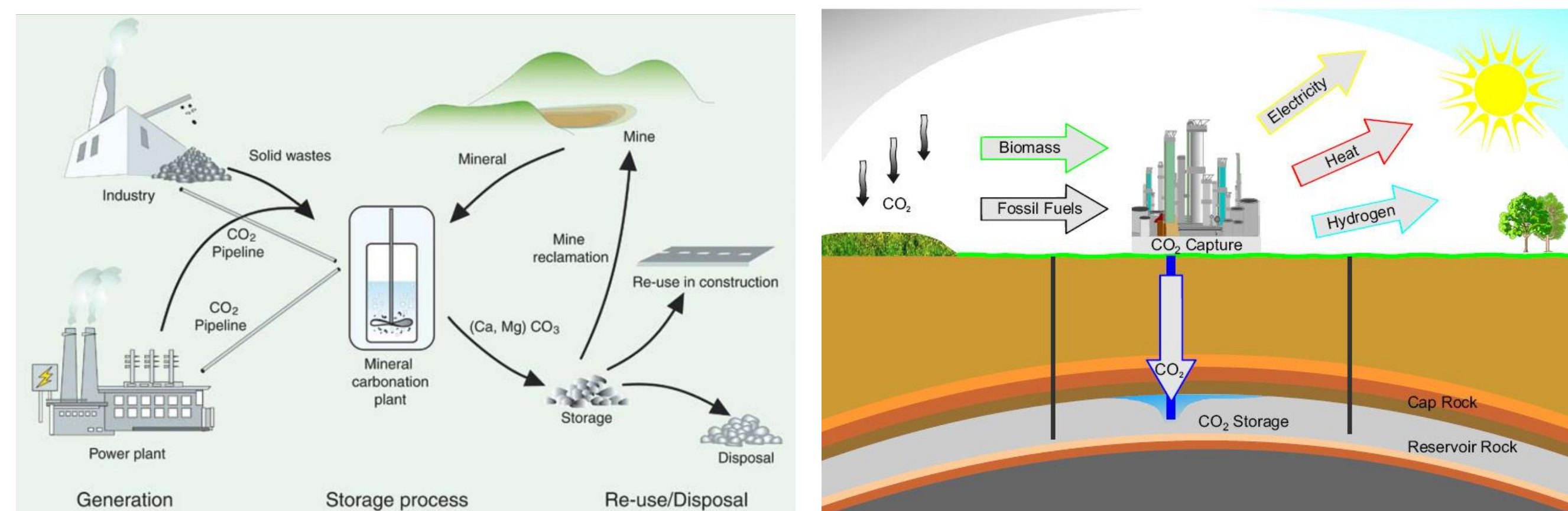


Introduction:

- This bibliographic study is focused on CO₂ storage possibilities in outcrop minerals through ex-situ methods.
- Carbon storage is an effective method for long-term storage of carbon dioxide into surface or subsurface rocks and minerals to combat climate change.
- At a large-scale, it can provide a viable solution to effectively lower GHGs emissions.
- Generally CO₂ storage is accomplished by two main methods:
 - ✓ **Surface process (*Ex-situ*)**
 - ✓ **Subsurface Injection (*In-situ*)**
- This CO₂ storage technique is accomplished through mineral carbonation. It is an attractive technology in which CO₂ chemically reacts with calcium, sodium, and magnesium containing materials to precipitate stable and environmentally harmless minerals, usually carbonates.



(Mazzoti et al, 2015)

Figure 1 and 2. Surface Ex-situ process (left) and Subsurface In-situ Injection process (right).

Materials and methods:

- There is a wide range of materials that can be used for mineral carbonation (Fig 3), divided into two main categories:
 - ✓ Naturally occurring formations such as:
 - ultramafic rocks (olivine, serpentine and wollastonite) and
 - zeolites
 - ✓ Wastes comprised of Mg/Ca-rich materials, such as:
 - fly ash,
 - iron and steel slags, and
 - ultramafic mine wastes (Li et al, 2018)
 - other wastes (red mud)
- Greece provides a variety of mine waste materials suitable for surface CO₂ storage, but this potential is yet underexplored.
- The present bibliographic study is focused on CO₂ sequestration by red mud, a solid residue produced during the alumina extraction.

