Prediction of transport and destination through subsoil, of lead from leaching of a landfill

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Introduction



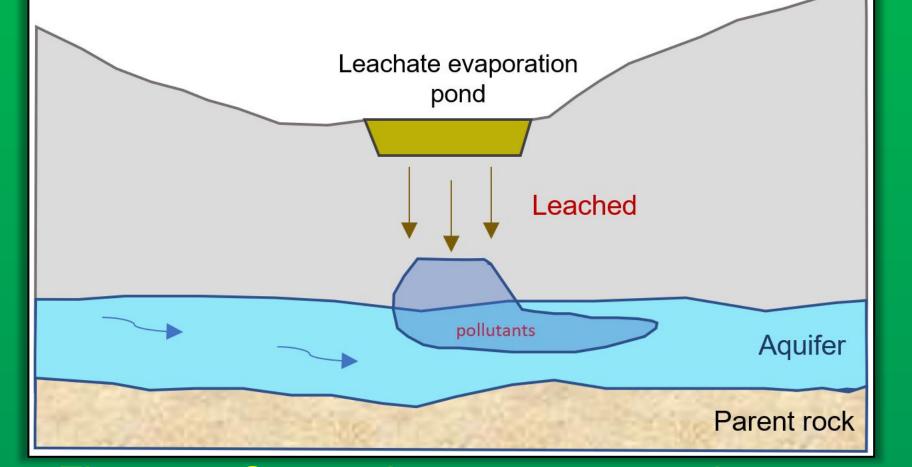


Figure 1: Contaminant transport scheme

The objective of this research work is to predict the transport and destination through the subsoil of lead from the leaching of a sanitary landfill, through the analysis and modeling of a scenario that puts at risk the health of the population and the integrity of the ecosystem. For this, a fictitious scenario was considered, assuming a leachate leak in the leachate evaporation pond.

The study site is a sanitary landfill located in the State of Mexico, Mexico; It is located in an area of high risk due to flooding and there is a neighboring population at a shorter distance than the regulation (Figure 2). To meet the proposed objective, the methodology shown in Figure 3 was followed.

