



Multivariate analysis of pharmaceutical pollutants adsorption in aqueous media with tailored waste-based carbonaceous adsorbent materials and commercial activated carbons

J. Lladó¹, F. López¹, J.M. Rossell¹, C. Lao-Luque¹, R.R. Gil², E. Fuente², B. Ruiz²

¹Department of Mining, Industrial and TIC Engineering (EMIT), Escola Politècnica Superior d'Enginyeria de Manresa, UPC, Manresa, Spain
²Biocarbon, Circularity and Sustainability Group (BC&S), Instituto de Ciencia y Tecnología del Carbono (INCAR), CSIC, Spain



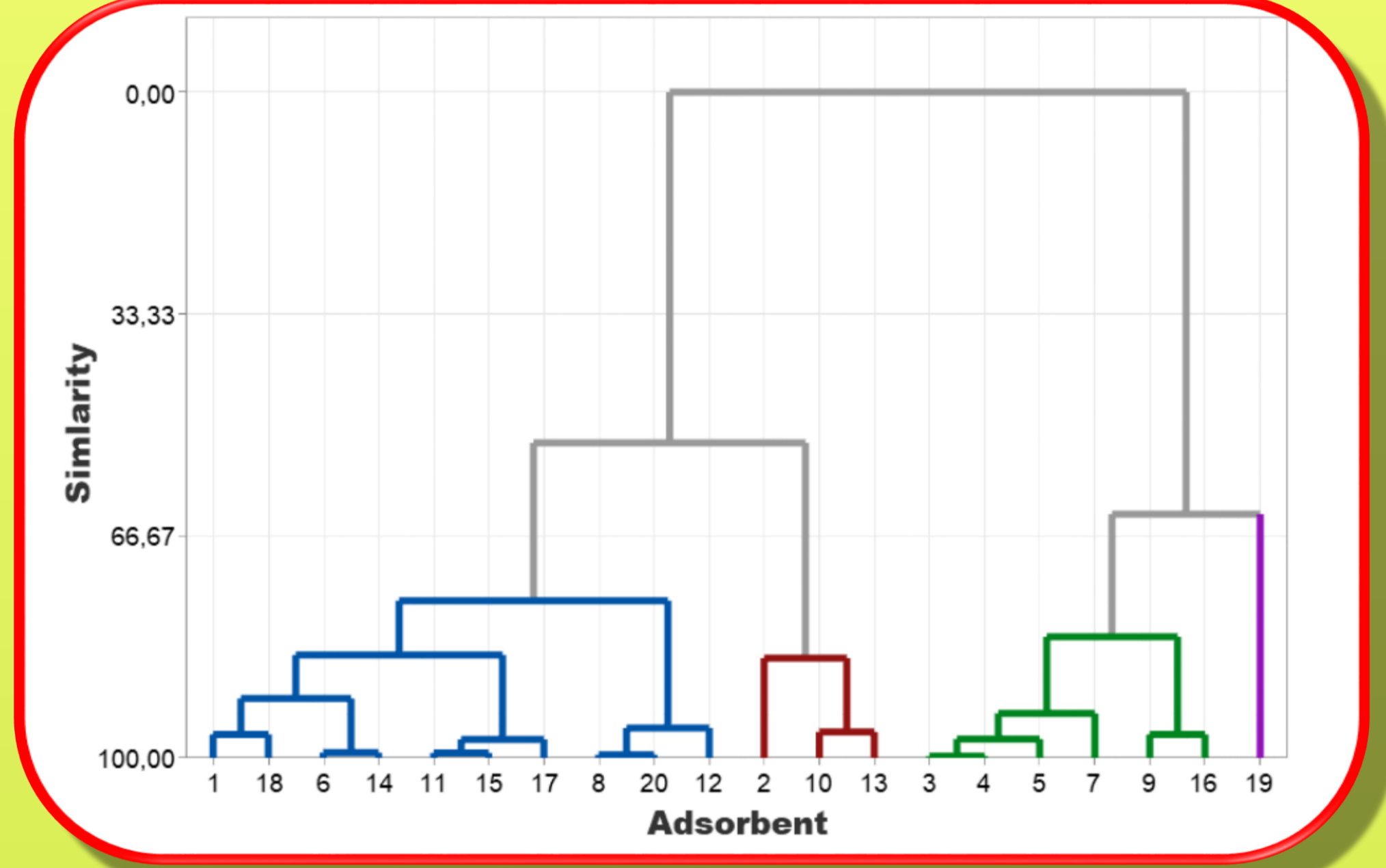
Introduction

- The purpose of this research is to determine the effectiveness of various biocollagenic waste-based activated carbons (BWAC) and a sludge biochar (SBC) in removing emerging pharmaceutical pollutants (phenol, salicylic acid, paracetamol, diclofenac and iodixanol) present in aqueous media and its comparison with commercial and manufactured adsorbent from different origins.
- Principal component analysis is applied to develop multiple lineal regression models to predict maximum adsorption capacities for future new waste based-activated carbons.