Υπόμνημα

- Βιομηχανικά Ορυκτά *
- Βωξίτης
- Ζεολίτις
- Μάρμαρα-Διακοσμητικά Πετρώματα**
- Μεταλλεύματα***
- Νεκρωσικοί Λαγέρτες
- Πετρώματα-Φυσικό Αέριο
- Εγκαταστάσεις Αιοληκτών
- Ενότητα Παρνασσού
- Υπερμαφικά και Μαφικά Πετρώματα
- Μασίφ Ροδόνης

Βιομηχανικά Ορυκτά* = Αταπούλιτης, Μπετονίτης, Λευκά Ανθρακικά, Διατομίτης, Ασπίς, Γρανάτης, Γραφίτης, Γύψος, Ορυκτό Άλας, Χονδρίτης, Καλκίτης, Μαγνησίτης, Ολιβίνη, Φωσφορίτης, Περίτης, Πολυάνη, Κίσηρης, Χαλαζίας, Αμορφή Πυρίτις, Τάλη, Βερμικουλίτης, Βολαστονίτης, Ζεολίτης

Μάρμαρα-Διακοσμητικά Πετρώματα** = Λευκά έως έγχρωμα μάρμαρα, Δολομίτες, Τραβερτίνες, Ονυχές, Αλάβαστρο, Ψαμμίτες, Σχιστόλιθοι, Ηφαιστειακά Πετρώματα, Ζεολιθοφόροι Τόφοι

Μεταλλεύματα*** = Άργυρος, Χρυσός, Βωξίτης, Χαλκός, Χρόμιο, Σιδηροσίδηρος, Μαγγάνιο, Νικέλιο, Μόλυβδος, Σιδηροπυρίτης, Ψευδάργυρος

