

# Parameters for removing sulfur from Dibenzothiophene using ultrasonication system and biodesulfurization microbes



Stylianou Marinos<sup>1,2</sup>, Samanides Charis<sup>3</sup>, Tsiampartas Antonios<sup>1</sup>, Kallis Christos<sup>2</sup>, Agapiou Agapios<sup>1</sup>, Vyrides Ioannis<sup>3</sup>, Damianou Christakis<sup>3</sup>



<sup>1</sup>Department of Chemistry, University of Cyprus, Nicosia, Cyprus  
<sup>2</sup>NORTEST Ltd, Cyprus  
<sup>3</sup>Department of Chemical Engineering, Cyprus University of Technology, Limassol, Cyprus

\*Corresponding author: e-mail address: stylianou.a.marinos@ucy.ac.cy



## Introduction:

### The Problem → Sulfur removal from oil

- A promising “Eco Technology” is to employ Biodesulfurization (BDS), a process where:
  - the bacteria (liquid phase) are mixed with oil at ambient temperature and pressure and remove selectively, the organosulfur components from oil fractions
  - without degrading the carbon skeleton of the compounds.

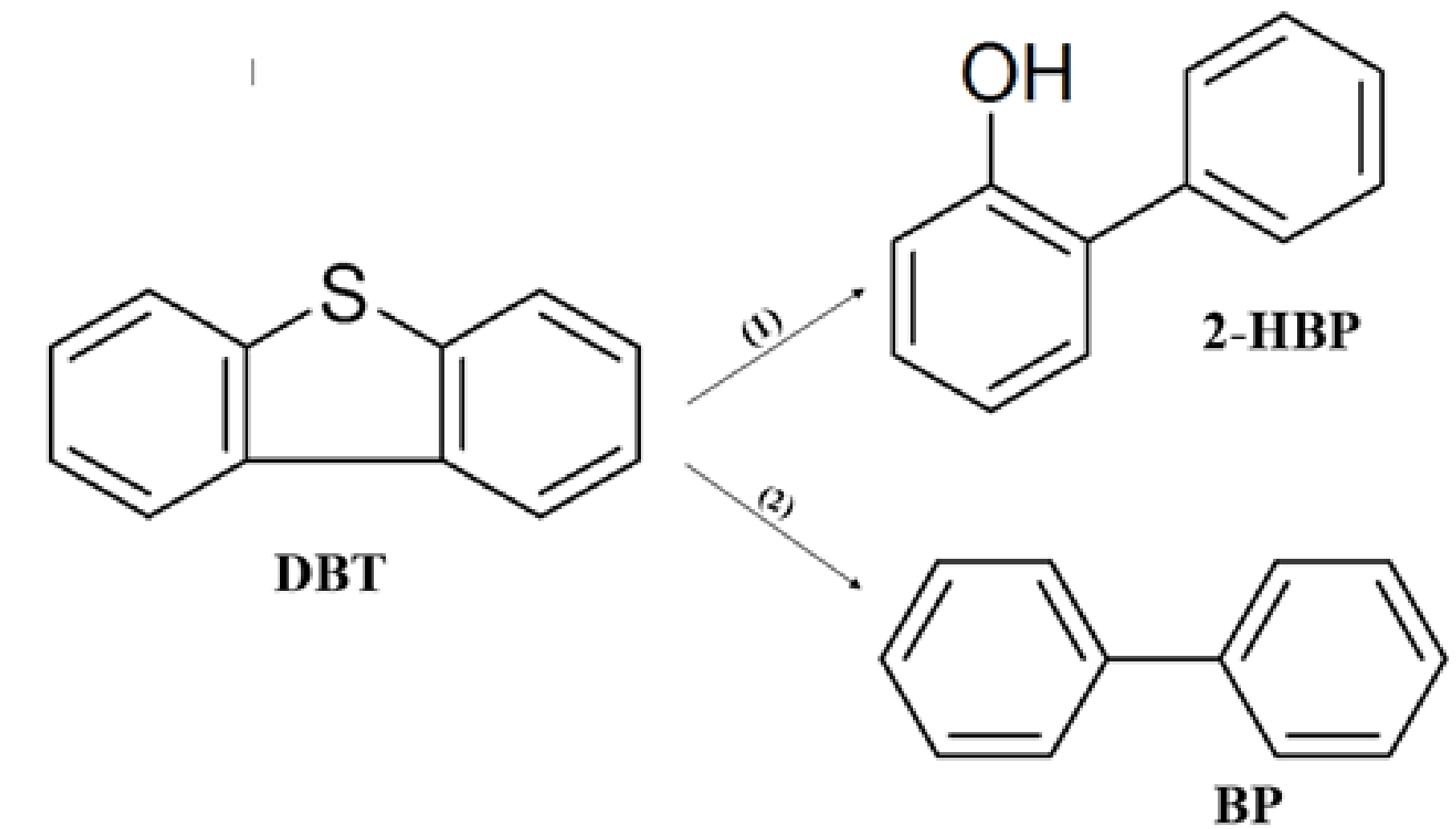


Figure 1. Desulfurization pathways: (1) aerobic microorganisms, and (2) anaerobic bacteria.

## Materials and methods:

### Ultra-sonication and Biodesulfurization



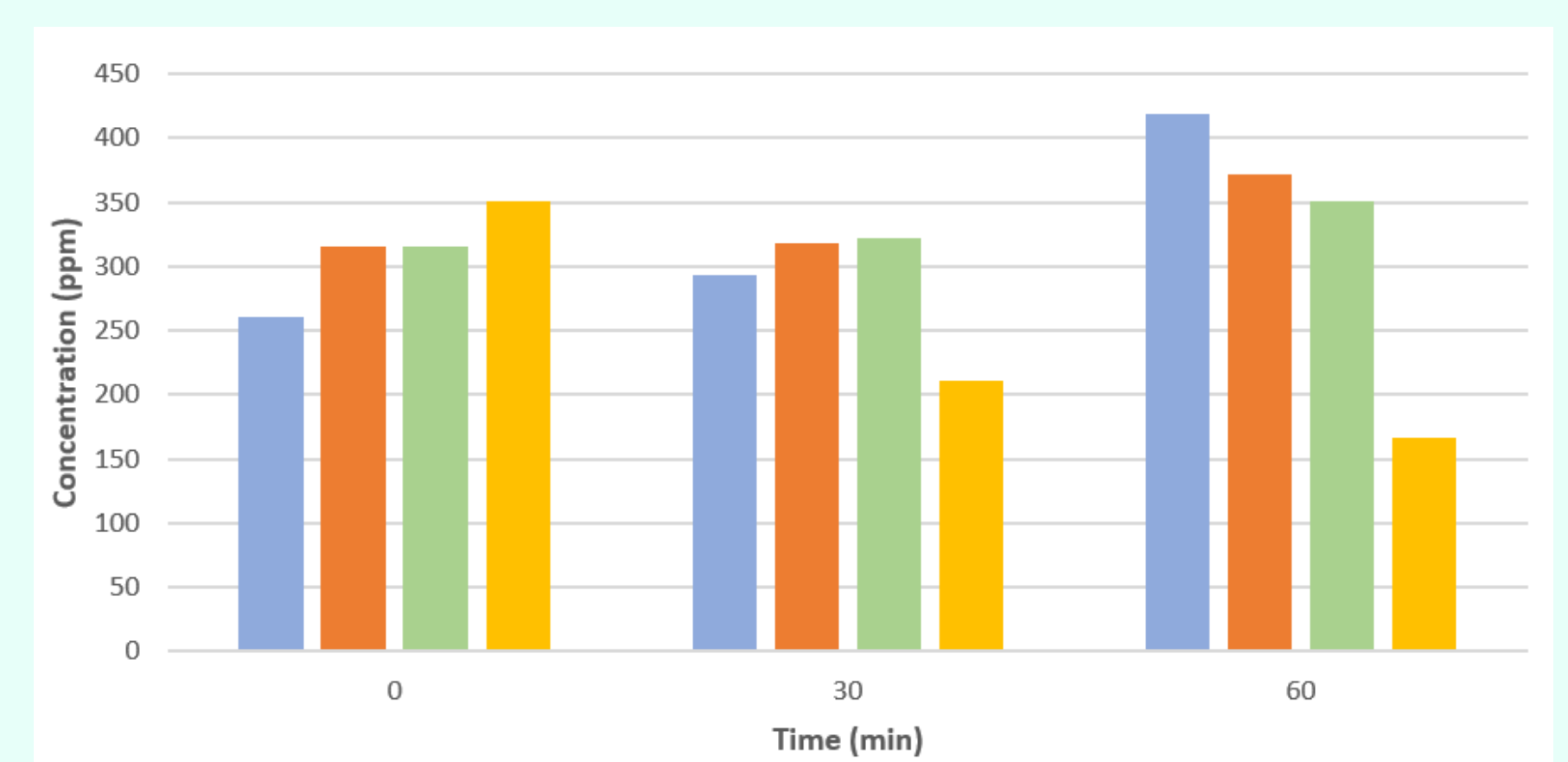
Fig. 4. Electronic system; front view.

