

# Effect of different iron sources on vivianite formation and anaerobic sewage sludge digestion

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### Introduction

Phosphorus is an essential nutrient for all living organisms and is a key element for global food production as it is widely used as fertilizer. The mineral fertilizer use increased due to the global population growing in numbers as well as due to the changes in food consumption habits over the years. So, it is of paramount importance to exploit secondary sources of phosphorus like sewage sludge. Phosphorus is present in wastewater mainly due to household cleaners and human excrements and is also identified as one of the main causes of eutrophication in water resources. While phosphorus removal technologies are well-established and widely applied, phosphorus recovery remains a challenge. Formation of Vivianite (Fe<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>·8H<sub>2</sub>O), a hydrated insoluble iron phosphate mineral, could increase the share of magnetically recoverable P in sludge by higher Fe dosing. The present study concentrates on the effects of different iron sources like Ferrous (Fe<sup>2+</sup>), Ferric (Fe<sup>3+</sup>), Metallic Iron (Fe<sup>0</sup>) and recycled ferric iron on vivianite formation and biomethane production during anaerobic sludge digestion. Aim

## Results

Vinit Incoulum Substrateinv Incoulum 6,74inv Incoulum 6,74inv Inv Intinv Inv <b< th=""><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th></b<>												
Image: bir	Sludge Characteristics											
IncoluinRefInfoIgf	Parameter	pН	ORP	Conductivity	TS	VS	TSS	VSS	TDS	Total P	Total COD	
3.9.2      3.9.3      3.1.1      3.9.3      2.4,47      2.9,60      1.9,60      1.9,60      1.9,60      0.1,10      0.1,10      4.1,10        Substrate      6,77      21,5      3,12      14,86      9,70      13,48      10,18      1,960      0,460      14,58        Parameter      K      Na      Ca      Mg      Fe      Dissolved TOC      Dissolved TN      Dissolved P      Dissolved Sulfate        Unit      mg/l      find      147,8      914,7      41,6      210,0      21,0        Substrate      312,2      402,2      166,8      22,4      n.d.      147,8      914,7      41,6      210,0      21,0        Parameter      Cr      Cu      Mn      Ni      Cd      Pb      Zn      K      Na      Ca        Jint      g/kg	Unit	-	mV	mS	g/l	g/l	g/l	g/l	g/l	g/l	g/l	
VisionVisio	Inoculum	8,24	-6,9	8,11	37,99	24,47	29,66	19,80	1,20	0,140	47,70	
Parameter UnitKNaCaMgFeDissolved TOCDissolved TNDissolved PDissolved PUnitmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lInoculua193,6251,1104,314,7n.d.147,8914,741,60210,0Substrate312,2402,2166,822,4n.d.414,1378,1121,0021,0ParameterCrCuMnNiCdPbZnKNaCaUnitg/kgg/kgg/kgg/kgg/kgg/kgg/kgg/kgg/kgg/kg	Substrate	6,77	21,5	3,12	14,86	9,70	13,48	10,18	1,96	0,460	14,58	
KNaCaMgFeDissolved TOCDissolved TNDissolved PSulfateUnitmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lInoculum193,6251,1104,314,7n.d.147,8914,741,6210,0Substrate312,2402,2166,822,4n.d.414,1378,1121,021,0ParameterCrCuMnNiCdPbZnKNaCaUnitg/kgg/kgg/kgg/kgg/kgg/kgg/kgg/kgg/kgg/kg	Liquid Phase											
Ing/i      Ing/i <th< th=""><th>Parameter</th><th>к</th><th>Na</th><th>Ca</th><th>Mg</th><th>Fe</th><th>Dissolved TOC</th><th>Dissolved TN</th><th>Dissolved P</th><th></th><th></th><th></th></th<>	Parameter	к	Na	Ca	Mg	Fe	Dissolved TOC	Dissolved TN	Dissolved P			
Substrate      312,2      402,2      166,8      22,4      n.d.      414,1      378,1      121,0      210,0        Parameter      Cr      Cu      Mn      Ni      Cd      Pb      Zn      K      Na      Ca        Jiii      g/kg	Unit	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l		
S12,2      402,2      100,3      22,4      Ind.      0 414,1      0 378,1      121,0      21,0      21,0        Parameter Unit      Cr      Cu      Mn      Ni      Cd      Pb      Zn      K      Na      Ca        Jnit      g/kg	Inoculum	193,6	251,1	104,3	14,7	n.d.	147,8	914,7	41,6	210,0		
Parameter    Cr    Cu    Mn    Ni    Cd    Pb    Zn    K    Na    Ca      Unit    g/kg	Substrate	312,2	402,2	166,8	22,4	n.d.	414,1	378,1	121,0	21,0	]	
Unit g/kg g/kg g/kg g/kg g/kg g/kg g/kg g/k	Solid Phase											
	Parameter	Cr	Cu	Mn	Ni	Cd	Pb	Zn	К	Na	Ca	
Inoculum      1,17      1,15      0,49      0,16      0,00      0,00      3,33      7,39      21,73      59,35	Unit	g/kg	g/kg	g/kg	g/kg	g/kg	g/kg	g/kg	g/kg	g/kg	g/kg	
	Inoculum	1,17	1,15	0,49	0,16	0,00	0,00	3,33	7,39	21,73	59,35	

