	1.003 (\pm 0.522)
Ketones	7	<chem>R-C(=O)-R</chem>	0.067 (\pm 0.036)
Hydrocarbons	9	<chem>R-R</chem> <chem>R=R</chem> <chem>R#R</chem>	0.137 (\pm 0.102)
Others	2	<chem>R-OH</chem> <chem>R-S-R</chem>	0.002 (\pm 0.001)
Total VOCs	41		2.588 (\pm1.616)



No	VOCs	No	VOCs
1	1,3-Cyclopentadiene, 1-methyl	11	Acetone
2	Hexanal	12	1-Hexene
3	Heptanal	13	n-Hexane
4	Octanal	14	Butanal
5	Hexanoic acid	15	1-Heptene
6	Nonanal	16	Heptane
7	Octanoic acid	17	Acetic acid
8	Nonanoic acid	18	2-Pentanone
9	2-Undecenal	19	Pentanal
10	n-Decanoic acid		

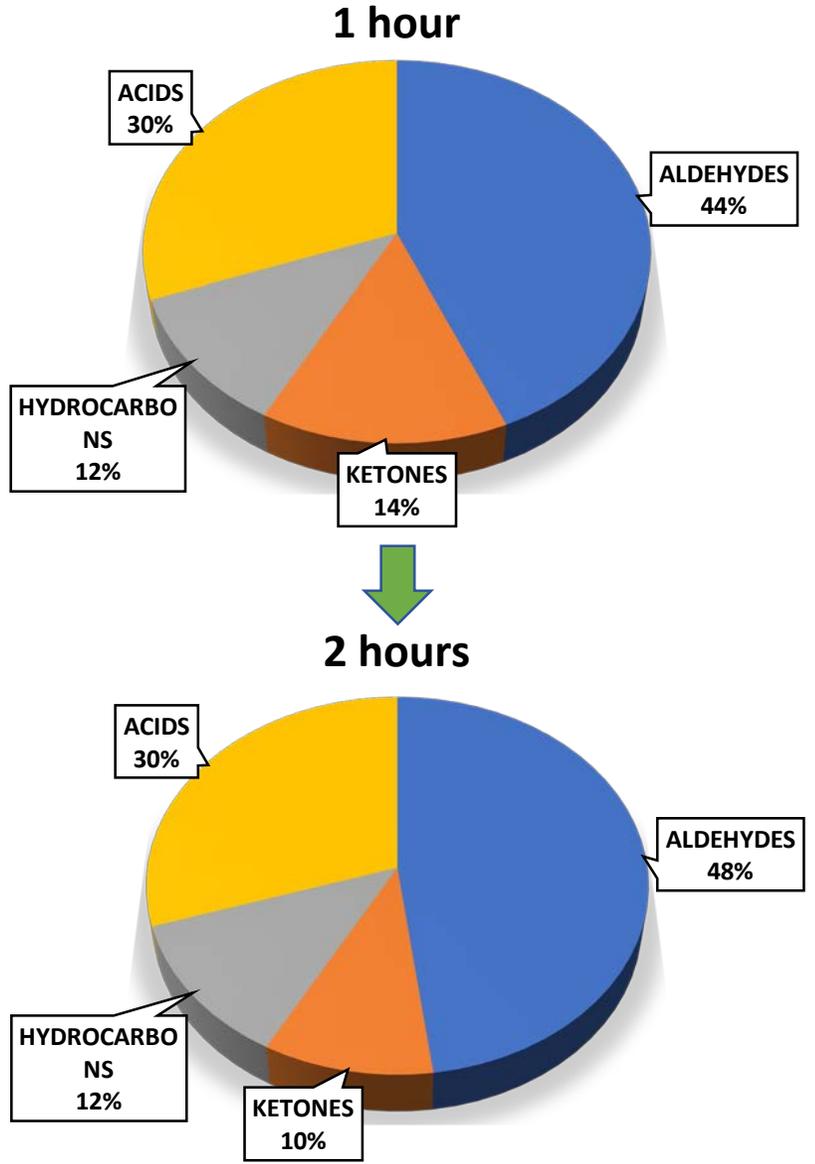
□ VOCs of EC-treated (2 h) cheese whey wastewater (n=3)

Categories of VOCs	Optimized EC process	
	Σ mg L ⁻¹ (±CL)	^b Efficiency (↓ η ↑)
Aldehydes	0.293 (±0.144)	↓ 78.8 %
Acids	0.183 (±0.115)	↓ 81.8 %
Ketones	0.062 (±0.036)	↓ 7.5 %
C-H	0.074 (±0.057)	↓ 46.0 %
Others	^a S/N	
Total VOCs	0.612 (±0.352)	↓ 76.4 %

