

Utilization of new TiO₂ photocatalyst in purifier for removal of hazardous wastes



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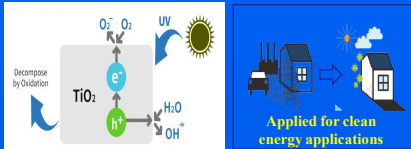
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Introduction & Experimental



TiO₂ is widely used in the field of photocatalyst because of its strong oxidizing power and excellent durability. But, it has a band-gap energy about 3.0 ~ 3.2 eV. This means that to generate photocatalytic activity, it is necessary to absorb the UV-light.

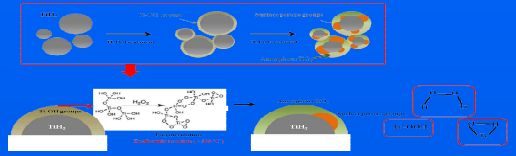
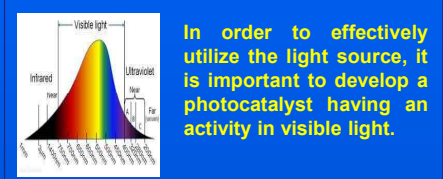
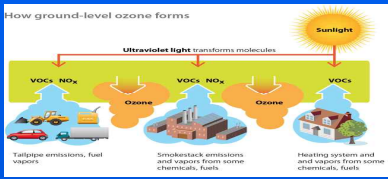


Fig. 1: Schematic of amorphous Ti-based peroxy complex (ATPC) synthesis



In order to effectively utilize the light source, it is important to develop a photocatalyst having an activity in visible light.

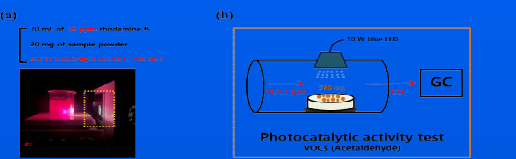


Fig. 2. Illustration of photocatalytic activity tests for removals of (a) RhB and (b) ACT.

How to remove hazardous wastes (VOCs) with new TiO₂ photocatalyst?

Results & Discussion

1. Photocatalytic activity test with Ti-based peroxy complex

2. Application test for air-purifier system

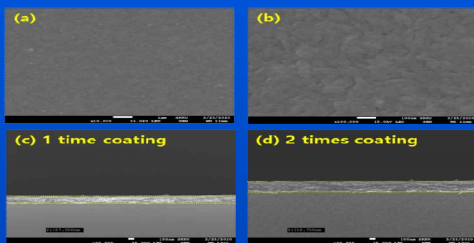


Fig. 3. SEM surface morphologies and cross-sectional images of spin-coated ATPC with different thickness.

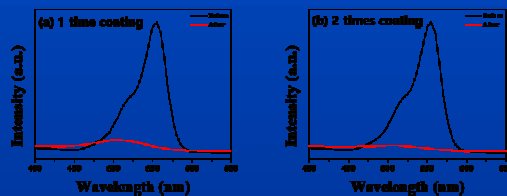
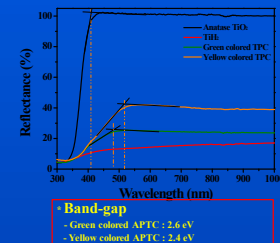


Fig. 4. UV-Vis absorption spectra measured before/after photocatalytic reaction of RhB dye on ATPC surfaces.



Band-gap
- Green colored ATPC : 2.6 eV
- Yellow colored ATPC : 2.4 eV

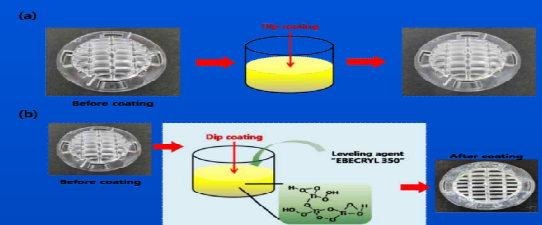


Fig. 7. Coating process of ATPC on ABS plastic filter in purifier with/without E-360 agent.

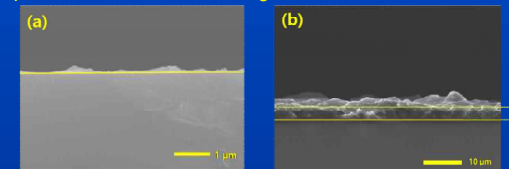


Fig. 8. Cross-sectional SEM images of dipping-coated ATPC coated layer on ABS plastic filter.

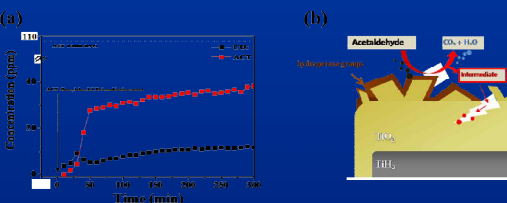


Fig. 5. Photocatalytic removal of ACT and reaction mech.

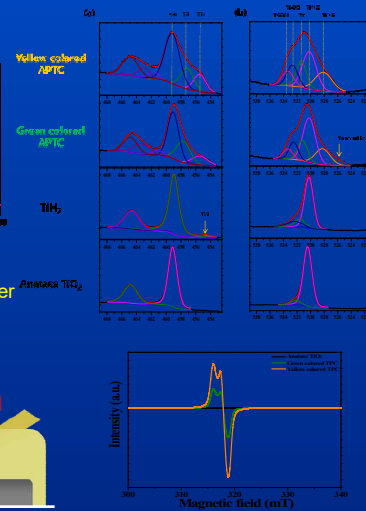
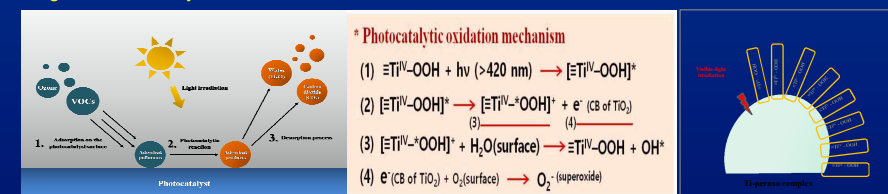


Fig. 6. XPS (upper) and EPR (lower)



Conclusions

In this study, amorphous Ti-based hydroperoxy complex (ATPC) was successfully synthesized by a facile method only using titanium hydride (metallic Ti) and H₂O₂ under mild conditions. TiH₂ is known to have many more reactive sites than metal oxides such as TiO₂. Therefore, TiH₂ was selected as a precursor and Ti based peroxy complexes were successfully synthesized for the first time. Qualitative and quantitative analyses through optical measurements revealed that the synthesized ATPC had many peroxy groups and various oxidation states of Ti (Ti²⁺, Ti³⁺ and Ti⁴⁺). The photocatalytic properties of successfully synthesized ATPC show very fast organic RhB decolorization rates, unlike other conventional visible-light catalysts. This is possible because many peroxy complexes can form more active radicals. Therefore, degradation experiment of VOCs such as acetaldehyde (ACT) in a gas phase was also carried out. Since an application test is one of the important items in the photocatalytic experiment as well as commercialization, we did fabricate a filter system in air-purifier by adapting synthesized ATPC, and then successfully removed air pollutants. Based on this study, we suggested a possible photocatalytic oxidation mechanism with ATPC.



Fig. 10. Photo images of air purifier adapted ATPC coated filter in a car.