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The main purpose of this research is to study the recovery of phosphorus from municipal waste in the form of crystalline vivianite from the sludge produced during aerobic digestion. Also, to identify the parameters that affect the production of biogas and vivianite and on the other hand to identify possible symbiotic link between a wastewater treatment plant and a water treatment plant, where the one will be the producer of mineral vivianite sustainably and the other will be the iron source provider (recycled ferric iron) so that the whole process will be in line with the circular economy model and sustainable development

### Materials & Methods

- Activated sludge samples were obtained from excess sludge (Larnaca WWTP) before mechanical thickening by a screw thickener. As Inoculum an anaerobic sludge from the Metamorphosis Wastewater Treatment Center (KELM) is used
- For vivianite formation sludge needs to be anaerobically digested, and iron must be added to the sludge.
- The experiments are performed in triplicates batch-wise [1]
  Different iron sources are tested for vivianite formation during anaerobic sludge digestion and compared to a control sample without iron addition.
  VS Inoculum to substrate Ratio is set to 1 and the digestion period was approximately 20 days.
  The amount of iron that is added is calculated based on the stoichiometrically molar ratio of Fe:P= 1.5 that is required for vivianite formation.

Substrate	0,05	0,18	0,24	0,04	0,00	0,00	0,89	15,31	13,28	203,12	20,26	20,48	
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Table 1: Characteristics of the Inoculum and activated sludge used in the batch experiments

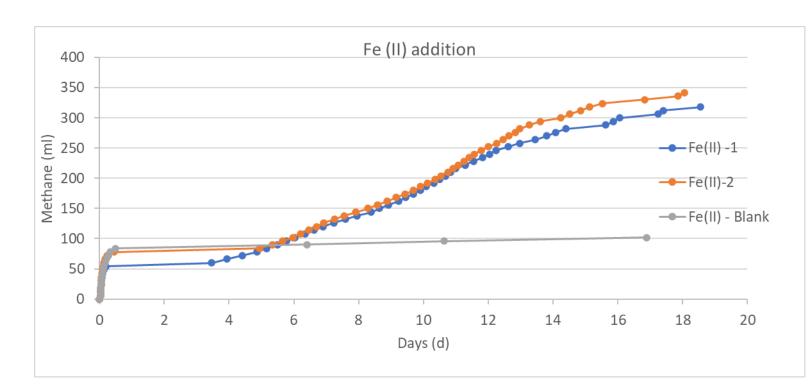


Figure 2: Methane production during anaerobic digestion of samples with excess of Fe (II) source

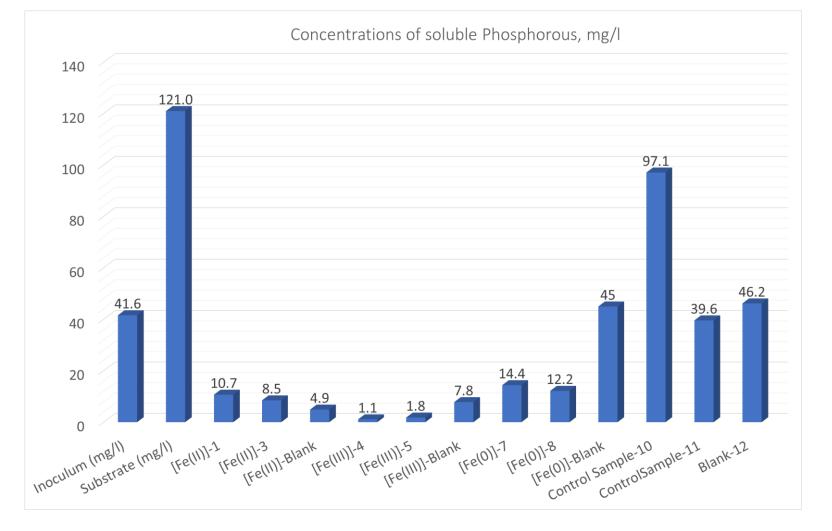


Figure 3: Concentration of soluble phosphorous before and after anaerobic digestion Figure 2 presents typical methane production during anaerobic digestion of samples with excess of FeCl<sub>2</sub> as Fe(II) source. After the 4<sup>th</sup> day, the methanogenic microorganisms start methane production. There is almost a linear increase in the methane production until about the 10th day when a decline in the production rate is observed up to the 18<sup>th</sup> day when methane production nearly stops. The initial linear methane production increase may be due to consumption of readily biodegradable organic material by microorganisms firstly, and then the biodegradable remaining organic fractions.