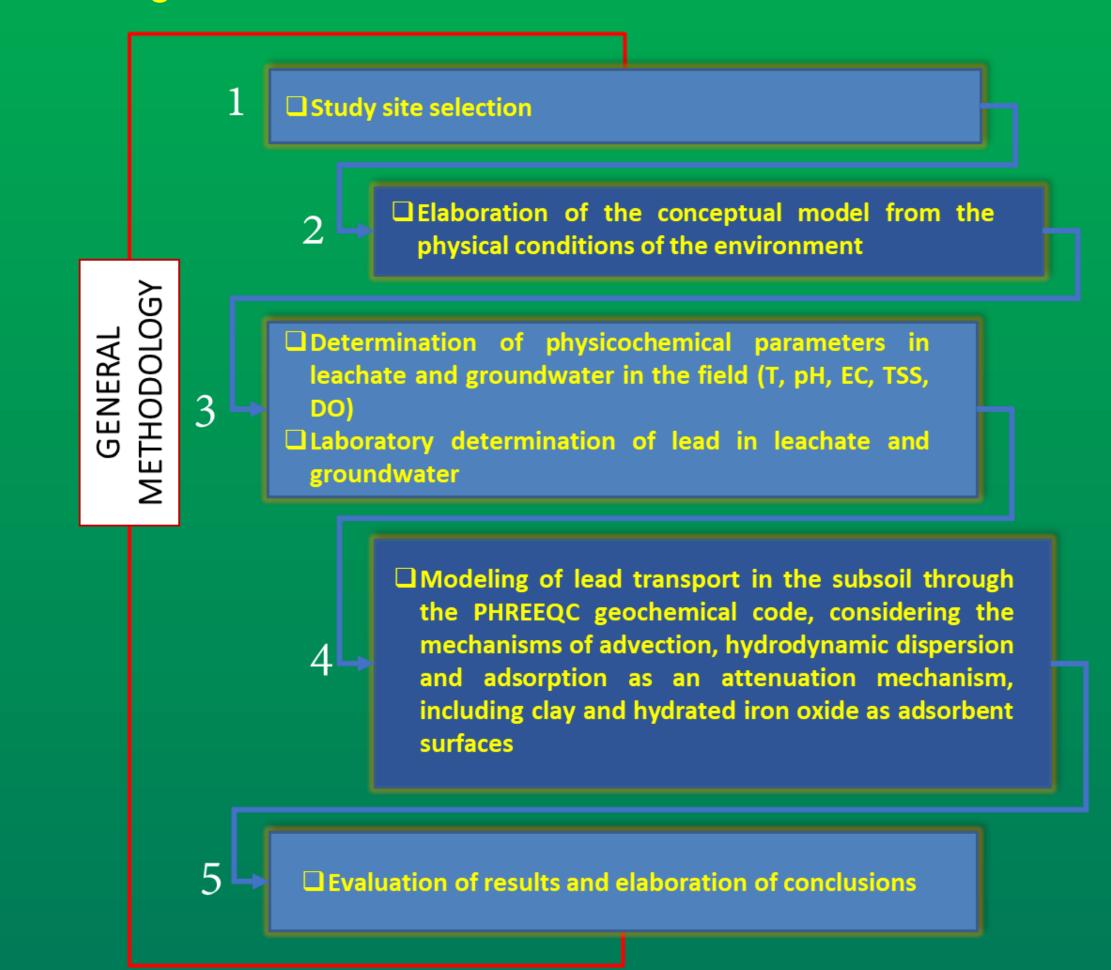
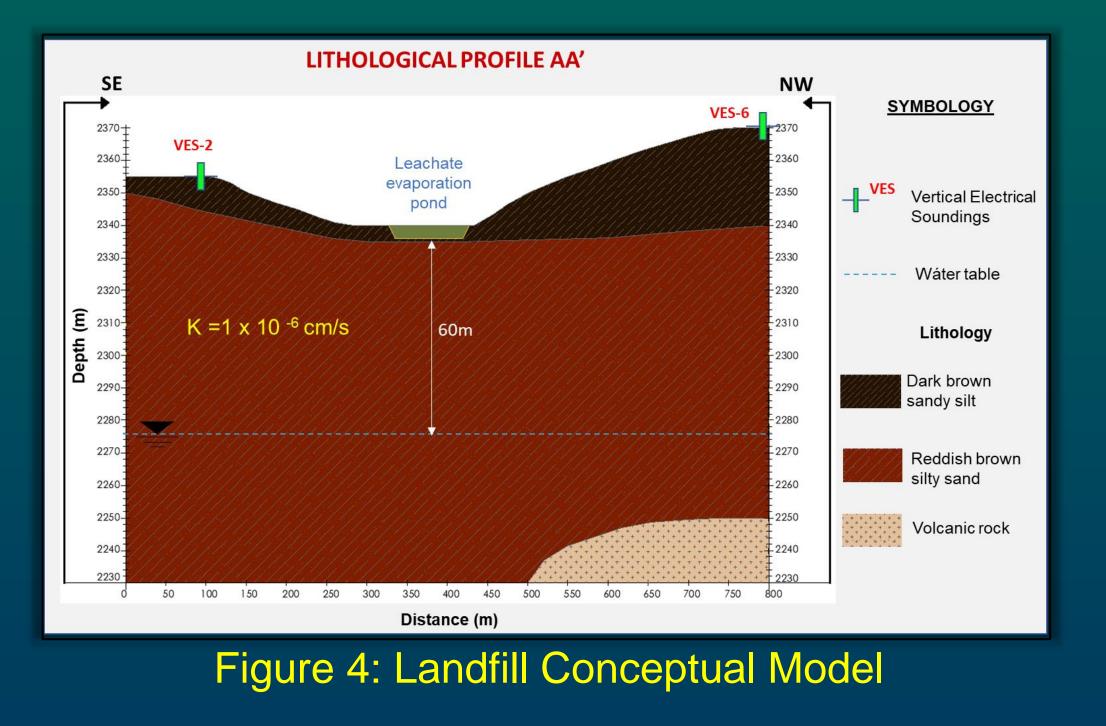


Figure 3: General methodology

Figure 2: Study Landfill

Results & Discussion

The conceptual model indicates that the subsoil of the study area is made up to a depth of 120 m of volcano-sedimentary materials, such as intercalations of firm to hard sandy silt tuffs and very compact brown silty sands. The permeability of the medium is low to medium, with a value of 1×10^{-6} cm/s and the water table is approximately 60 m deep (Figure 4).