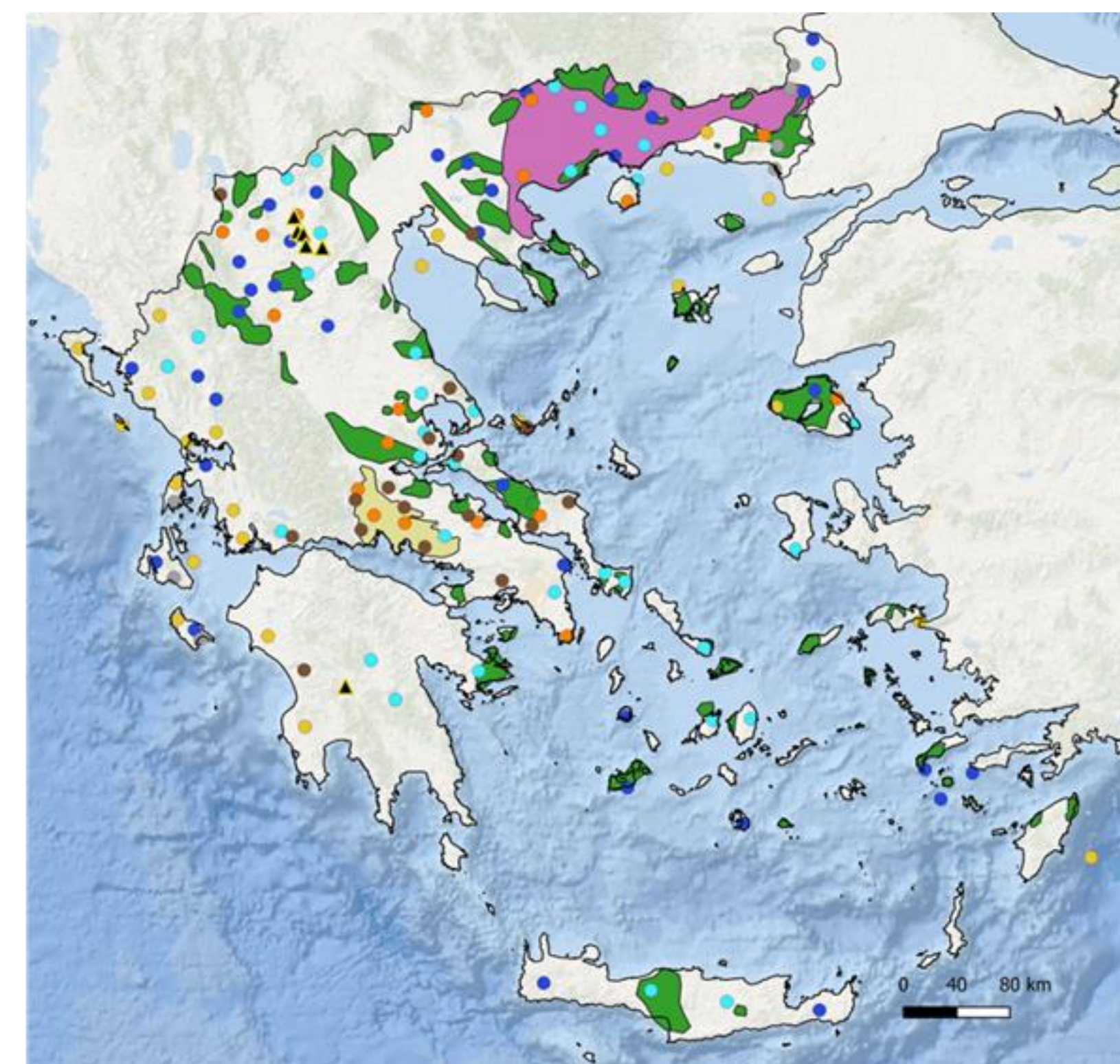


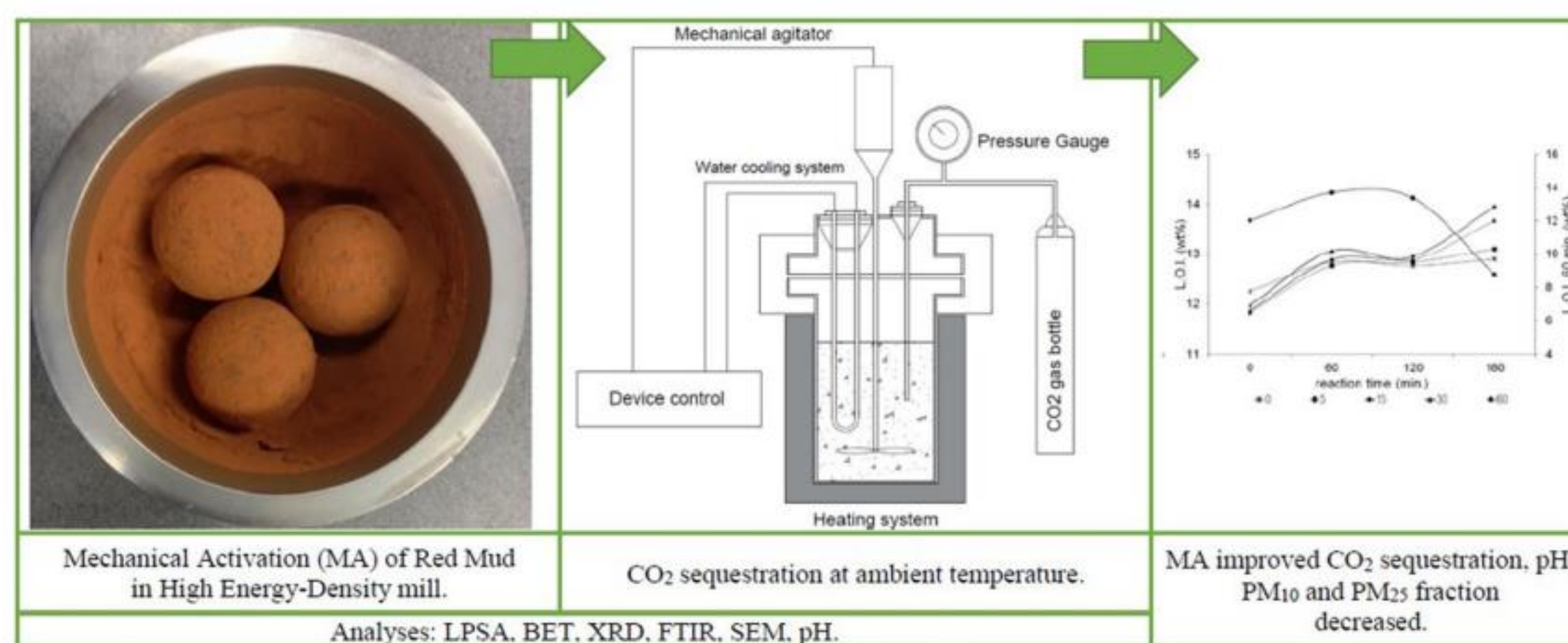
Figure 3. Map demonstrating possible material for CO₂ mineral carbonation

Results:

Table 1. Main properties and carbonation conversions of waste materials demonstrating their CO₂ storage effectiveness.