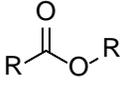
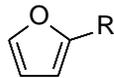
^a Low signal to noise ratio (S/N≤3)

$$^b \% \text{ Efficiency } (\downarrow) = \frac{C_{RAW} - C_{EC-treated}}{C_{RAW}} \times 100$$

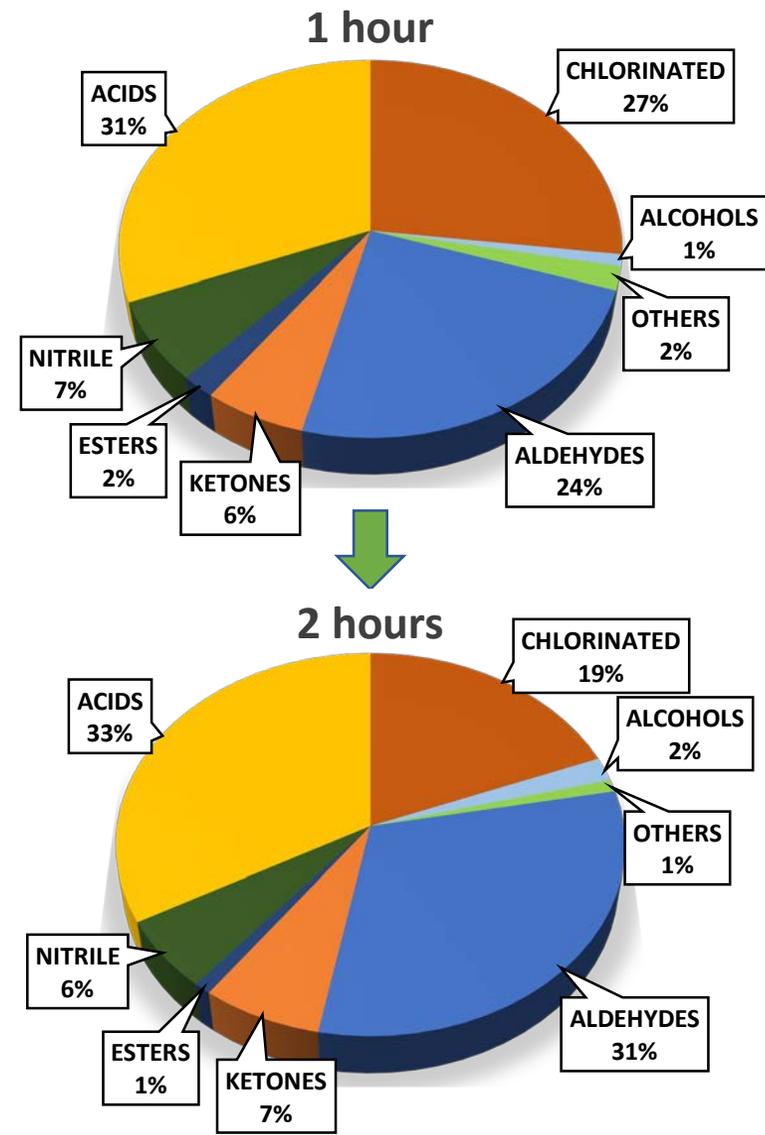
Electrocoagulation (EC)