The results of the Pb transport modeling indicate that, for a 60 m column, the amount of dissolved lead that would reach the end of the column would be 3.54×10^{-10} mg/L, contained in different species; value that can be considered negligible. Regarding the amount of lead retained on the two adsorbent surfaces considered, an amount of 0.89 mg/l was obtained, with the hydrated ferric oxide surface being the one with the highest Pb retention and to a much lesser extent by clay (Table 2).

	Dissolved		Adsorbed			
	Molality (mol/kg)	mg/L	Adsorbent surface	Molality (mol/kg)	mg/L	
Total, disolved	1.71E-15	3.54E-10	Hydrated Ferric Oxide	4.315E-06	0.8941	
			Montmorillonite	2.091E-09	0.0004	
			Total, adsorbed	4.32E-06	0.894	

Table 2: Dissolved and adsorbed lead concentrations

The following graph shows the transport profile of Pb in the leachate for 36 years, the time it will take for the lead to travel through the 60m column; where a reduction of the aqueous concentrations is observed due to the

The pH of the leachate from the site is in the range of 6.5 to 7.89; therefore, the leachate is in a methanogenic phase. The analytical results of the leachate samples indicate the presence of lead in all of them, with a maximum concentration of 0.894 mg/L (concentration used for modeling) and a minimum of 0.559 mg/L (Table 1).

Table 1: Lead concentrations in leachate and groundwater

ID		Sample análisis (mg/L)		
ID	Matrix	Pb	SD	LQ
LIX-01-LAG	Leached	0.590	0.008	0.2
LIX-02-CARC1	Leached	0.894	0.003.	0.2
LIX-03-CARC2	Leached	0.559	0.002	0.2
MA-01-POZO VIRREYES	Groundwater	< 0.2	0.001	0.2
LQ= Limit of Quantification. SD=	Standard Deviation			

adsorption of lead in the medium, which limits its transport to only 1.5 m (Figure 5).

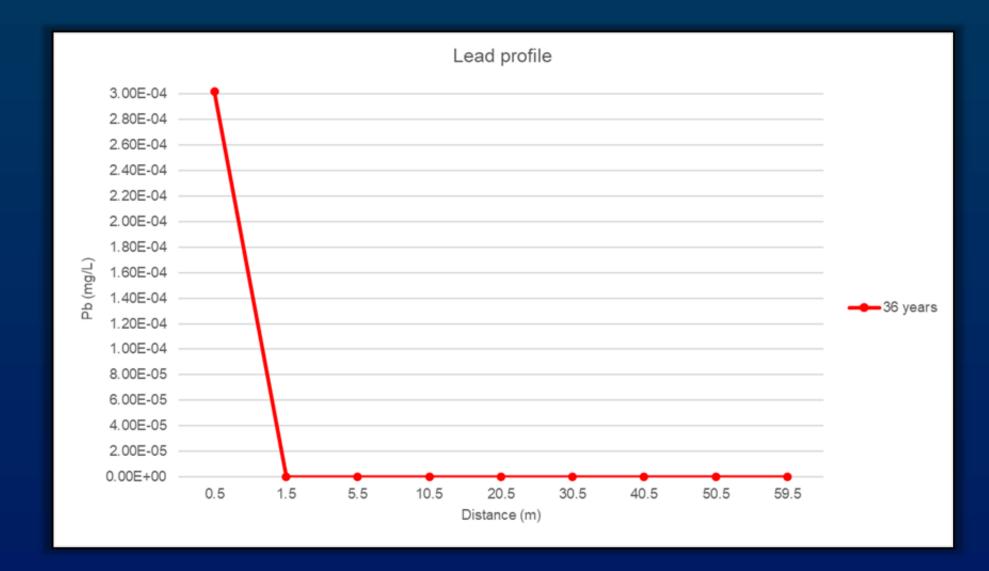


Figure 5: Transport of Pb in leachate at 36 years

Conclusions

The results obtained from the modeling of lead transport in the unsaturated zone indicate that all the lead contained in the leachate that infiltrates the subsoil of the sanitary landfill will be adsorbed in the medium, and will not reach dissolved until a depth of 60 m, so the lead will not reach the water table and will not impact the groundwater. It should be noted that the characteristics of the environment inhibit the dispersion of the pollutant.

The first author wishes to express his gratitude to CONACYT for the scholarship provided; to the Postgraduate Studies Support Program (PAEP) of the UNAM, for the economic support provided and to the Tersa del Golfo Company, for its openness and collaboration for access to the landfill and its technical information.