#### (A) High intensity ultrasonic system for removing sulfur from DBT

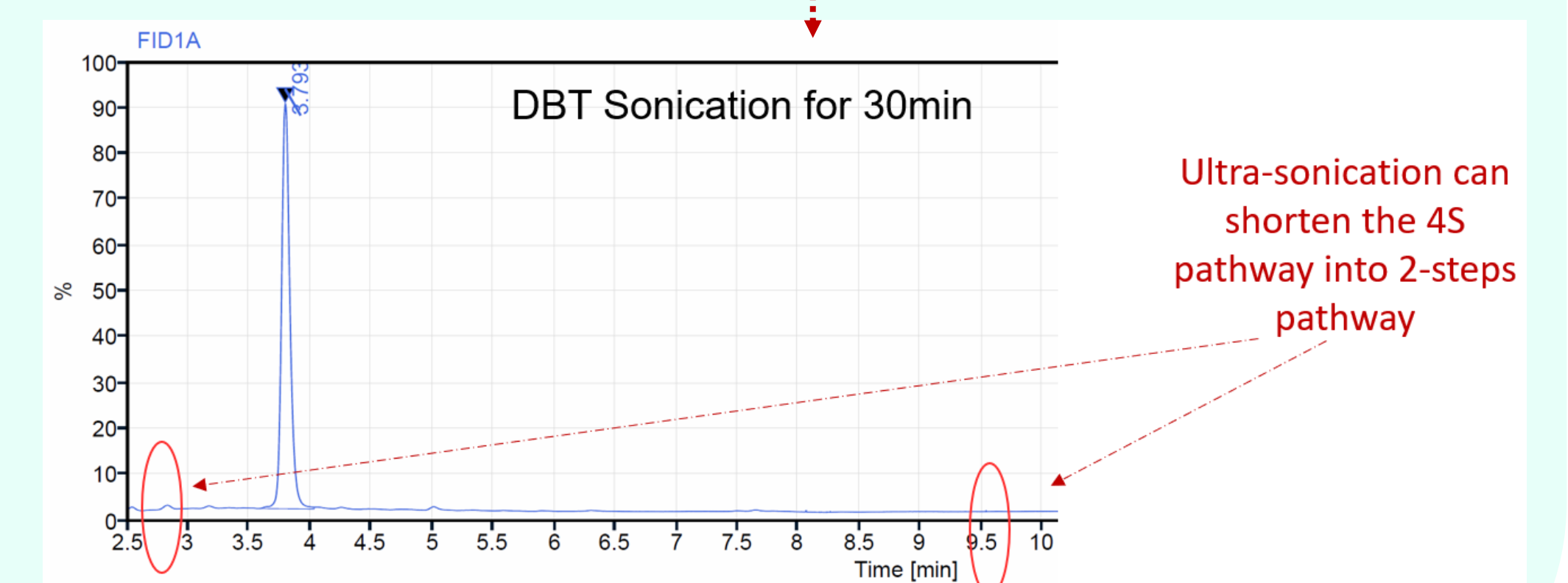
A circular unfocused ultrasonic transducer operating at 28 KHz was used (Fig. 4)