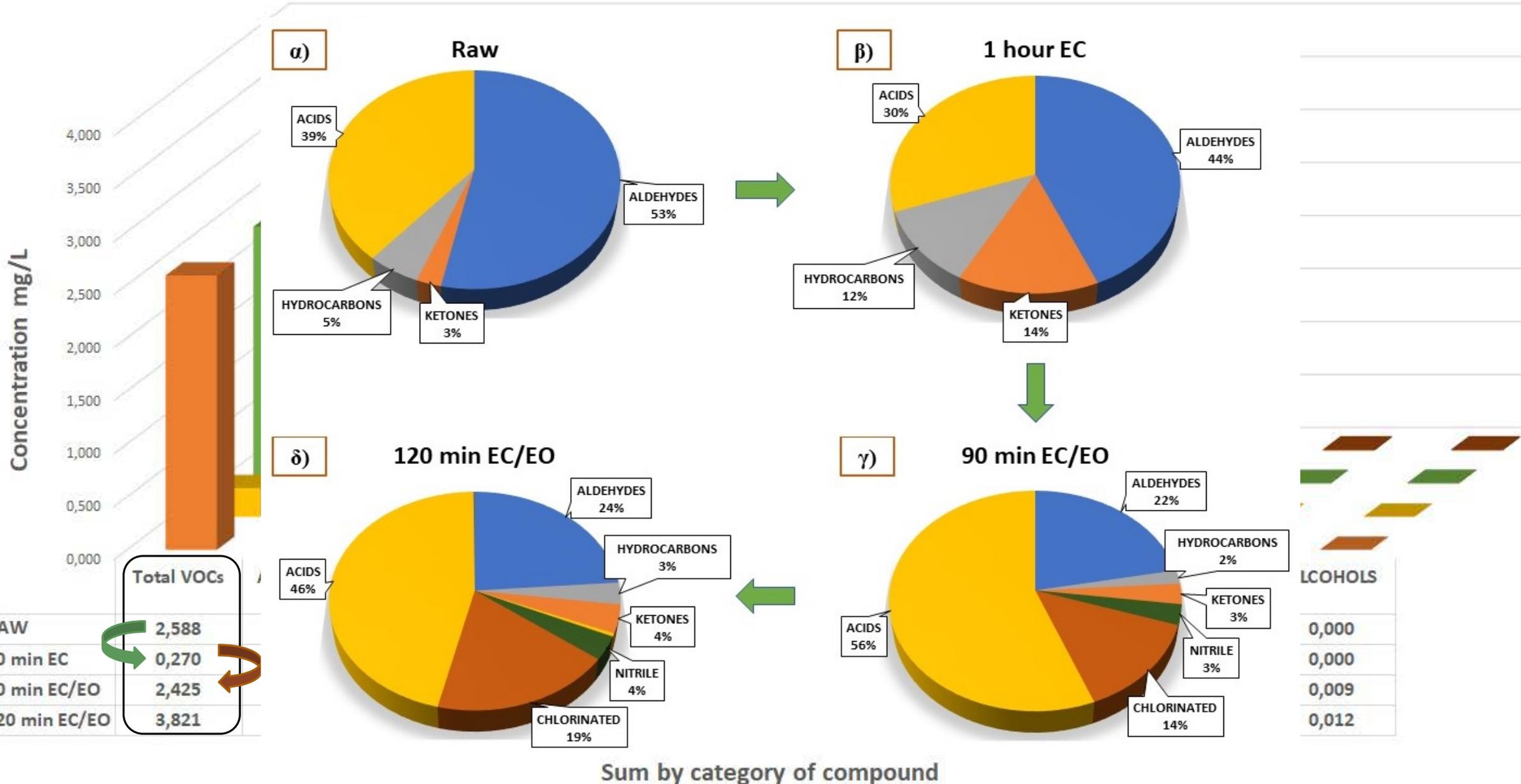
□ **VOCs of EO-treated (2 h) cheese whey wastewater (n=3)**

Categories of VOCs	Optimized EO process	
	Σ mg L ⁻¹ (±CL)	Efficiency (↓ ↑)
Aldehydes	1.074 (±0.253)	↓ 22.1%
Acids	1.155 (±0.281)	↑ 13.2 %
Ketones	0.236 (±0.050)	↑ 71.6 %
C-H	0.010 (±0.003)	↓ 92.7 %
Alcohols R-OH	0.062 (±0.019)	
Esters 	0.042 (±0.007)	
Nitriles R-C≡N	0.212 (±0.012)	
Chlorinated R-Cl	0.659 (±0.150)	
Furans 	0.021 (±0.005)	
Total VOCs	3.471 (±0.816)	↑ 25.4 %

Electrooxidation (EO)