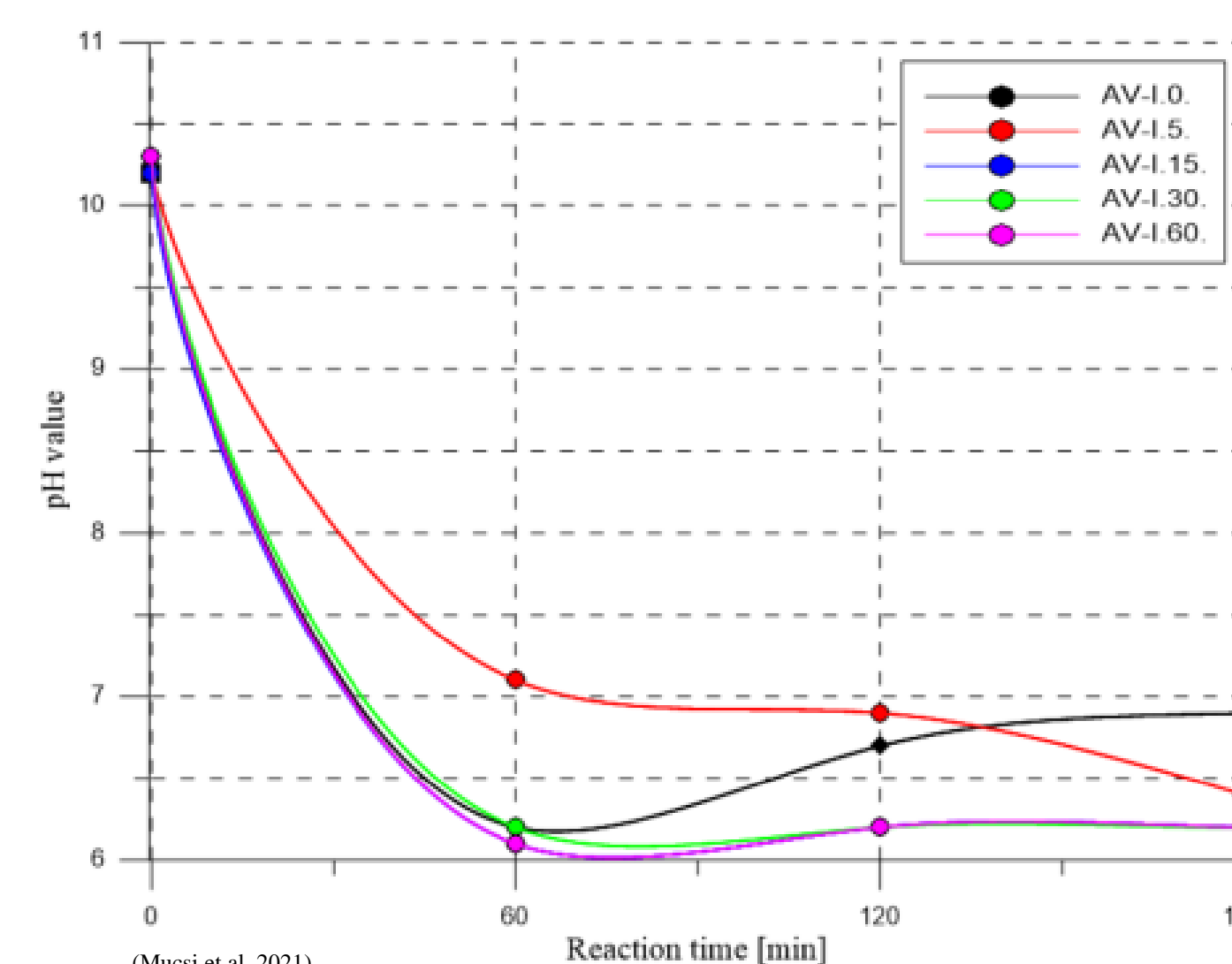
Material	CaO (%)	MgO (%)	Theoretical maximum CO ₂ uptake (tCO ₂)	Experimental CO ₂ Uptake (ECO%)
Fuel Combustion Ashes	1.3-10	1-3	6-9	2.6
Lignite Fly Ash	27.5	6.5	43	23
Ni Tailings	3.4	21-40	43	29
Red Mud	2-7	<1	7-19	4.15-7.2

(Data acquired from Sanna et al, 2014)



(Mucsi et al, 2021)

Figure 4. Mechanical activation of Red Mud to enhance CO₂ sequestration of red mud as shown by Mucsi et al, 2021.



(Mucsi et al, 2021)

Figure 5. Variation of pH as function of grinding time and reaction time that proved the CO₂ sequestration of red mud as shown by Mucsi et al, 2021.

Conclusions:

- Therefore, establishing CO₂ storage as a technique will:
 - ❑ Low CO₂ storage through mineral carbonation
 - ❑ Neutralize the waste
 - ❑ Low contribution in minimizing the Greenhouse Gas Emissions (GNG) emissions

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