Samples	Solution	Concentration	Temperature (°C)	Time (min)	kHz	Parameter measured
1	DBT	Aqueous	25; 70	30; 60	28	DBT (GC-FID)
2	DBT	Heptane	25	30; 60	28	DBT (GC-FID)

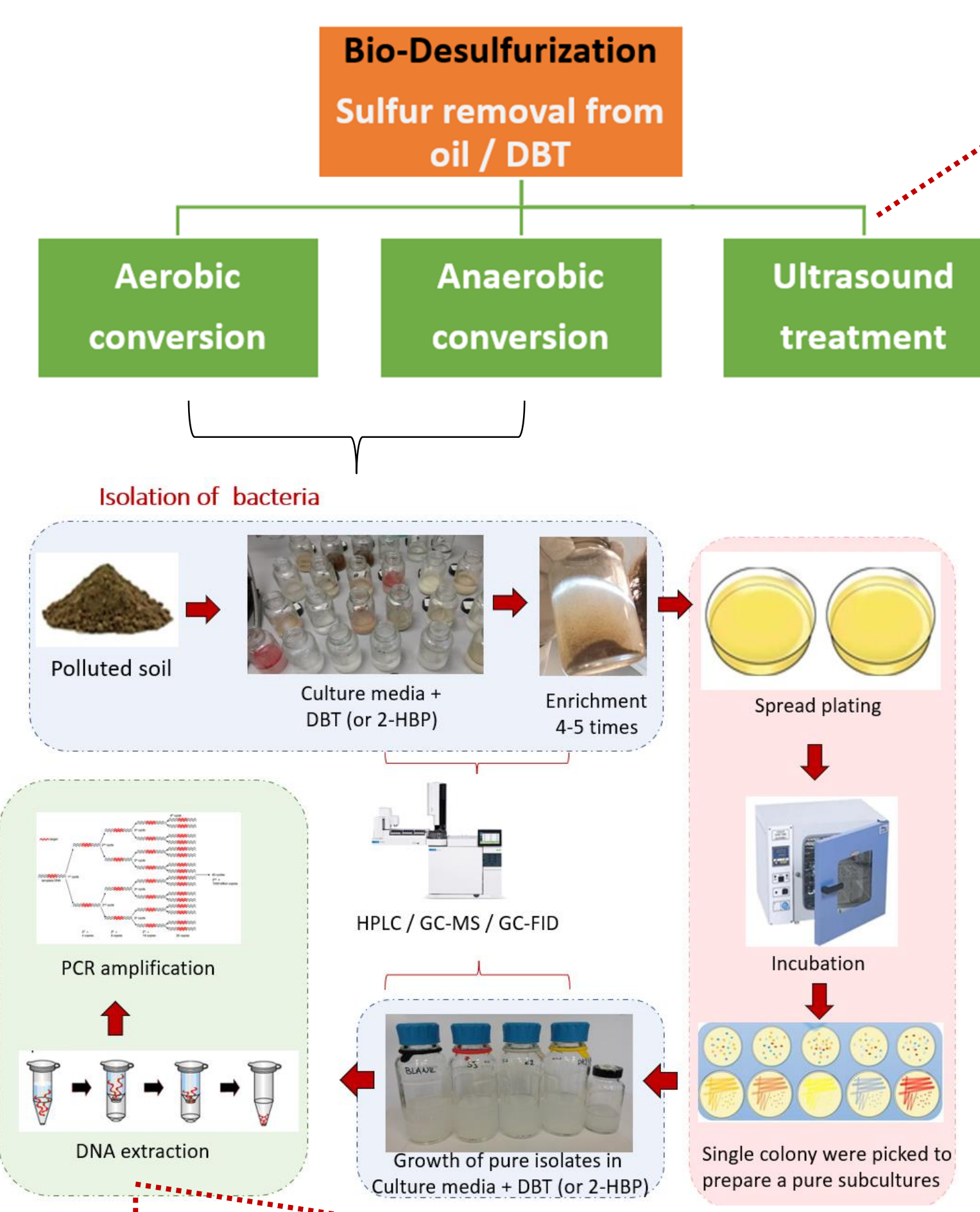
## Results:



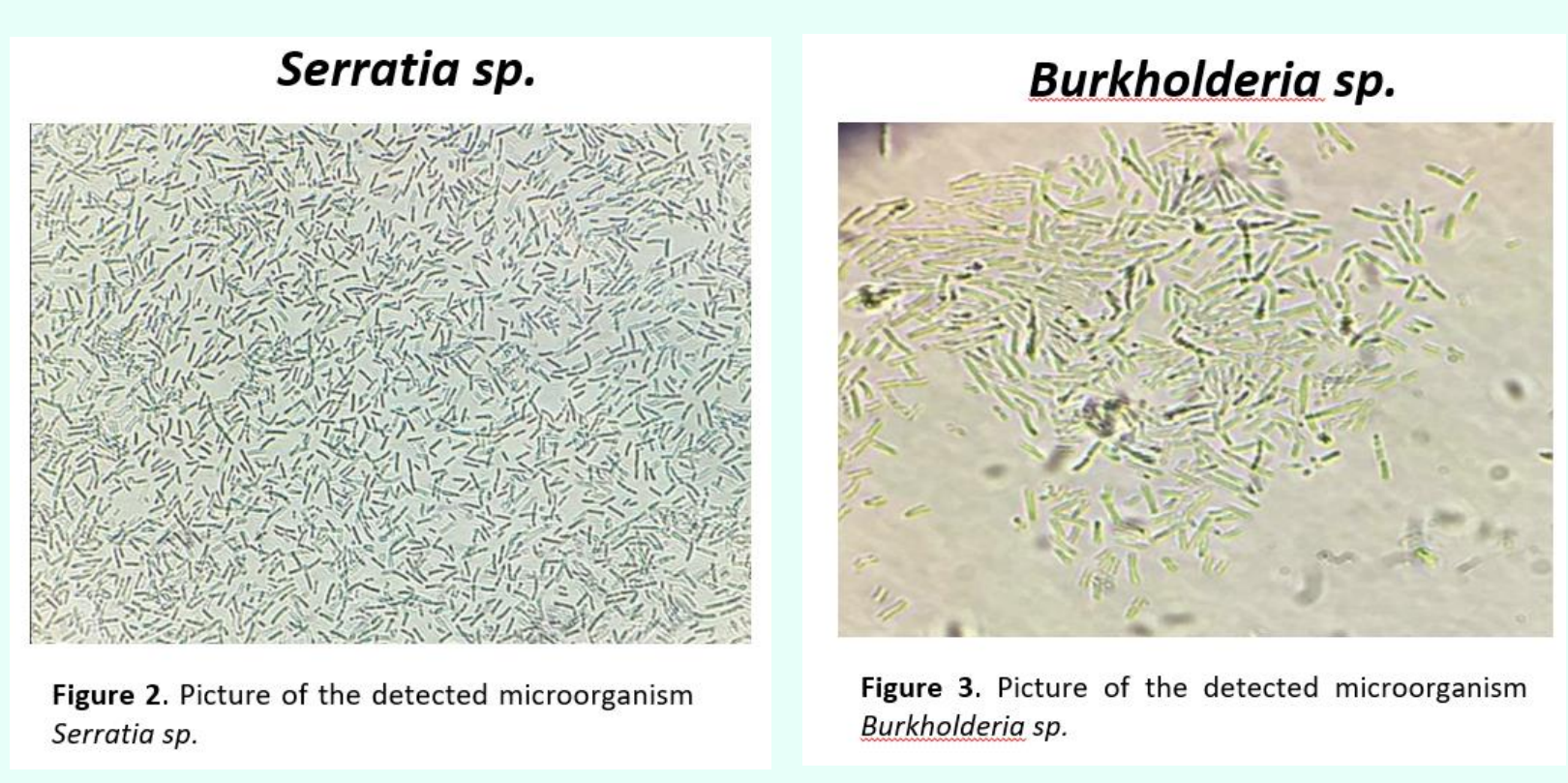
- An increase in DBT concentration is observed in both temperatures that were performed.
- On heated (T=70°C) experiment there is a lesser increase of DBT concentration
- Decrease of 2-HBP concentration