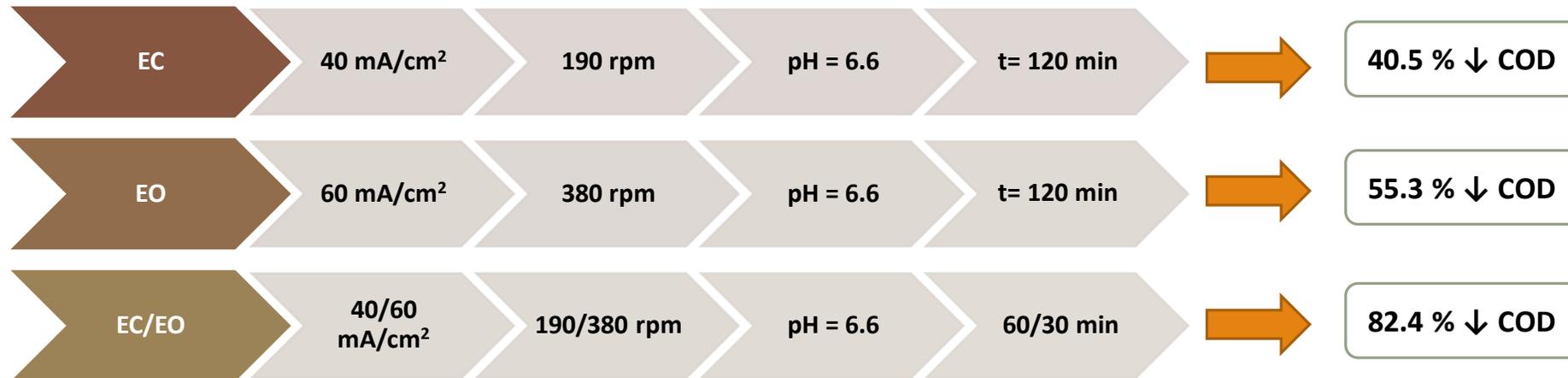


□ VOCs of EC/EO treated cheese whey wastewater (n=3)



Conclusions

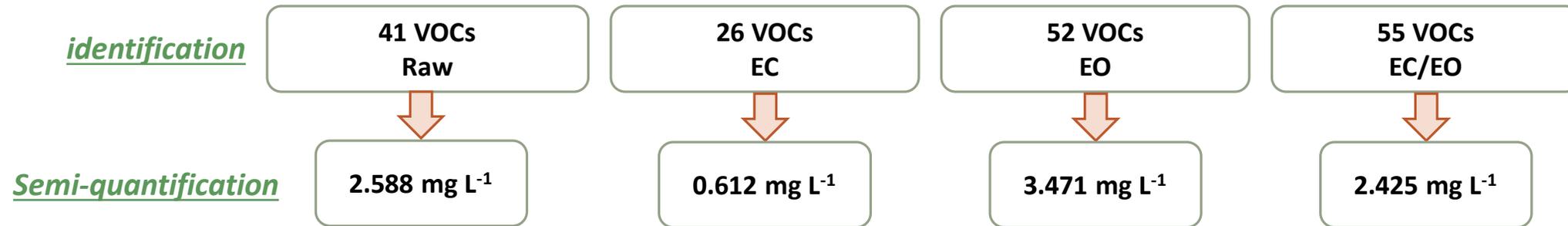
1) Implementation and optimization of EC and EO operating parameters



2) EC showed good efficiency in reducing all metered parameters, including VOCs

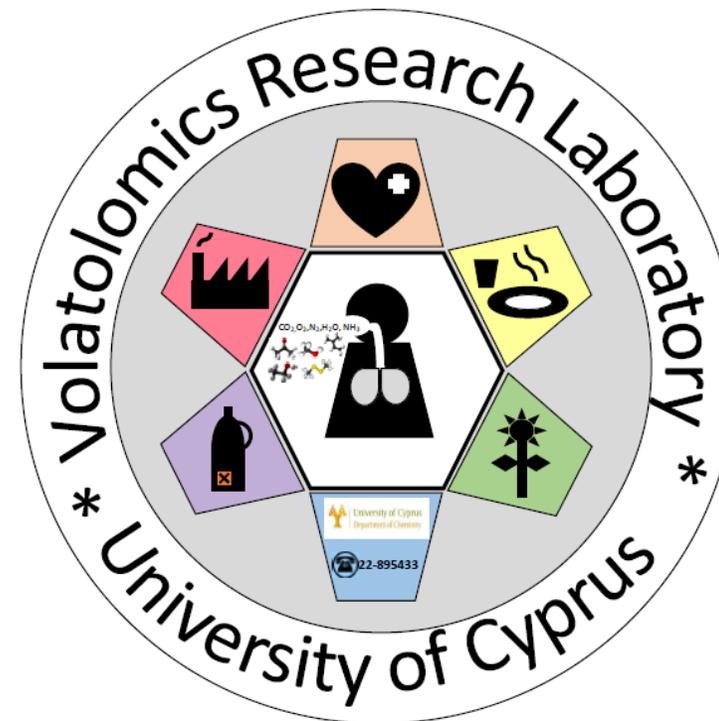
3) EO failed to reduce TPCs, PO₄⁻³, NO₃⁻ and Cl⁻

4) The HiSorb TD-GC / MS method identified and semi-quantified (Total VOCs):



In conclusion

EC appeared to be a favorable process for the reduction of VOCs emitted, while conjugated EC/EO as more effective for the reduction of cheese whey wastewater COD



Thank you!