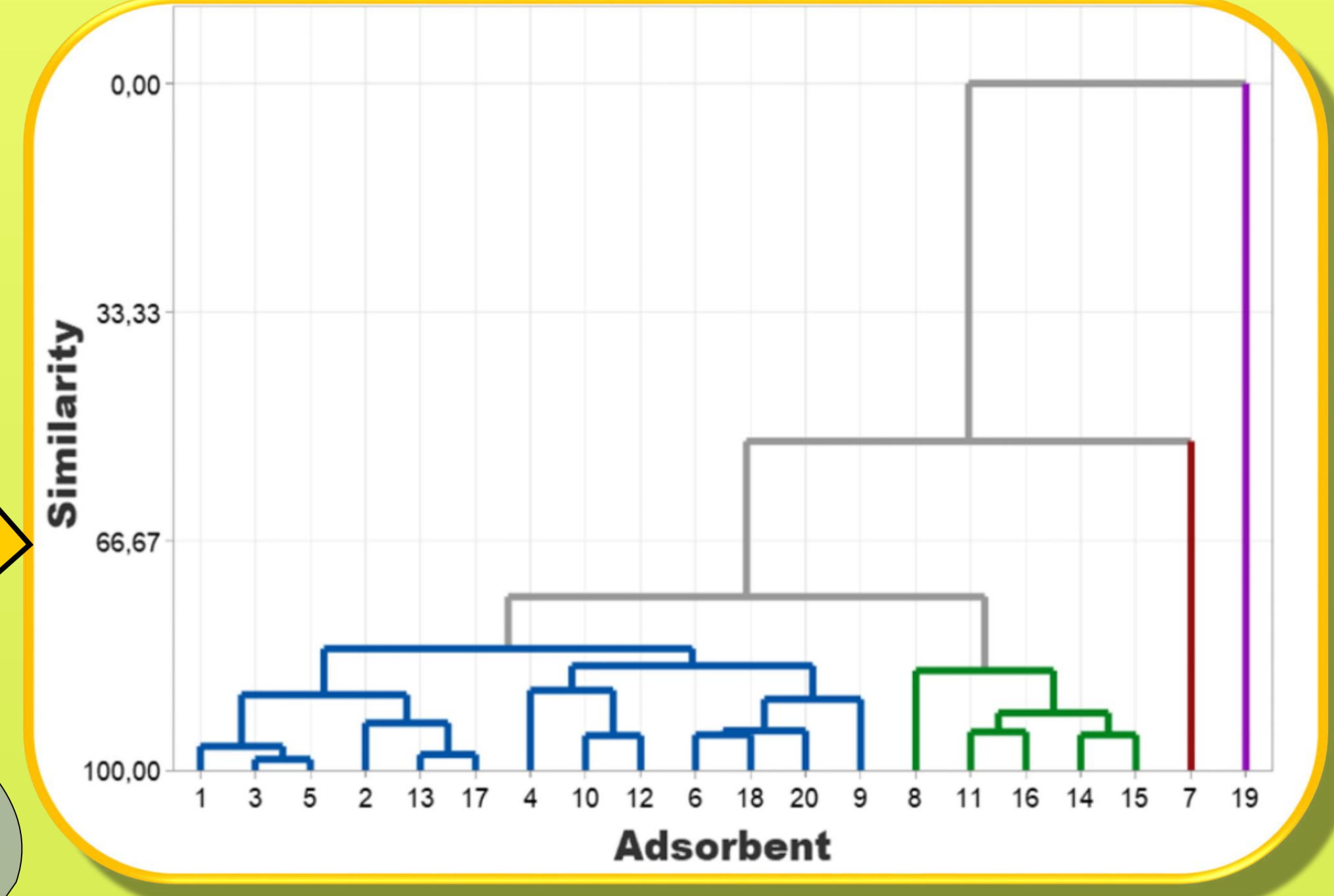


Nº	Name	Kind	Raw material	Activation
1	BAC	Commercial	Petroleum pitch	Industrial process
2	CNAC	Experimental	Anthracite	NaOH
3	CX19	Experimental	Xerogel	Microwave
4	CX45	Experimental	Xerogel	Microwave
5	CX8	Experimental	Xerogel	Microwave
6	F400	Commercial	Bituminous coal	WA
7	HYDC	Commercial	Lignite	vapor
8	MAC	Experimental	Lignite	KOH
9	NPK	Commercial	Peat	vapor
10	BWA	Experimental	Biocollagenic waste	KOH
11	BWB	Experimental	Biocollagenic waste	KOH
12	BWPC	Experimental	Biocollagenic waste	KOH
13	BWPD	Experimental	Biocollagenic waste	KOH
14	BWD	Experimental	Biocollagenic waste	K ₂ CO ₃
15	BWE	Experimental	Biocollagenic waste	K ₂ CO ₃
16	BWPF	Experimental	Biocollagenic waste	K ₂ CO ₃
17	BWPG	Experimental	Biocollagenic waste	K ₂ CO ₃
18	ROW	Commercial	Charcoal	vapor
19	SBC	Experimental	Sludge WWTP	vapor
20	YAO	Commercial	Coconut shell	vapor