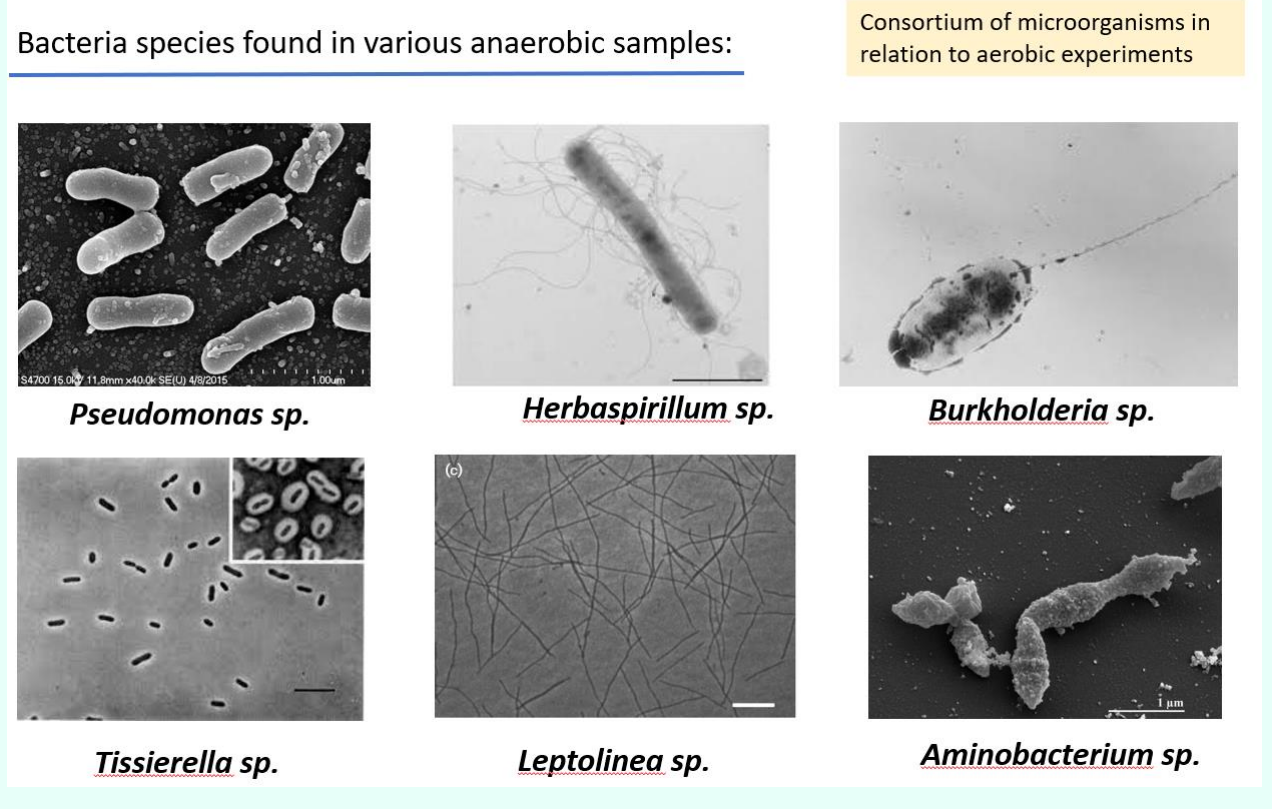
Ultra-sonication can shorten the 4S pathway into 2-steps pathway