After anaerobic digestion, soluble phosphorus (Fig. 2) in all samples occurs in lower concentrations than initially, which indicates the formation and presence of vivianite in the sludge





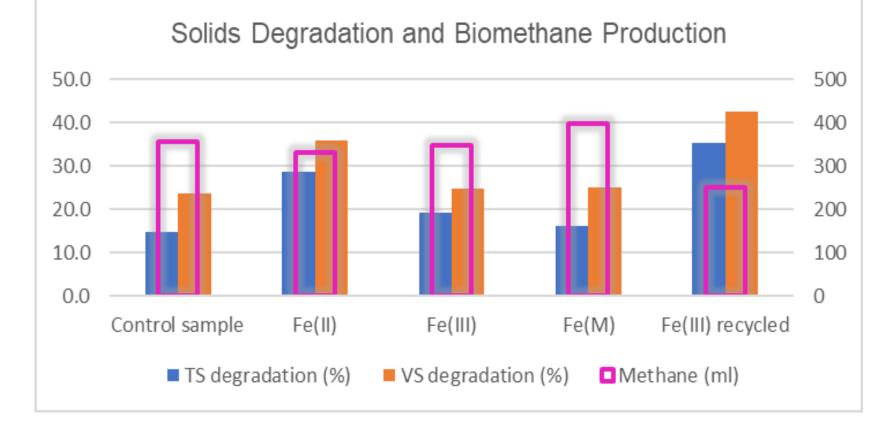


Figure 4: Solids degradation and Biomethane production per category iron source added

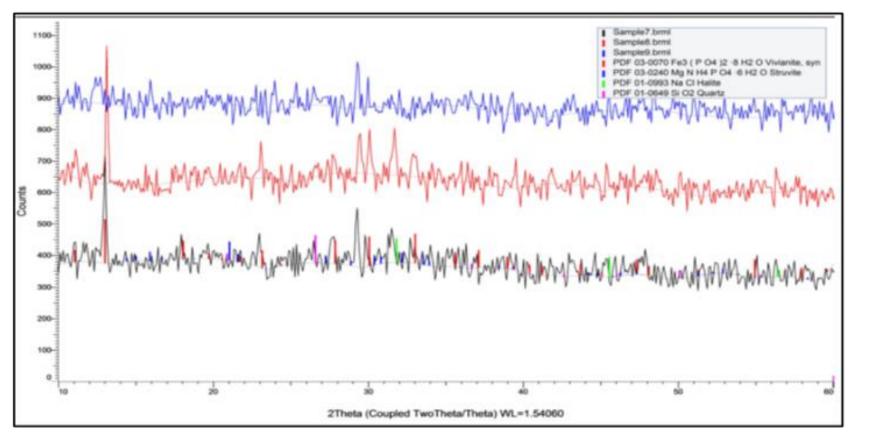


Figure 4: XRD analysis of samples with additional metallic iron

# Conclusions

This study shows that the use of extra iron during anaerobic digestion of the sludge can affect to some extent biomethane production i.e. 60-70 mlCH4/gVS were produced when an excess of ferrous or ferric iron is added, while 90-100 mlCH4/gVS were produced by control samples (without excess iron) and samples with metallic iron source. Vivianite formation is demonstrated in digested sewage sludge for each iron source tested. Almost all phosphate in sewage sludge can be bound in vivianite. XRD analysis proves presence of vivianite and Optical Microscope analysis indicates its presence as free particles. This fact could potentially allow vivianite separation from the sludge and high phosphate recovery.

The highest methane production per g TS and VS occurs in samples with metallic iron. It seems that the presence of ferric and ferrous iron slightly suppresses the production of biomethane. The highest solids degradations (Fig.4) were observed in the samples with these iron sources which may indicate that the production of CO<sub>2</sub> is favored over biomethane.

Vivianite formation with the characteristic peak at  $2\theta \cong 13^{\circ}$  can be observed in all samples with excess iron. Apart from, vivianite the crystallin structure of struvite, halite, calcite, and quartz can also be noticed. No formation of vivianite is detected in control samples (Fig.4).

Analysis performed for sample characterization before and after anaerobic digestion:

- Determination Of Total Solids (TS)
- Determination Of Volatile Solids (VS)
- Determination Of Total Suspended Solids (TSS)
- Determination Of Volatile Suspended Solids (VSS)
- Determination Of Total Dissolved Solids (TDS)
- Determination Of Metals Cd, Cr, Cu, Mn, Ni, Pb, Zn With Atomic Absorption
- Determination Of Total Organic Carbon (TOC)
- Determination Of Total Nitrogen (TN)
- □ XRD ANALYSIS
- Optical Microscope Analysis (OM)

# References

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