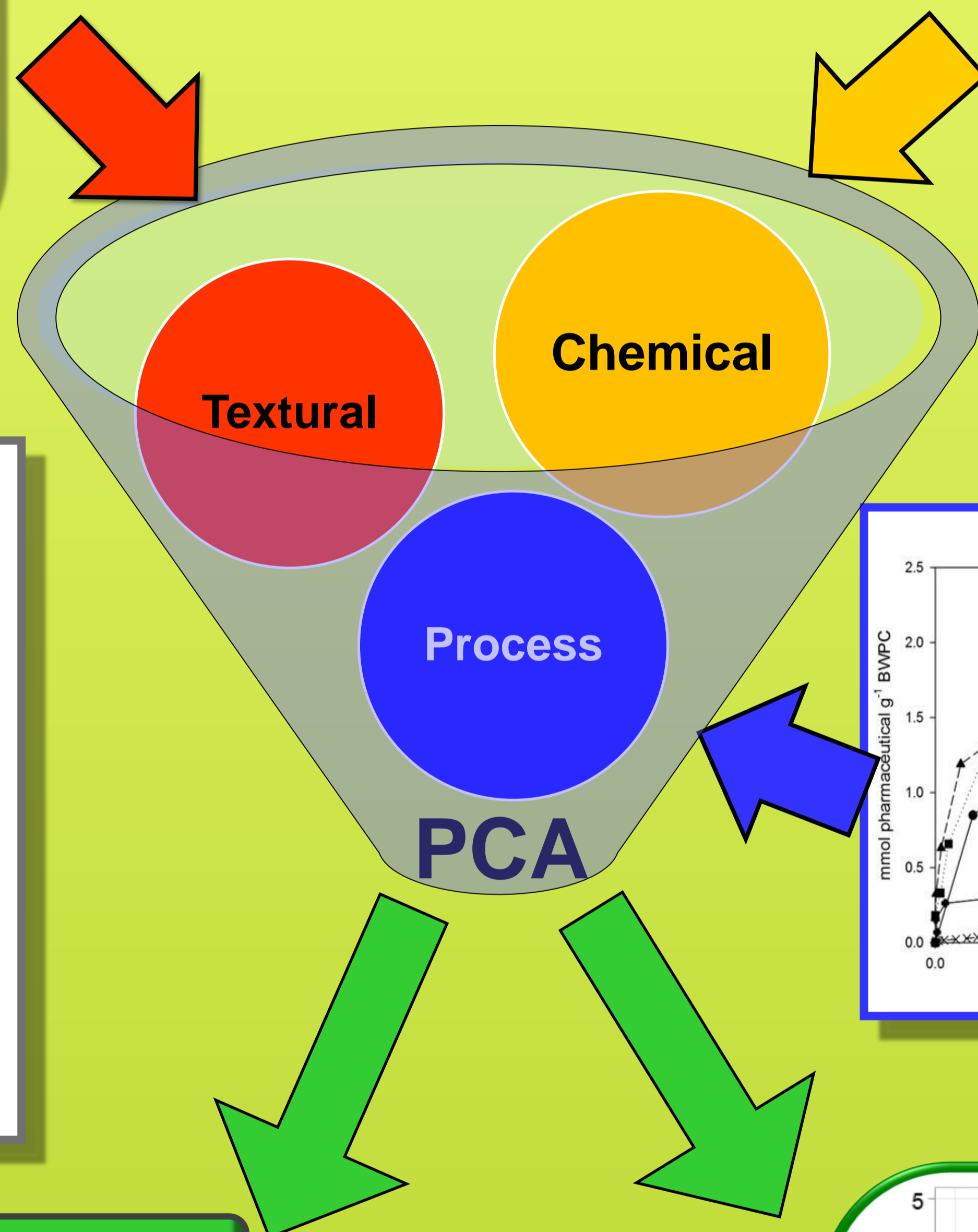
TEXTURAL CHARACTERIZATION



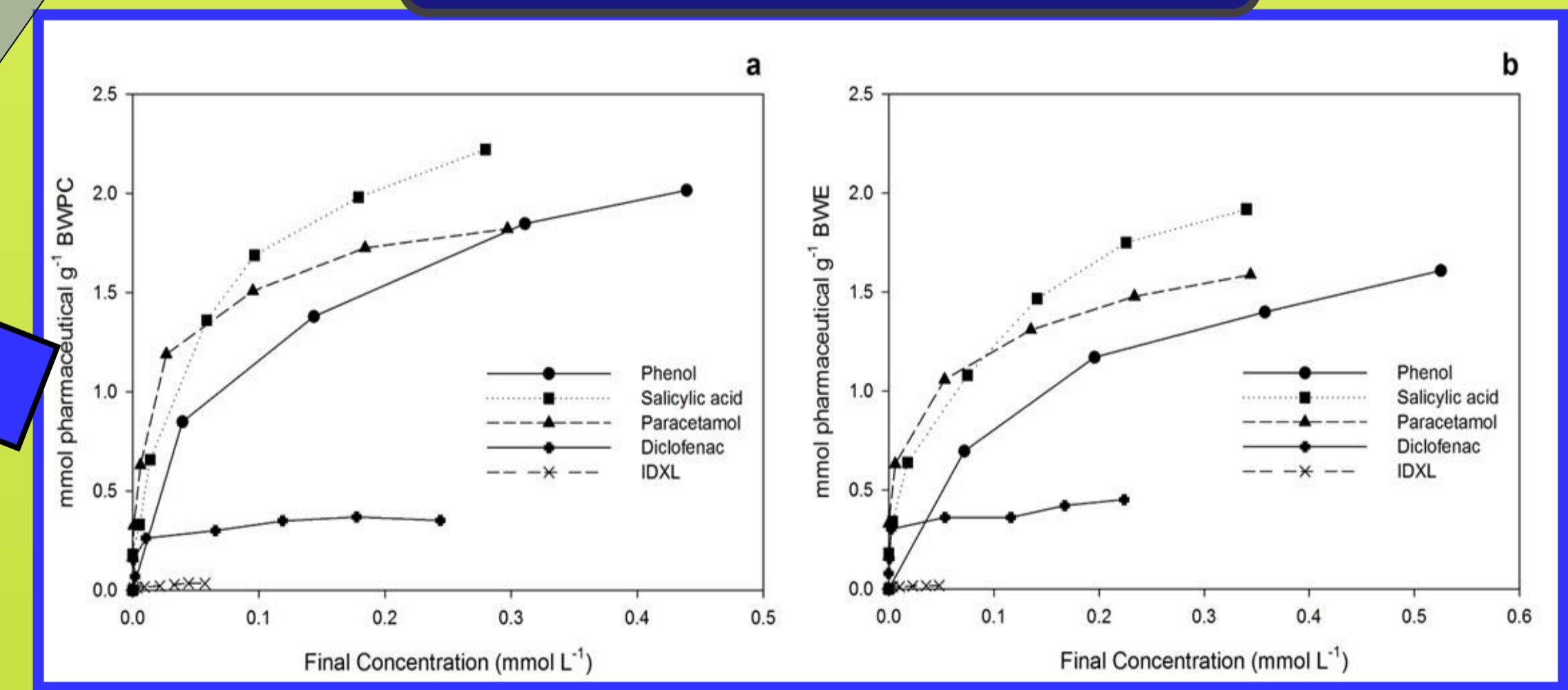
CHEMICAL CHARACTERIZATION



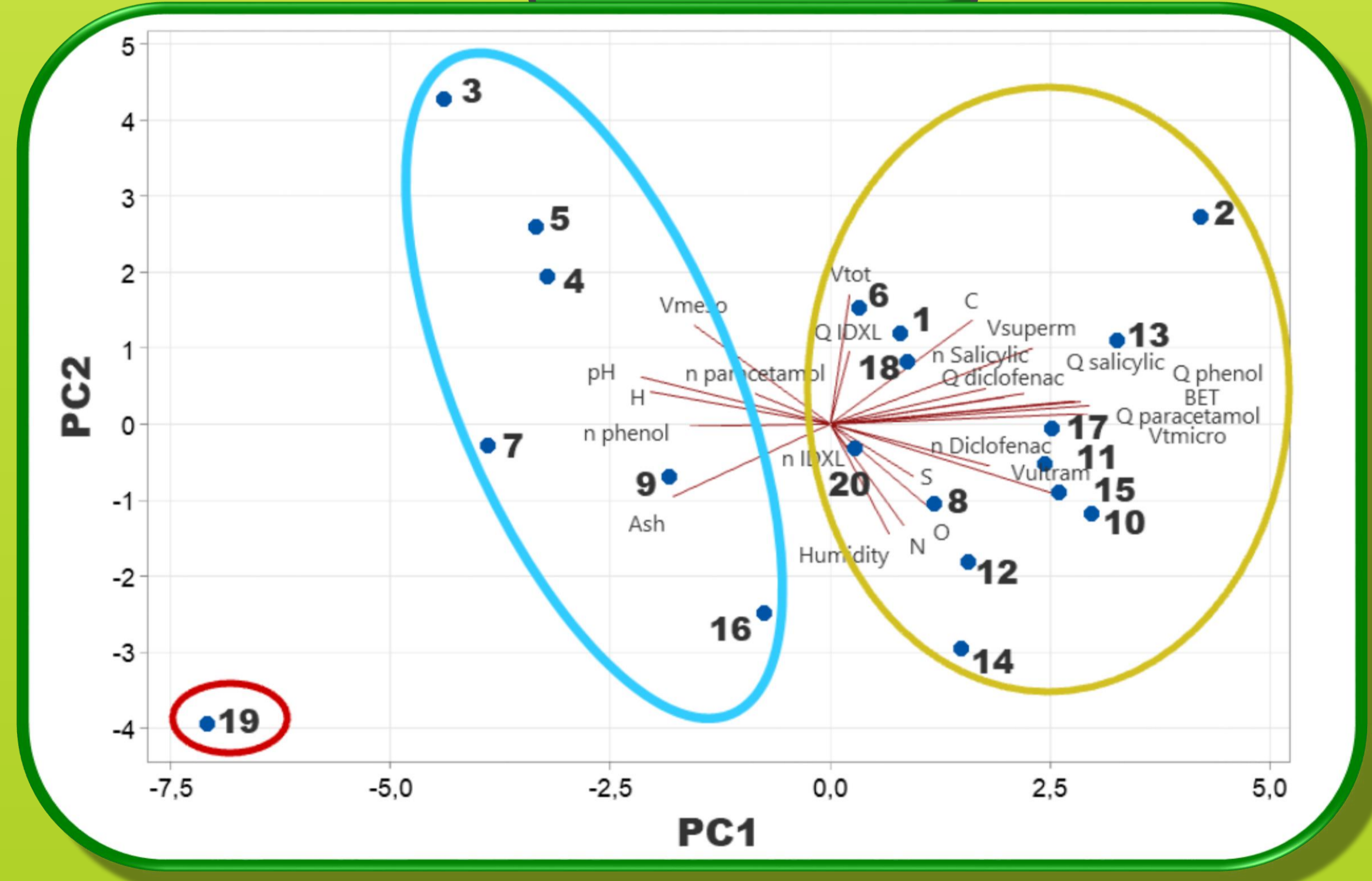
Results & Discussion



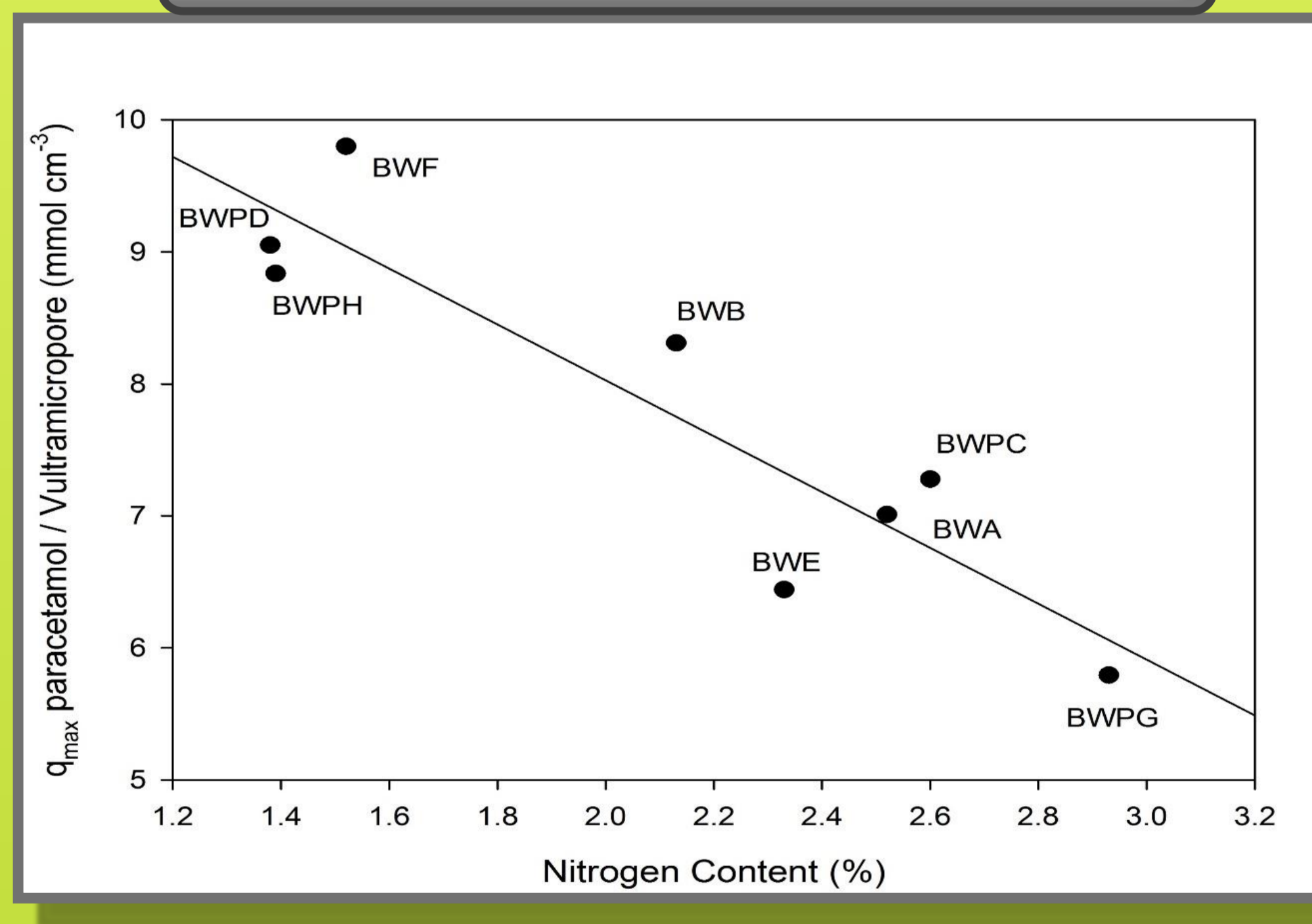
ADSORPTION PARAMETERS



PCA RESULTS



CHEMICAL INFLUENCES ON BWACs



Qmax MULTIPLE REGRESSION MODEL