### Isolation of aerobic bacteria

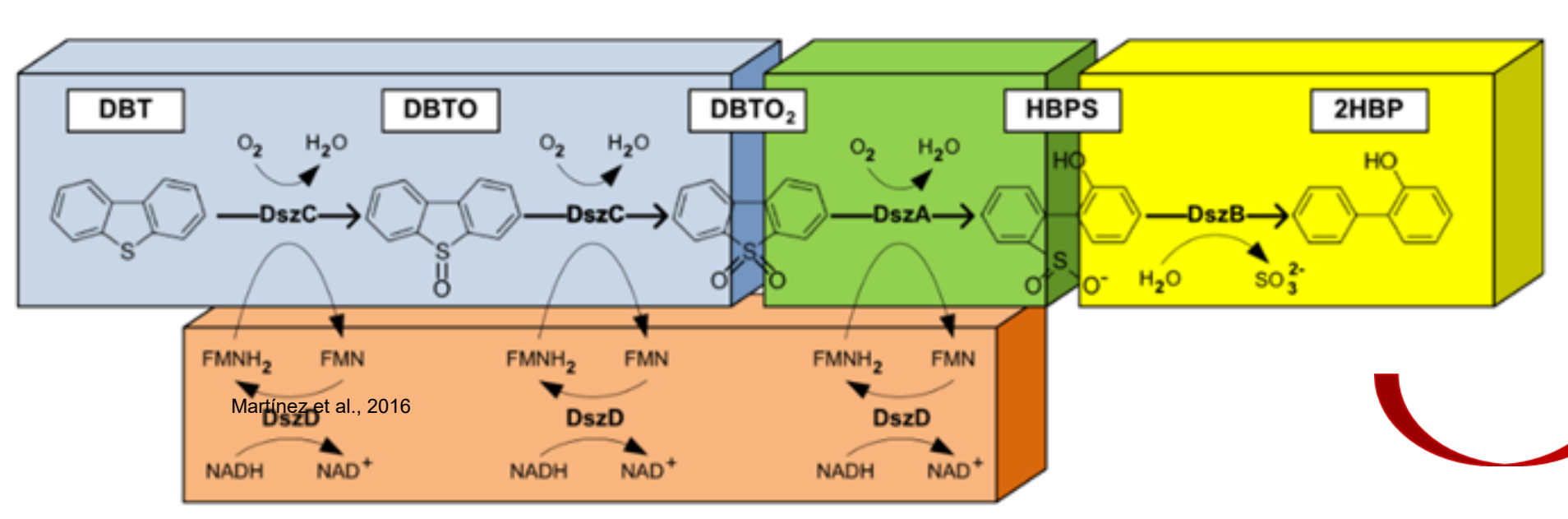


### Isolation of anaerobic bacteria



## Conclusions:

1. Sonication may enhance DBT solubility in an aqueous solution resulting in better performance through the 4S pathway
2. Through sonication process temperature is increased which may affect the solubility of the studied organic compounds



