





Fermentation of CO<sub>2</sub> and H<sub>2</sub> for formic acid production by *Thermoanaerobacter kivui* 

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### **Carbon dioxide**

- Most abundant anthropogenically produced greenhouse gas
- Rapid increase in CO<sub>2</sub> concentration
- Direct correlation between the increase in the CO<sub>2</sub>
  concentration and the rise in surface earth temperature



### CO<sub>2</sub> can be converted into fuels or chemicals



"Microbial cell factory (MCF) is a production facility of organisms that aims at achieving a high yield of metabolites by rewiring and optimizing the production process with the help of engineering and biology. "

### **Biorefinery concept**



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### Literature background

- Resting cells of *T. kivui* were proven by F.M. Schwarz and V. Muller as highly efficient whole-cell biocatalysts for the direct hydrogenation of CO<sub>2</sub> to formate.
- The addition of 300 mM of KHCO<sub>3</sub> switches the culture to the production of formate instead of acetate by inhibiting the further downstream processing of formate.



10.1186/s13068-020-1670-x

## **Goal of this study**

Prove that the addition of 300 mM of  $KHCO_3$  leads to the production of formate not only in the case of resting cells but also of replicating cells.

10.1186/s13068-020-1670-x



### **Experimental set-up**

Bacteria in autotrophic conditions in serum bottles placed in an incubator

Spectrophotometer used to measure optical density

High-performance liquid chromatography (HPLC) used to quantify acetic acid and formic acid



Autotrophic  $(H_2:CO_2 = 80:20)$ 

Autotrophic + 300m M KHCO<sub>3</sub> Autotrophic + 300m M KCl



### **Effect of bicarbonate on the Optical Density**



time (h)

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## Effect of bicarbonate on acetate and formate production





### Conclusions

*T. kivui* is able to convert  $H_2$ +CO<sub>2</sub> to formic acid in presence of bicarbonate



- Fest in laboratory scale bioreactor
- Investigation of the production of formate varying the pressure of CO<sub>2</sub> and H<sub>2</sub>





![](_page_11_Picture_1.jpeg)

![](_page_11_Picture_2.jpeg)

## Thanks for the attention!

![](_page_12_Picture_0.jpeg)

# Any questions?

![](_page_12_Picture_2.jpeg)

![](_page_13_Picture_0.jpeg)

![](_page_13_Picture_1.jpeg)

![](_page_13_Picture_2.jpeg)

## Appendix

### Formic acid production methods

#### > acidolysis of formate salts, which are in turn by-products of other processes

- Important in the 1990s
- The disadvantage is the unavoidable production of sodium sulfate.

### coproduct with acetic acid in the liquid-phase oxidation of hydrocarbons

- Important before 1980s.
- Between 0.05 and 0.25 tons of formic acid are produced for every ton of acetic acid. The reaction product is a highly complex mixture, and a number of distillation steps are required to isolate the products and to recycle the intermediates.
- carbonylation of methanol to methyl formate, followed either by direct hydrolysis of the ester or by the intermediacy of formamide
  - practiced industrially

#### electro-chemical reduction of CO<sub>2</sub>

- Not yet produced in large scale
- $\sim$  require only a two-electron reduction of CO<sub>2</sub>

![](_page_14_Picture_12.jpeg)

## Resting cell

- ➢ For the preparation of resting cells, *T. kivui* was cultivated in 1L flasks in the growth media to the late exponential growth phase.
- Glucose and fructose grown cells were harvested at an OD600 of 1.7–2.0, CO-grown cells were harvested at OD600 of 0.6.
- The culture was centrifuged under anoxic conditions at 11,500 g and 4 °C for 10 min and was washed twice in imidazole buffer (50 mM imidazole– HCl, 20 mM MgSO4, 20 mM KCl, 2 mM DTE, 4 µM resazurin, pH 7.0). Afterwards, the cells were resuspended in the same buffer to a protein concentration of 1 mg/mL and kept in gas-tight Hungate tubes.
- All preparation steps were performed under strictly anoxic conditions at room temperature in an anaerobic chamber.
- > The cells were directly used for the subsequent cell suspension experiments.

![](_page_15_Picture_6.jpeg)

### KHCO<sub>3</sub> effects

- > ATP hydrolysis
- Inhibition of ATP synthesis

The bicarbonate modulate the relative affinities of the catalytic site for ATP and ADP

![](_page_16_Picture_4.jpeg)