Model	r ²
$Q_{phenol} = 0.059 + 0.02496 \cdot C - 1.101 \cdot H - 0.0565 \cdot O + 0.863 \cdot V_{SUPERM}$	87.71%
$Q_{salicylic} = 0.087 + 0.0238 \cdot C - 0.985 \cdot H - 0.3215 \cdot S + 0.0455 \cdot O + 2.55 \cdot V_{ULTRAM} + 1.51 \cdot V_{SUPERM}$	86.67%
$Q_{paracetamol} = 1.920 + 0.0103 \cdot C - 0.476 \cdot H - 0.1155 \cdot S - 0.0321 \cdot O - 0.3081 \cdot N - 0.1206 \cdot pH + 3.065 \cdot V_{ULTRAM}$	98.30%
$Q_{diclofenac} = 0.998 - 0.00445 \cdot C - 0.861 \cdot H + 0.0243 \cdot O + 1.76 \cdot V_{SUPERM} + 0.672 \cdot V_{MESO}$	71.31%
$Q_{iodixanol} = -0.1962 + 0.00164 \cdot Ash + 0.001922 \cdot S_{BET} - 5.83 \cdot V_{ULTRAM} - 5.10 \cdot V_{SUPERM} + 0.01288 \cdot Humidity$	75.12%

Conclusions

- The presence of nitrogenated groups in BWACs decrease the adsorption of paracetamol due to water compete for the same sites of adsorption.
- The adsorption of phenol, salicylic acid and paracetamol was on micropores. Diclofenac and IDXL adsorption was preferably physical in the wider micropores and narrower mesopores.
- Multiple lineal regression models were proposed to predict maximum adsorption capacities of pharmaceutical. In the IDXL model the textural properties predominated, while in the paracetamol model the nitrogen content had a negative influence.

Acknowledgements

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