3. Ultra-sonication → from 4S pathway to 2-steps pathway

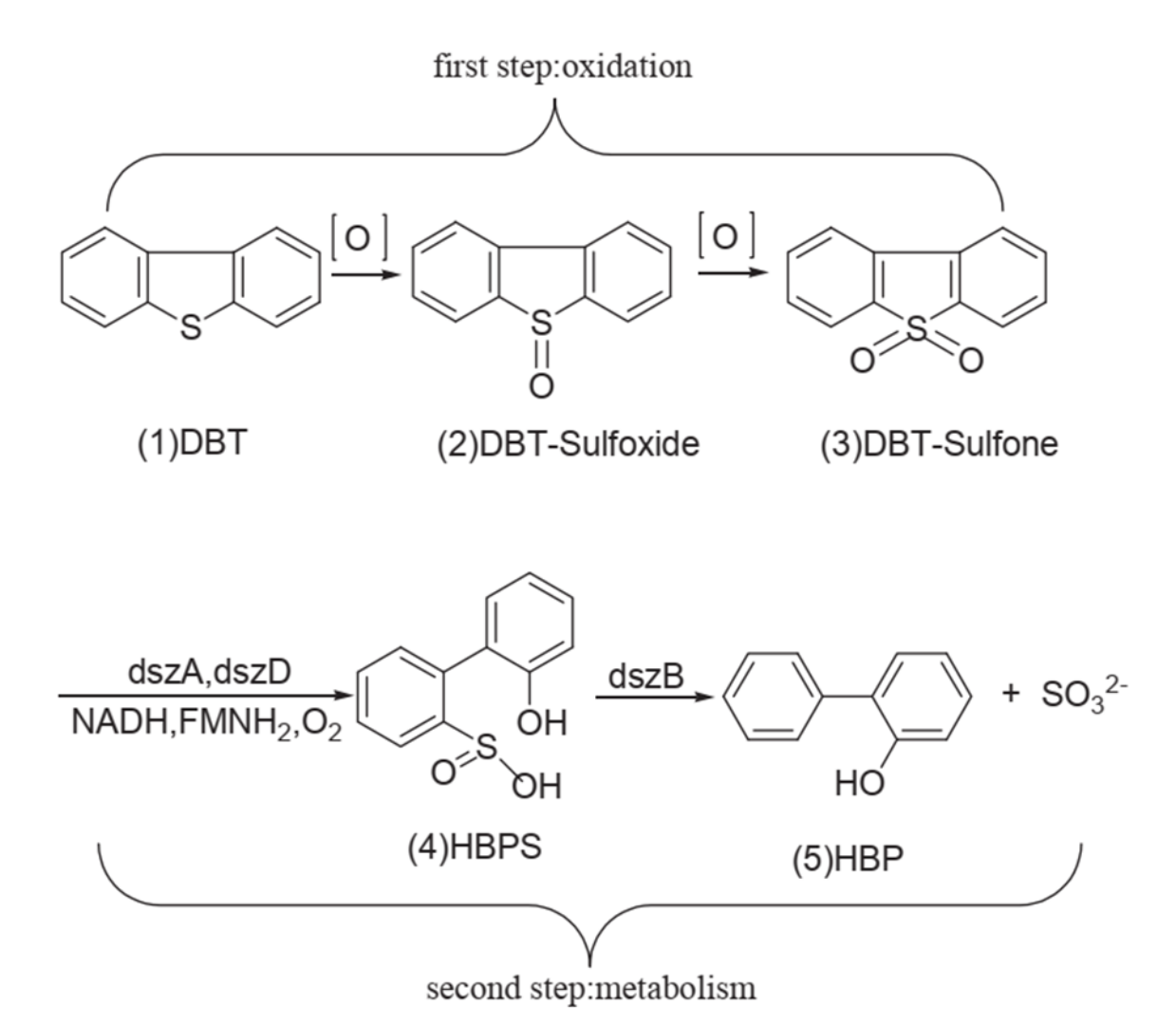


Fig. 5. The pathway of biodesulfurization of DBT by ultrasonic pre-oxidation [where [O] represent O<sup>o</sup> and -OH and H<sub>2</sub>O<sub>2</sub>] (Yi et al., 2019)

## References

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